



# Synchrotron EXAFS & XANES techniques for Chemical Speciation on Environmental Systems

ALBA, 6/10/2014 – 9/10/2014

## X-ray absorption spectroscopy (XAS – XANES/EXAFS)

***Laura Simonelli***

*CLÆSS (Core Level Absorption & Emission Spectroscopies)  
beamline responsible*

# OUTLINE

- The absorption process
- The multiple scattering region (XANES)
- The single scattering region (EXAFS)
- Detection methods
- Research field
- Summary

# X-rays

## X-ray Absorption Spectroscopy (XAS)

**probe**

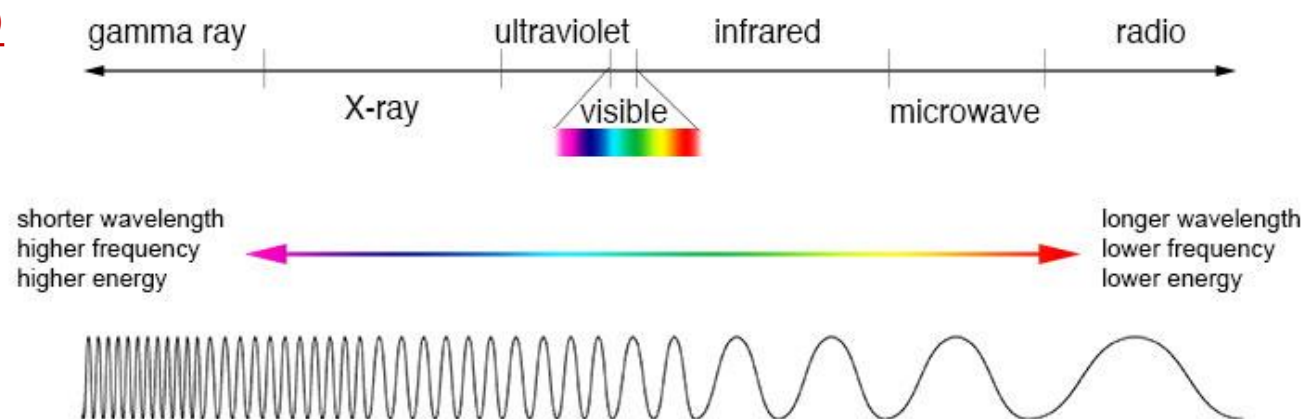
electromagnetic radiation

E from 120 eV (soft) to 120 keV (hard);

$\lambda$  from 100 Å to 0.1 Å

$$E \text{ (keV)} = \frac{12.4 \text{ (keV)}}{\lambda \text{ (Å)}}$$

measure the energy  
of an interaction

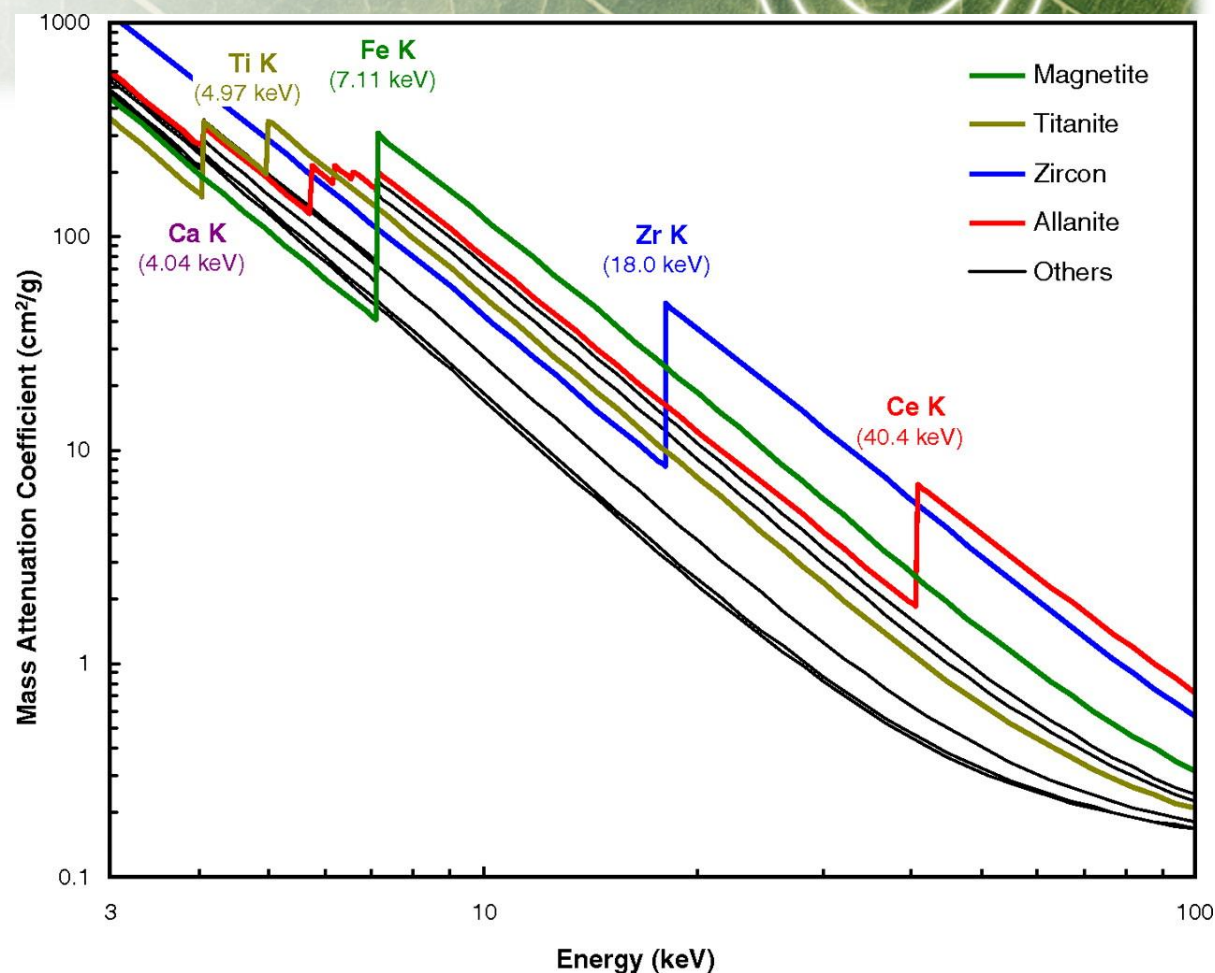


# Photon absorption

The oscillating electric field of the electromagnetic radiation interacts with the electrons bound in an atom.

Scattered radiation

Absorbed radiation





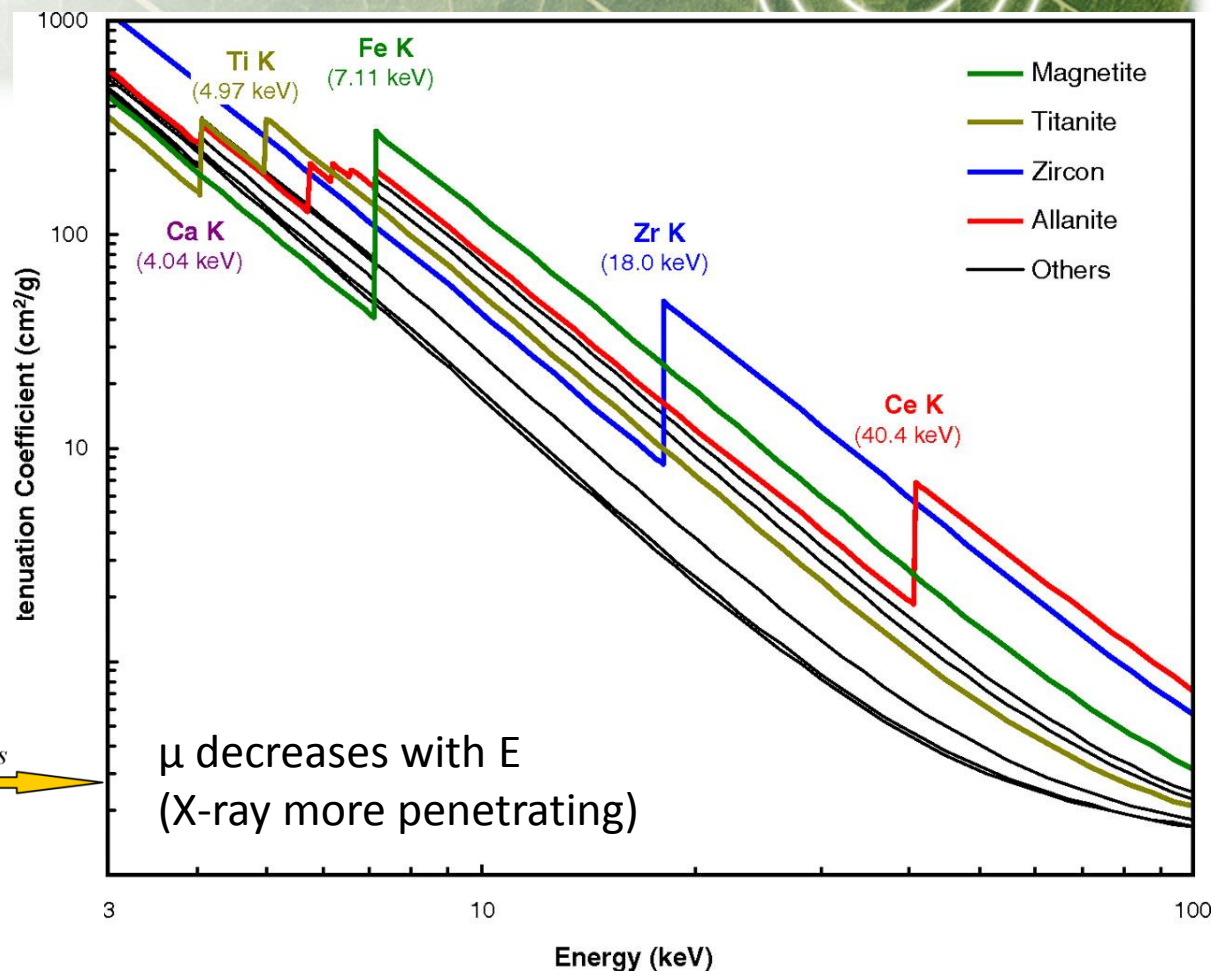
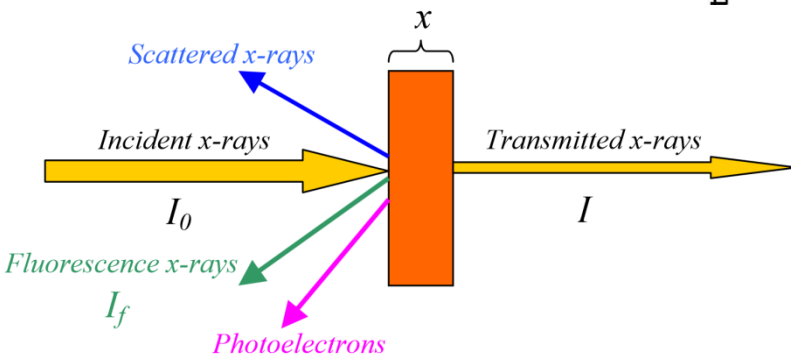
# Photon absorption



A monochromatic x-ray beam of intensity  $I_0$  passing through a sample of thickness  $x$  will get a reduced intensity  $I$  according to:

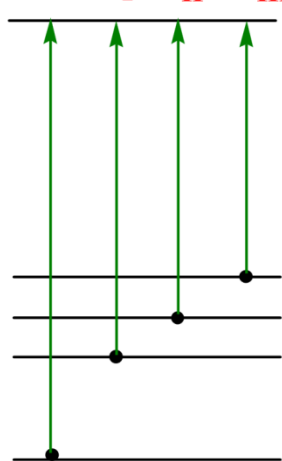
$$\ln(I_0/I) = \mu x$$

$\mu$  = linear absorption coefficient



# Photon absorption

edge: **K** **L<sub>I</sub>** **L<sub>II</sub>** **L<sub>III</sub>**

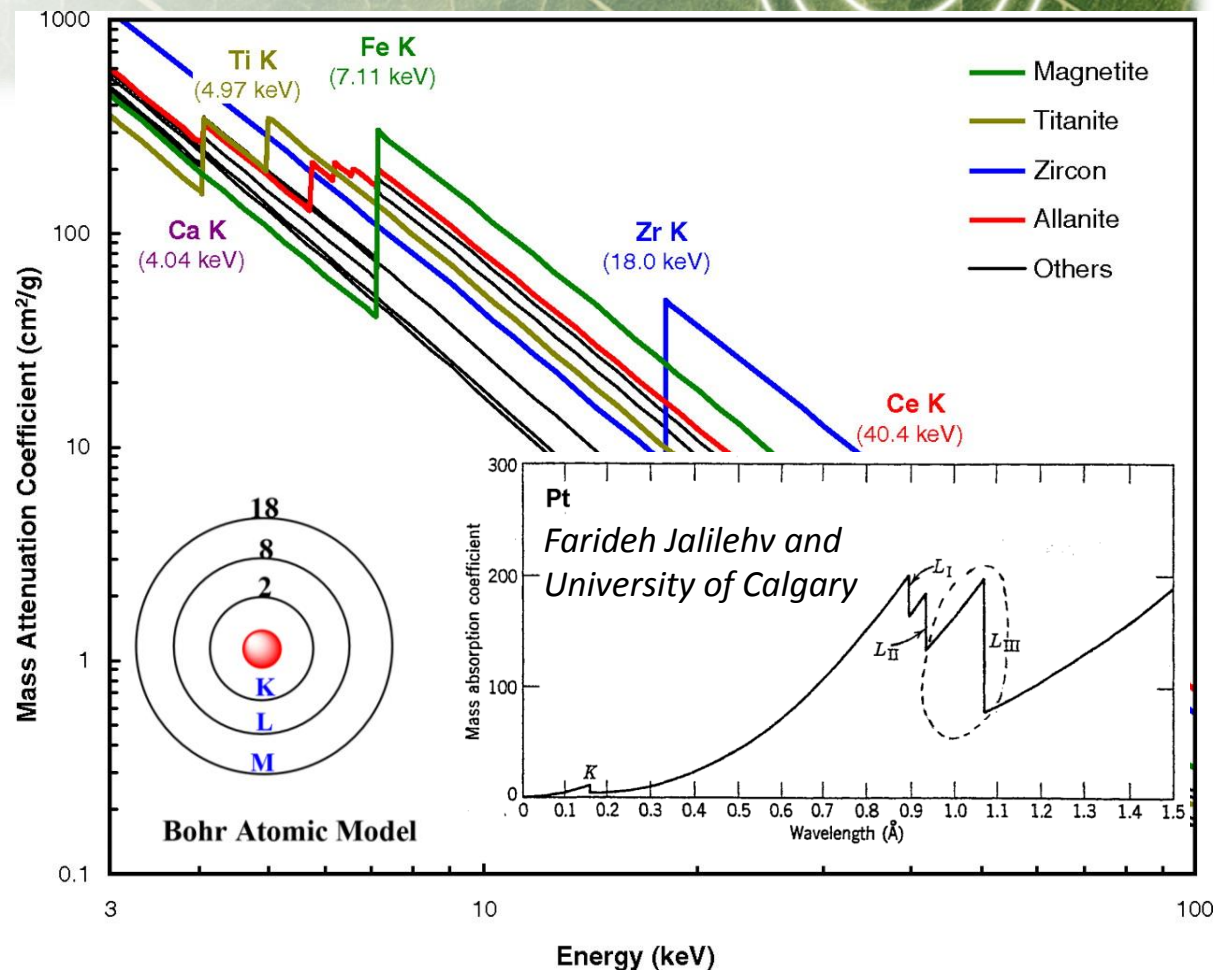


Continuum

$2P_{3/2}$   
 $2P_{1/2}$   
 $2S_{1/2}$   
 $2S_{1/2}$

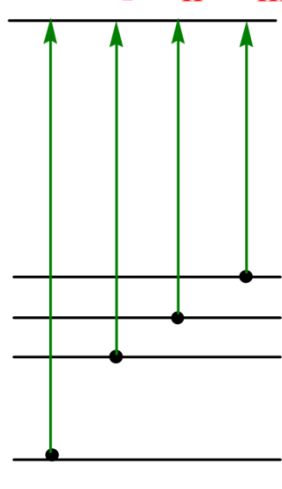
Absorption edges occur for each shell of core electrons when  $E = \text{binding energy of core electron}$

**K:**  $1s_{1/2}$  **L<sub>1</sub>:**  $2s_{1/2}$  **L<sub>2</sub>:**  $2p_{1/2}$  **L<sub>3</sub>:**  $2p_{3/2}$   
( $nl_j$  notation)



# Photon absorption

edge: **K** **L<sub>I</sub>** **L<sub>II</sub>** **L<sub>III</sub>**

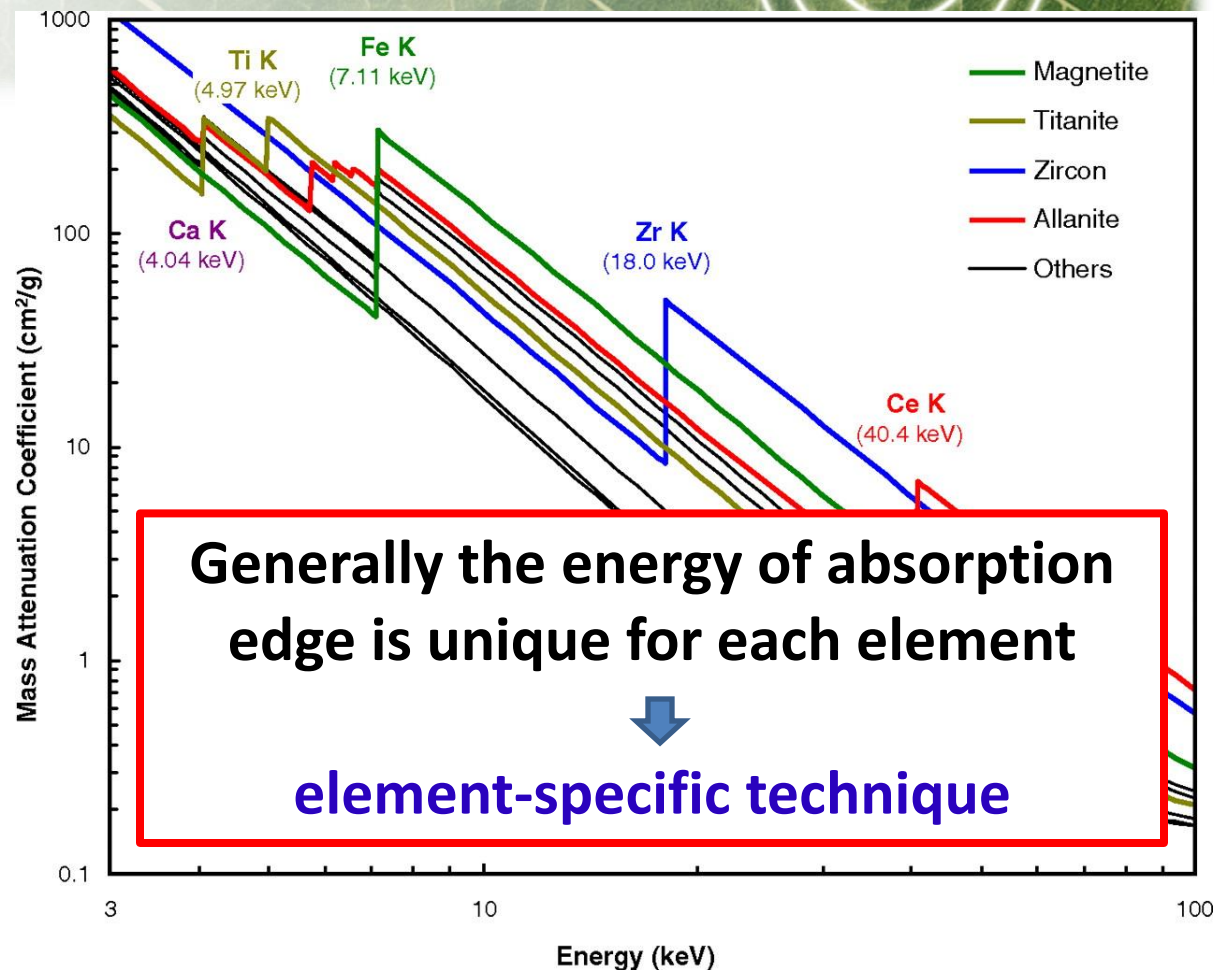


Continuum

$2P_{3/2}$   
 $2P_{1/2}$   
 $2S_{1/2}$   
 $2S_{1/2}$

Absorption edges occur for each shell of core electrons when  $E = \text{binding energy of core electron}$

K:  $1s_{1/2}$  L<sub>I</sub>:  $2s_{1/2}$  L<sub>II</sub>:  $2p_{1/2}$  L<sub>III</sub>:  $2p_{3/2}$   
( $nl_j$  notation)

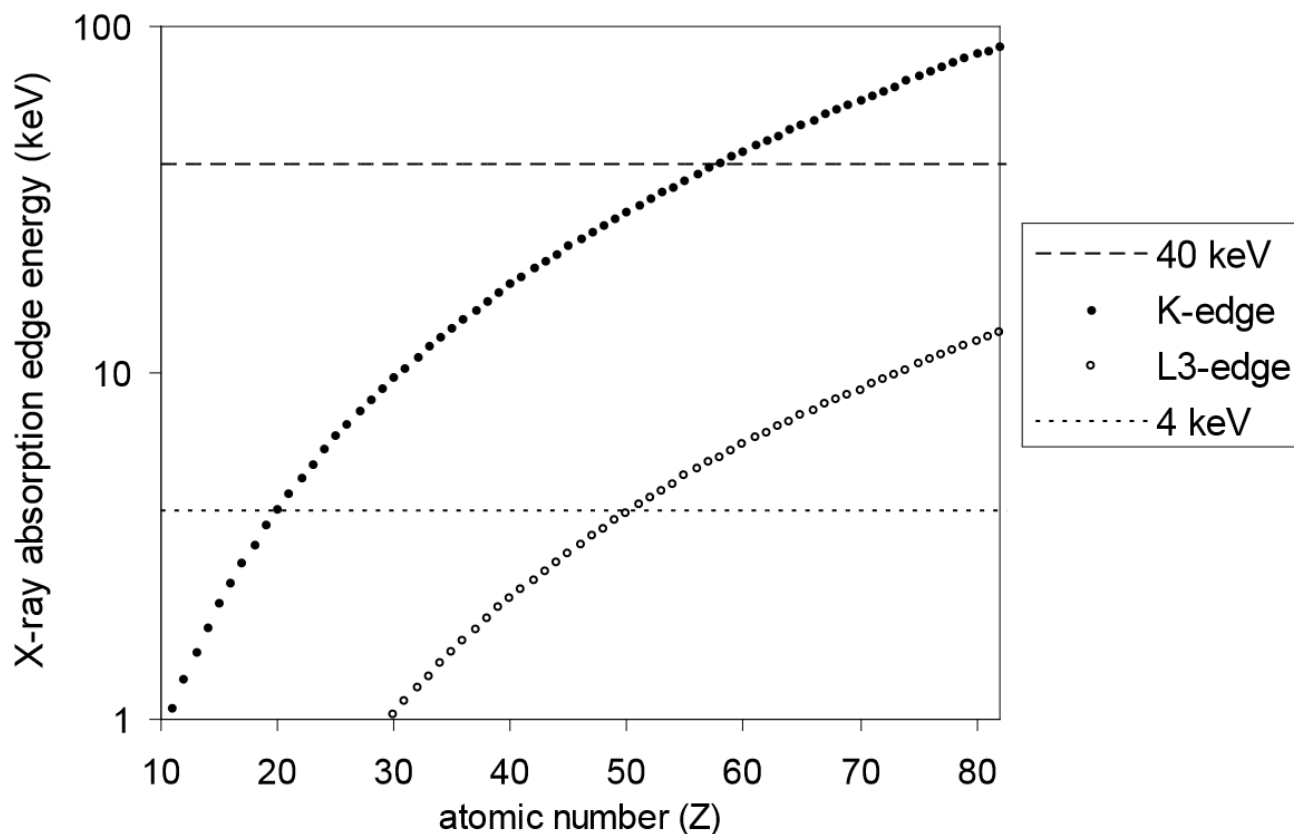




# Photon absorption:

difficult for light elements, e.g.  $Z < 22$  (Ti)

for heavy elements use L3 edge, e.g.  $Z > 57$  (La)





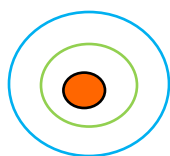
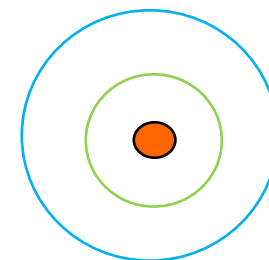
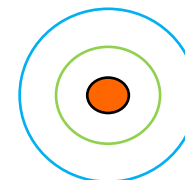
# Photon absorption: Photoelectron like a wave



photo-electron excited to state significantly above  
binding energy – becomes "quasi" free-electron

The photo-electron leaves the excited atom with  
circular wavefunction – wavelength  $\lambda = \frac{12.2}{\sqrt{KE(eV)}}$   
(kinetic energy  $KE = E - \text{binding energy}$ )

Atomic absorption

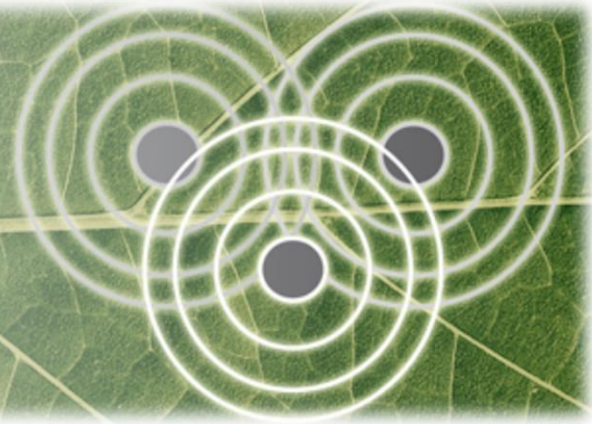


higher E,  
larger KE,  
Smaller  $\lambda$

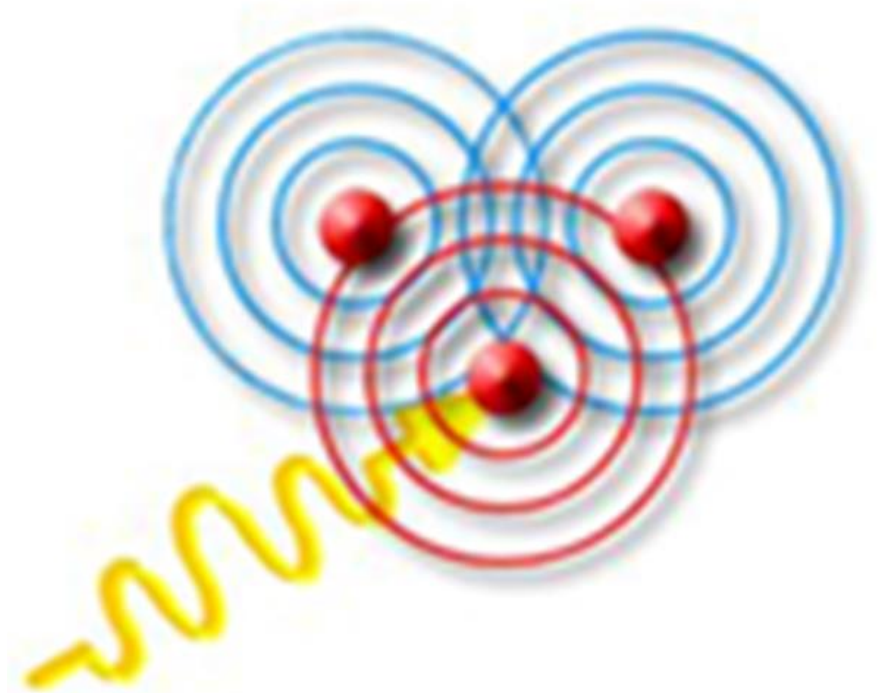


lower E,  
smaller KE,  
larger  $\lambda$

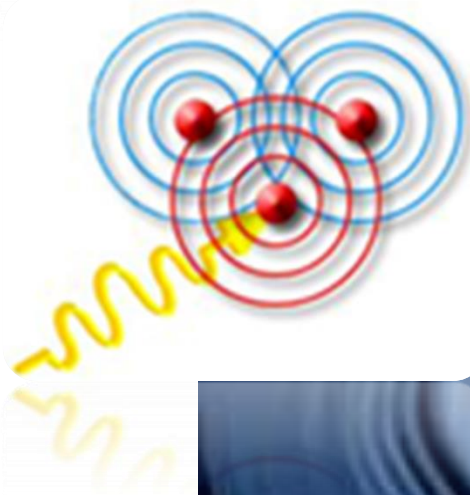
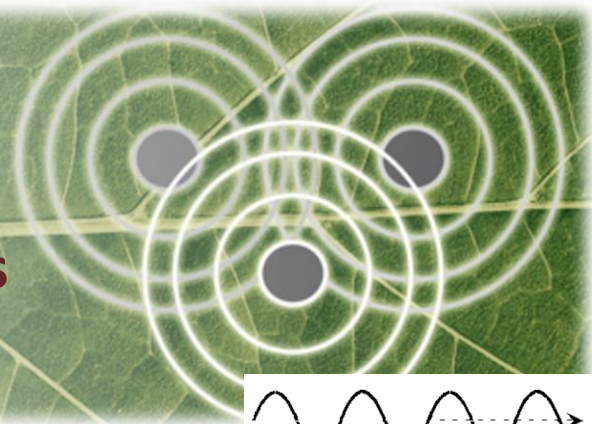
# Photon absorption: Photoelectron like a wave



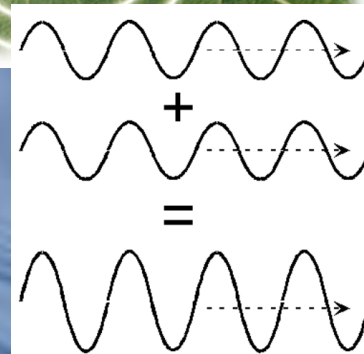
When the photoelectron leaves the absorbing atom, its wave is backscattered by the neighboring atoms.



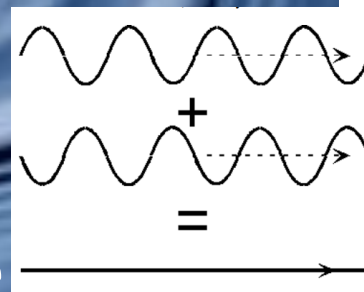
# Photon absorption: Interference of scattered waves



Interference is essentially an energy redistribution process: the energy which is lost at the destructive is regained at the constructive interference.



Constructive



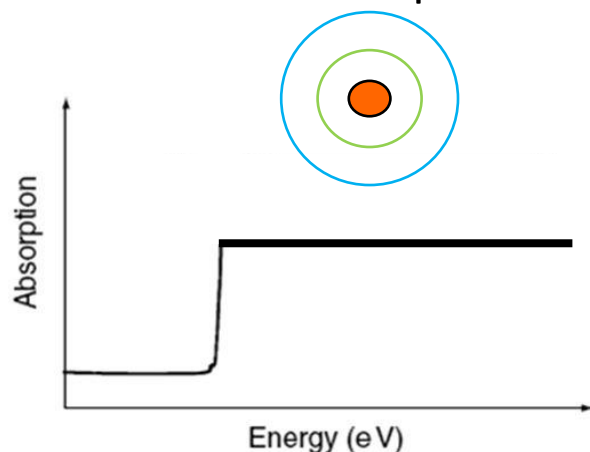
destructive

**Constructive and destructive interference  
between the outgoing photoelectron wave  
and backscattered wave**



# Photon absorption: Absorber isolated or embedded in the matter

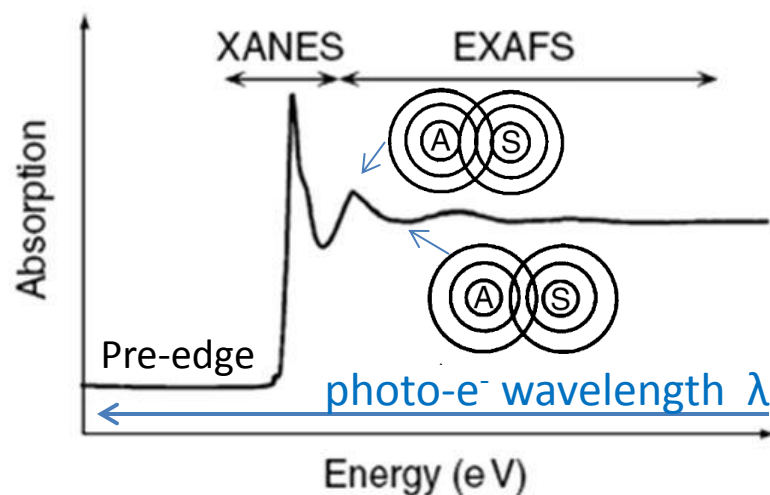
## Atomic absorption



The absorption of an isolated atom shows a step-like behavior.



When the absorber is embedded in the matter, the photoexcitation cross-section is modulated by the interference between the outgoing and the back-scattered photoelectron waves.



# Photon absorption: XANES & EXAFS

## XANES (or NEXAFS)

(X-ray Absorption Near Edge Structure):

$$\approx -10 \text{ eV} < E - E_0 < 50 \text{ eV}$$

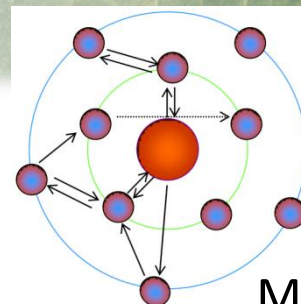
*Lower energy*  $\Rightarrow$  *Higher  $\lambda$*   $\Rightarrow$  *multiple scattering*

## EXAFS

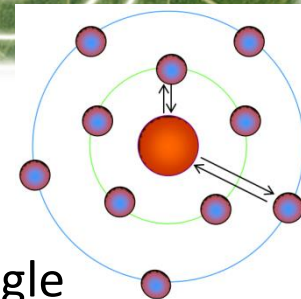
(Extended X-ray Absorption Fine Structure):

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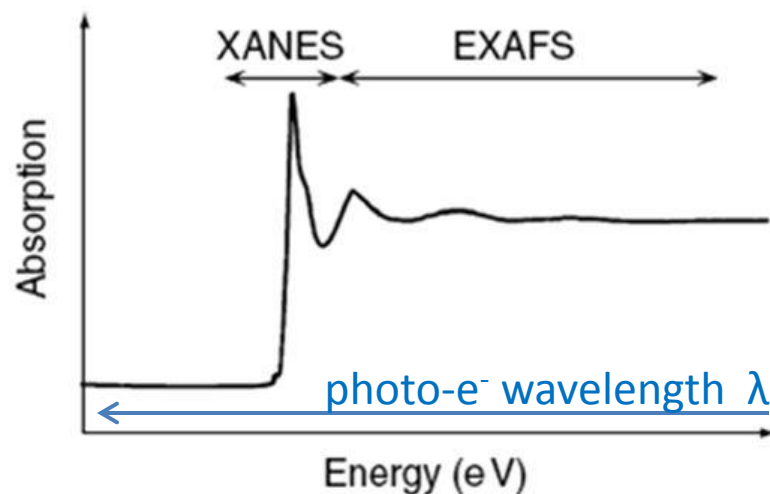
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Multiple  
scattering  
region



Single  
scattering  
region



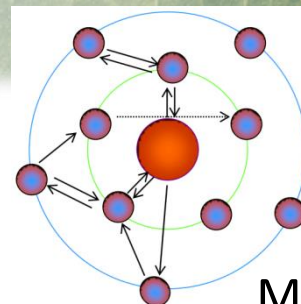
# Photon absorption: XANES

## XANES (or NEXAFS)

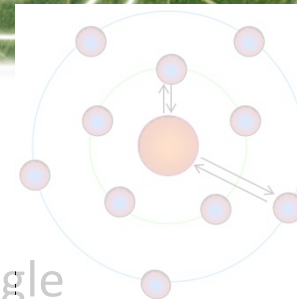
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Multiple  
scattering  
region



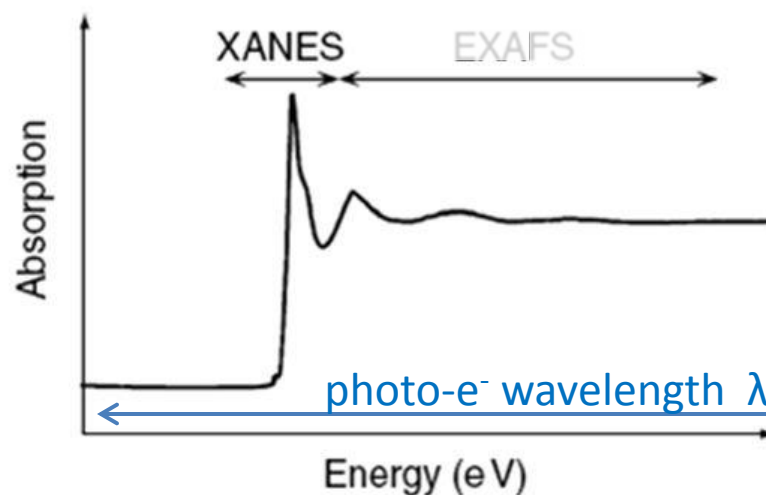
Single  
scattering  
region

## EXAFS

(Extended X-ray Absorption Fine Structure):

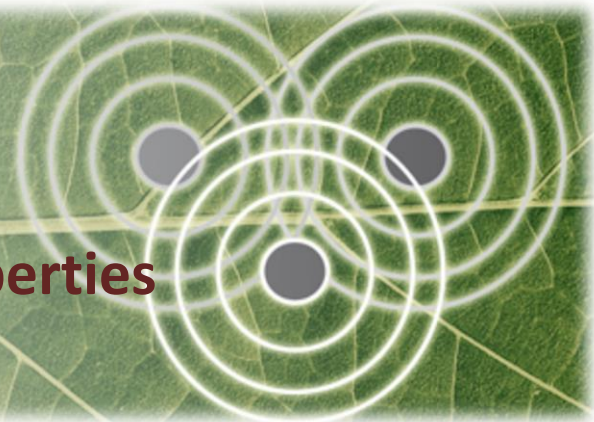
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*Higher energy*  $\Rightarrow$  *Lower  $\lambda$*   $\Rightarrow$  *single scattering*





# XANES: structural and electronic properties



## ELECTRONIC AND STRUCTURAL PROPERTIES

Energy, width, and  
intensity



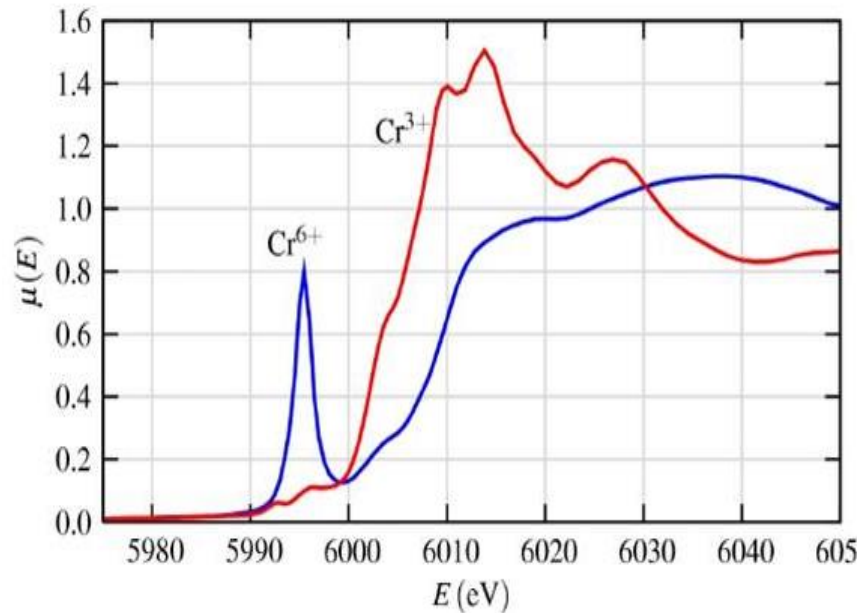
- **Oxidation state**
- **Unoccupied electronic states**
- **Spin state**
- **Local structure**

## STRUCTURAL PROPERTIES

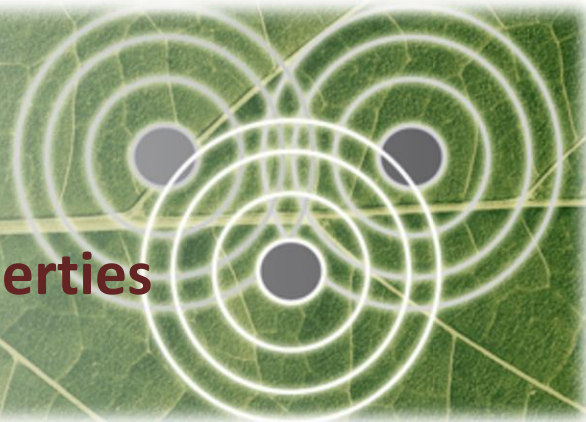
Angle dependence of  
multiple scattering



- **direct information about bond angles.**



# XANES: structural and electronic properties



## ELECTRONIC AND STRUCTURAL PROPERTIES

Energy, width, and  
intensity



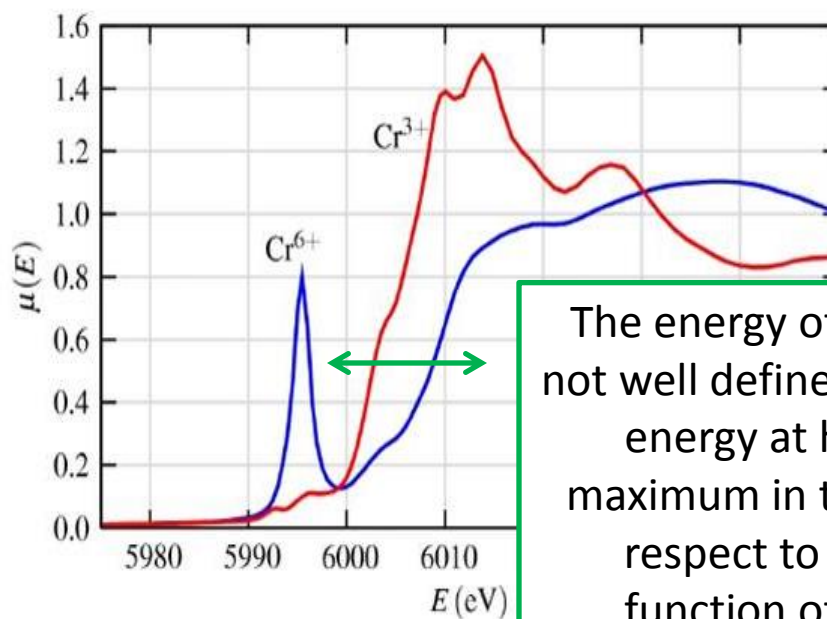
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## STRUCTURAL PROPERTIES

Angle dependence of  
multiple scattering

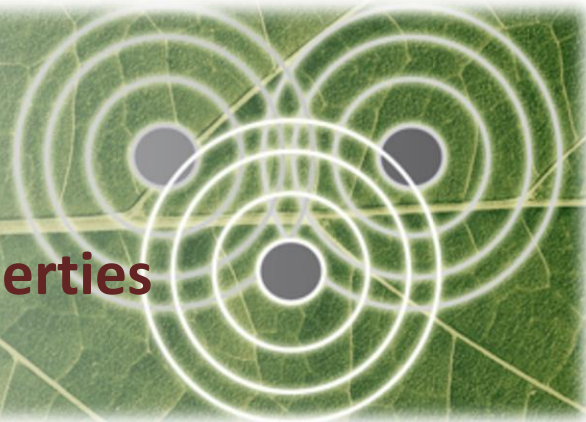


- **direct information about bond angles.**



The energy of an absorption edge is not well defined. It can be taken as the energy at half-height or as the maximum in the first derivative with respect to energy. It shifts as a function of the oxidation state

# XANES: structural and electronic properties



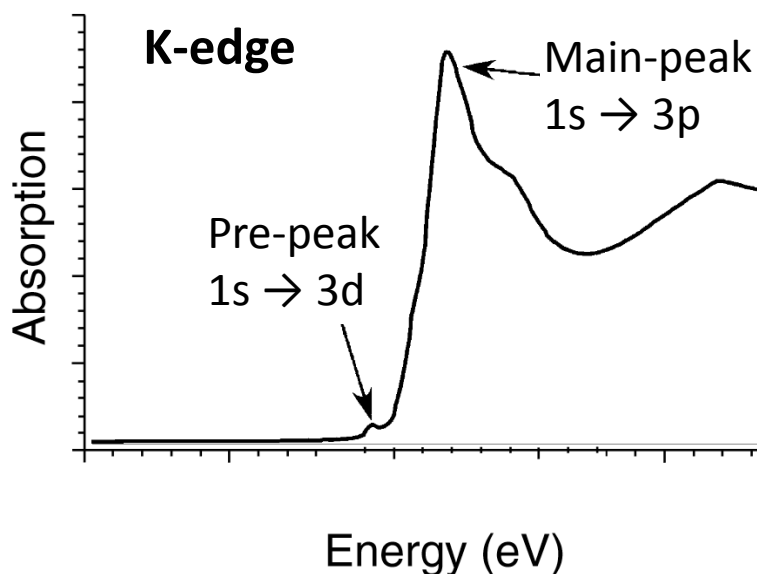
## ELECTRONIC AND STRUCTURAL PROPERTIES

Energy, width, and  
intensity



- **Oxidation state**
- **Unoccupied electronic states**
- **Spin state**
- **Local structure**

## STRUCTURAL PROPERTIES



The most intense features are due to electric-dipole allowed transitions (i.e.  $\Delta \ell = \pm 1$ ) to unoccupied final states. The most intense features of a K-edge are due to transitions from 1s → p-like final states, while the most intense features of the  $L_3$ -edge are due to 2p → d-like final states.



# XANES:

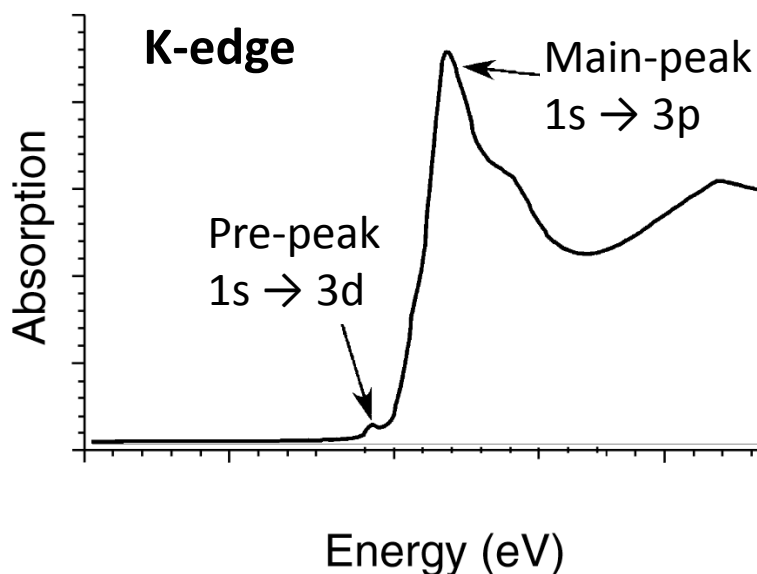
## structural and electronic properties

### ELECTRONIC AND STRUCTURAL PROPERTIES

Energy, width, and  
intensity

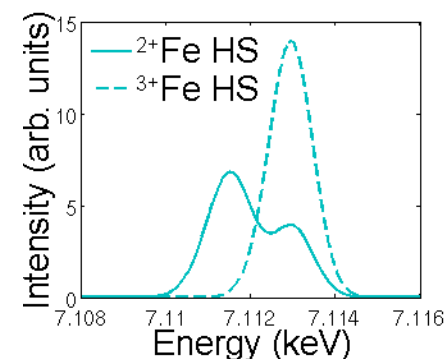


- **Oxidation state**
- **Unoccupied electronic states**
  - **Spin state**
- **Local structure**



### STRUCTURAL PROPERTIES

Theoretical simulation of  
the pre-edge region for  
tetrahedral high-spin  $2^+Fe$   
(solid line) and  $3^+Fe$   
(dashed line) complexes



# XANES:

## structural and electronic properties

### ELECTRONIC AND STRUCTURAL PROPERTIES

Energy, width, and  
intensity



- **Oxidation state**
- **Unoccupied electronic states**
  - **Spin state**
  - **Local structure**



### STRUCTURAL PROPERTIES

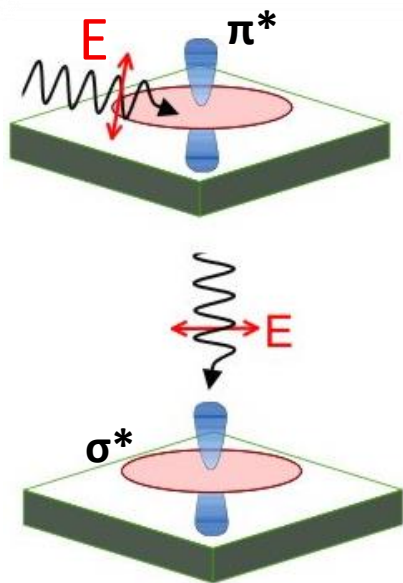
Angle dependence of  
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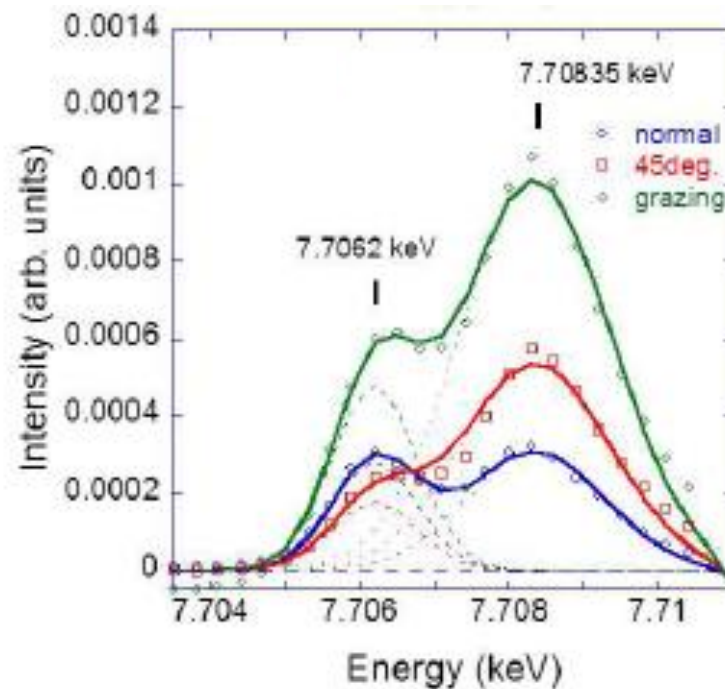
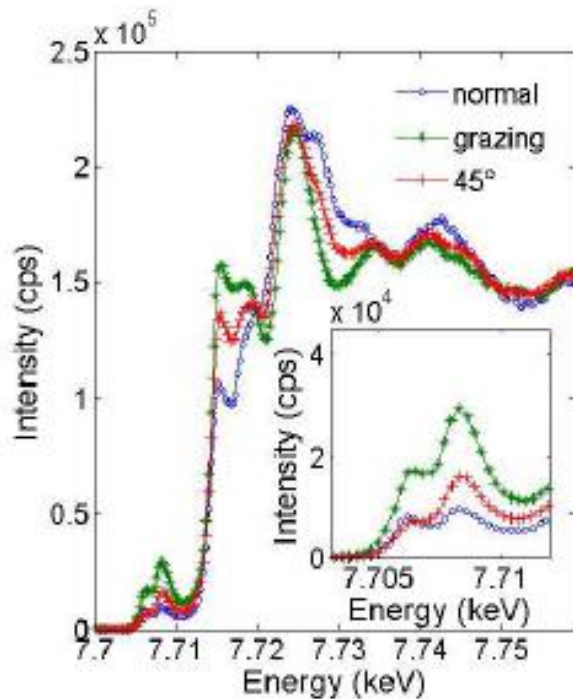
- **direct information about bond angles.**

XANES often used as simple  
“fingerprint” to identify presence  
of a particular chemical species.

# XANES: Polarization analysis



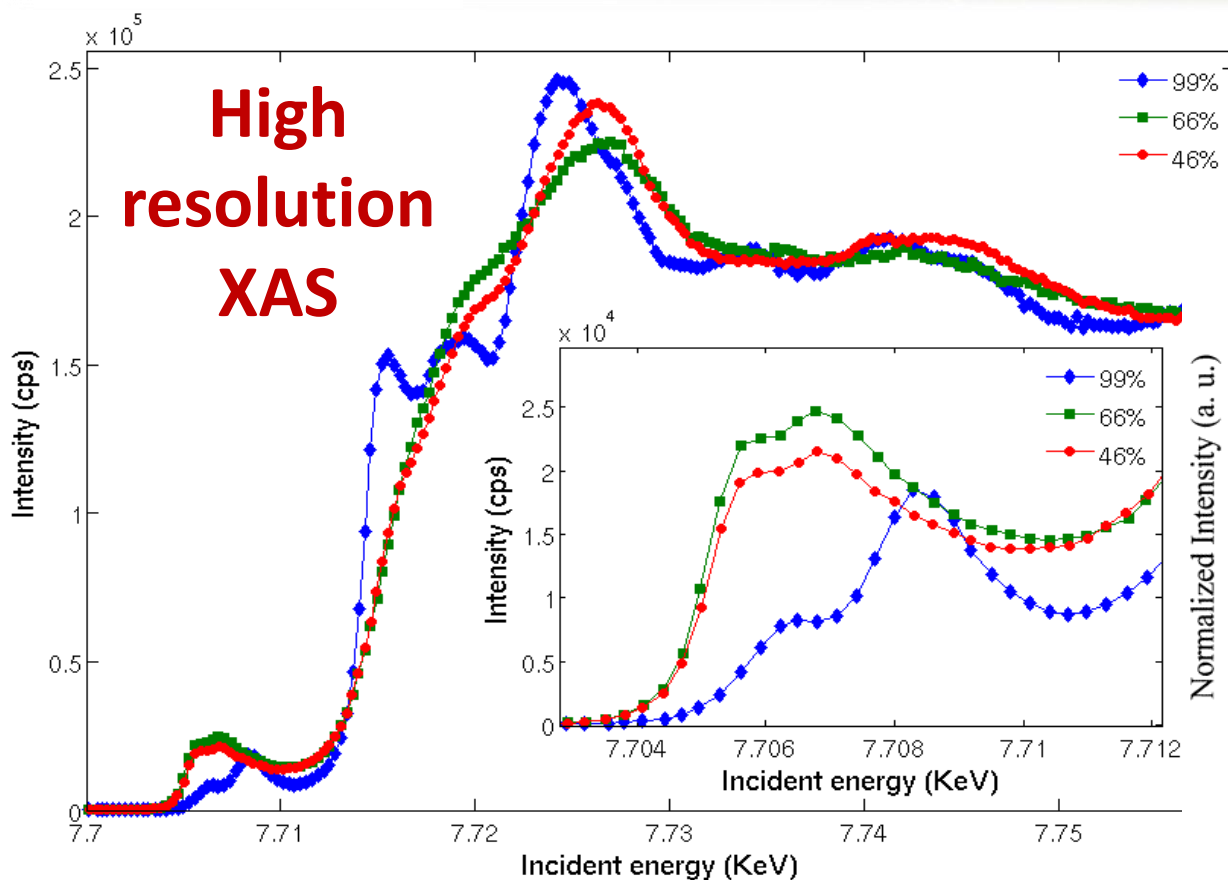
*The chosen angle of incidence favors either excitations into the  $\pi^*$  (top) or the  $\sigma^*$  (bottom) orbitals.*



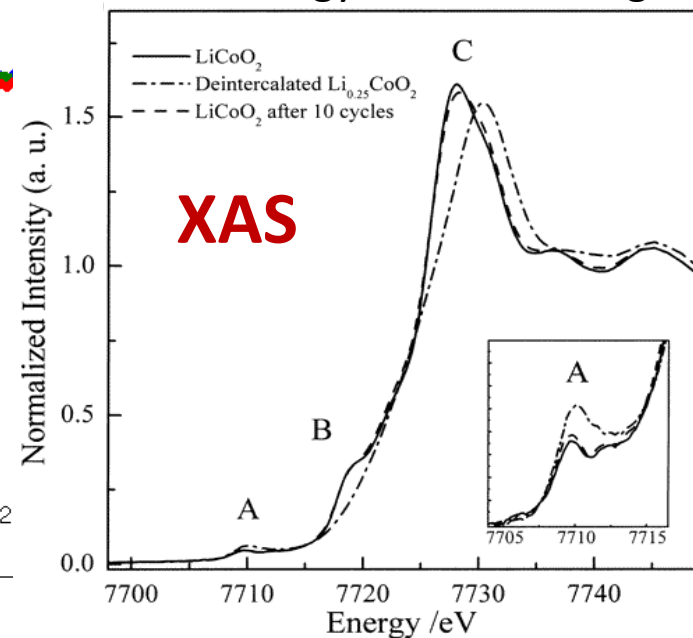


# XANES:

## High resolution x-ray absorption or Partial Fluorescence Yield (PFY)



By collecting an emission intensity line and scanning the incident energy across the edge.



# Photon absorption: EXAFS

## XANES (or NEXAFS)

(X-ray Absorption Near Edge Structure):

$$\approx -10 \text{ eV} < E - E_0 < 50 \text{ eV}$$

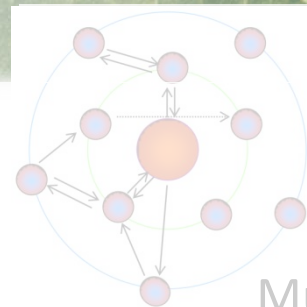
*Lower energy*  $\Rightarrow$  *Higher  $\lambda$*   $\Rightarrow$  *multiple scattering*

## EXAFS

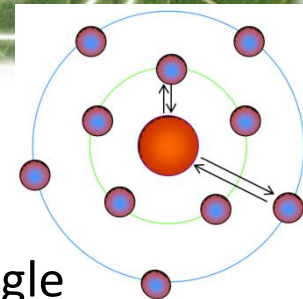
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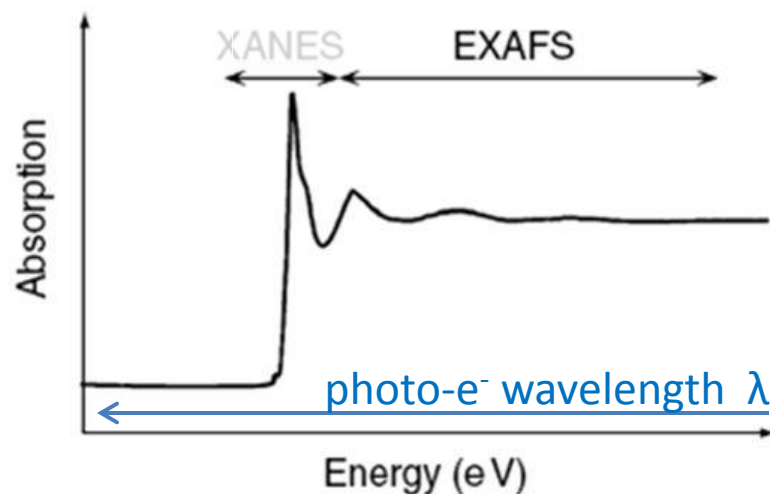
*Higher energy*  $\Rightarrow$  *Lower  $\lambda$*   $\Rightarrow$  *single scattering*



Multiple  
scattering  
region



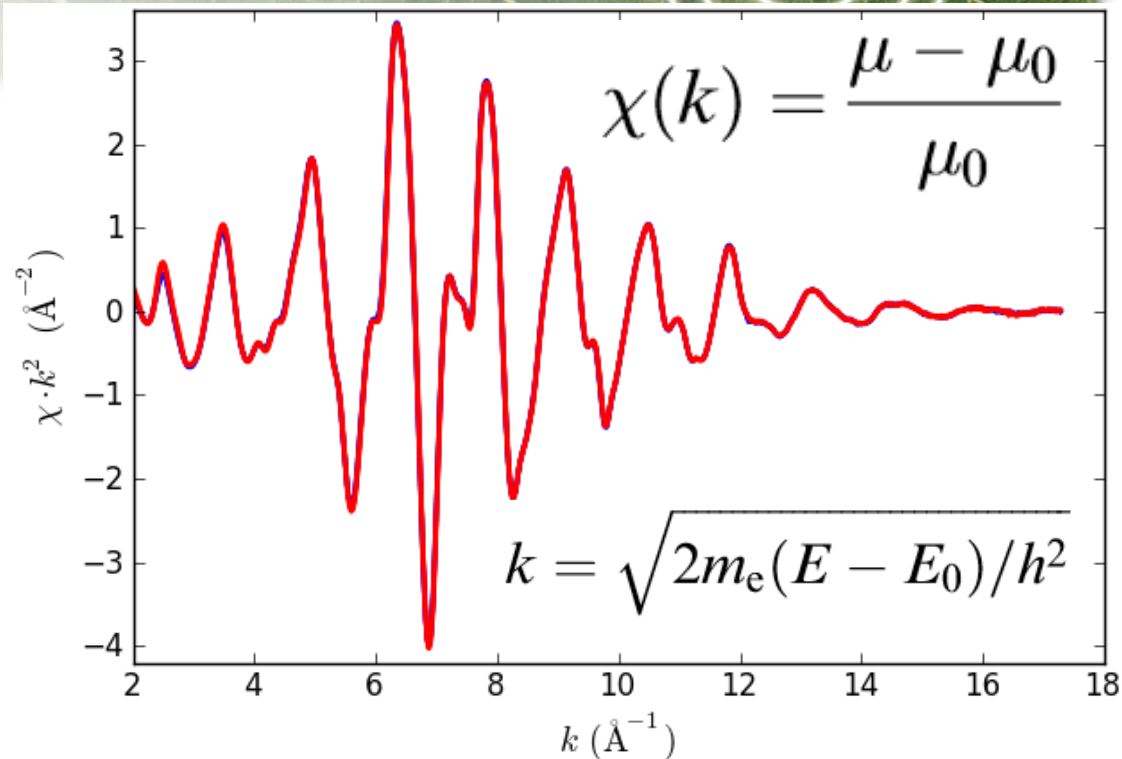
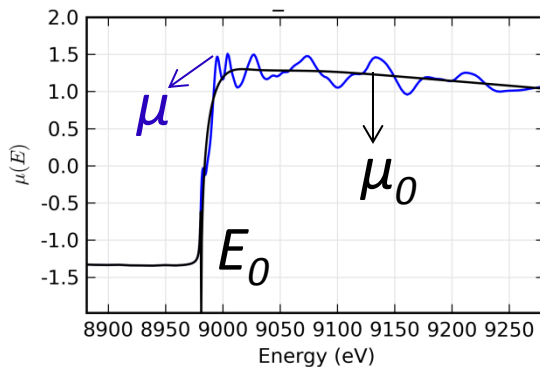
Single  
scattering  
region



# EXAFS:

## The $\chi(k)$ function

The structural information are encoded in the amplitude, shape, phase, and frequency of the oscillations represented by  $\chi$ .

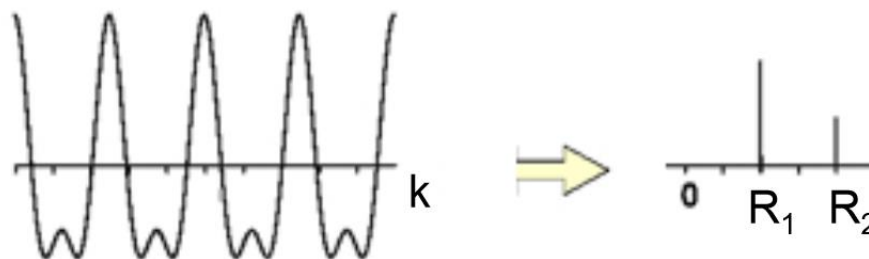
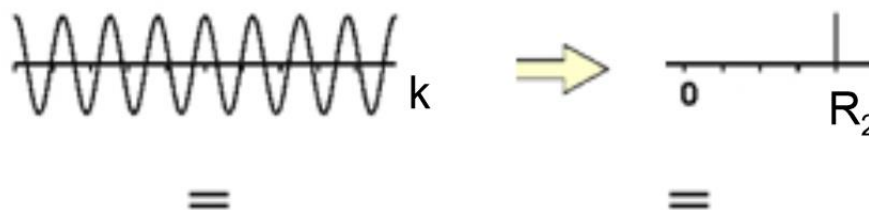
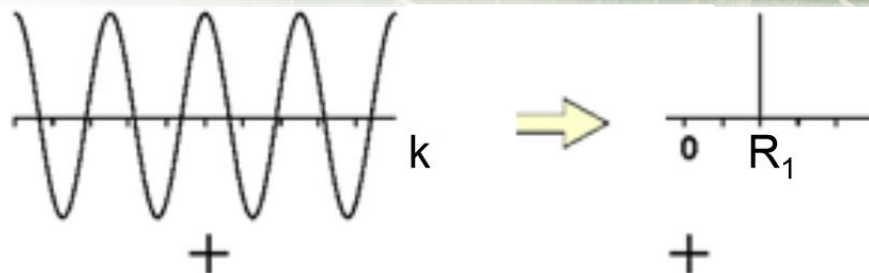


For a single absorber–scatterer pair the interferences will give rise to sinusoidal oscillations in  $\mu$  if the energy is given in units proportional to the photoelectron wavevector,  $k$ , and the threshold energy,  $E_0$ , is the binding energy of the photoelectron.



# EXAFS:

## The Fourier transform of $\sin(2kR)$



Interference gives  
an oscillating  
function  
of wavevector  $k$

Fourier Transform  
Gives a peaked  
function of  
distance  $R$

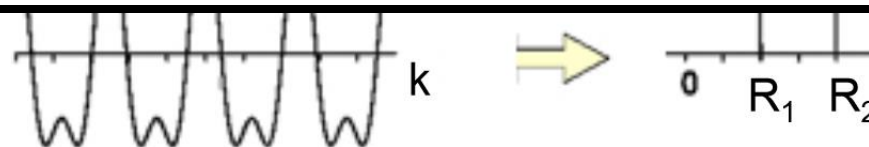
# EXAFS:

## The Fourier transform



A good Fourier Transform requires large value of  $k$ ,  
hence large range of  $E$

scan X-ray energy up to  $E_0 + 1000$  eV  
photo-electron  $k$  up to  $16 \text{ \AA}^{-1}$



Interference  
an oscillatory  
function of wave

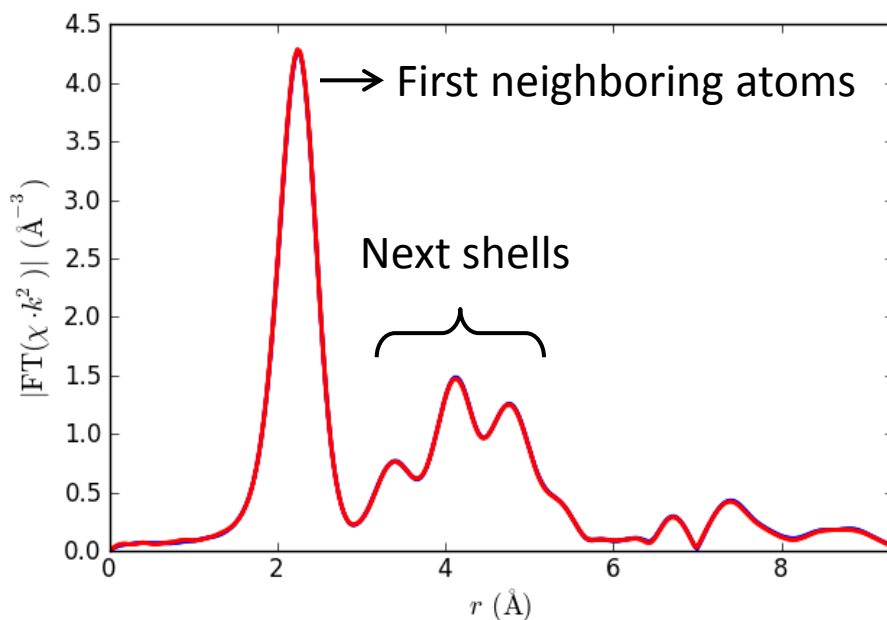
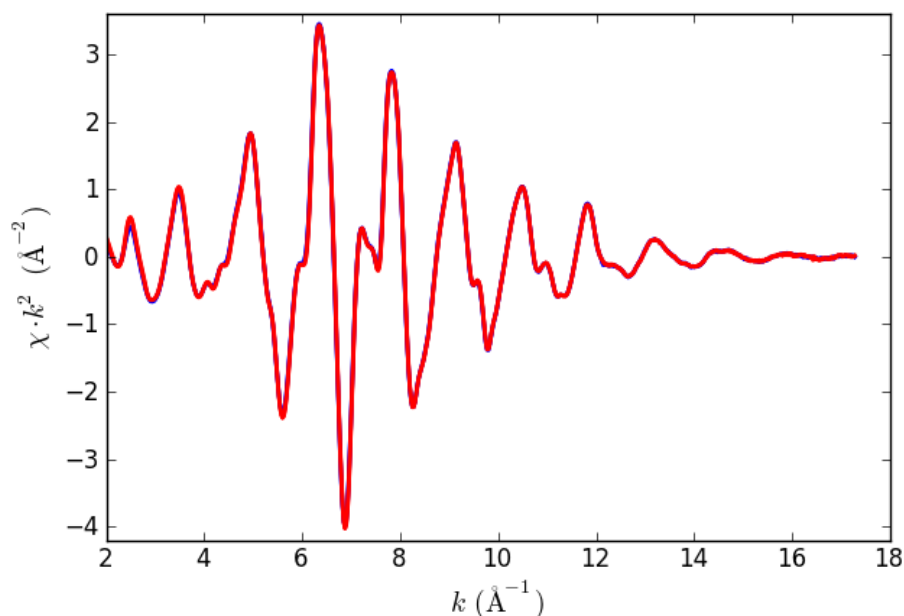
transform  
peaked  
function of  
distance  $R$

# EXAFS:

## The Fourier transform



Fourier transformation can be used to decompose a k-space signal into its different constituent frequencies. The Fourier transform (FT) of an EXAFS spectrum gives a pseudo-radial distribution function.



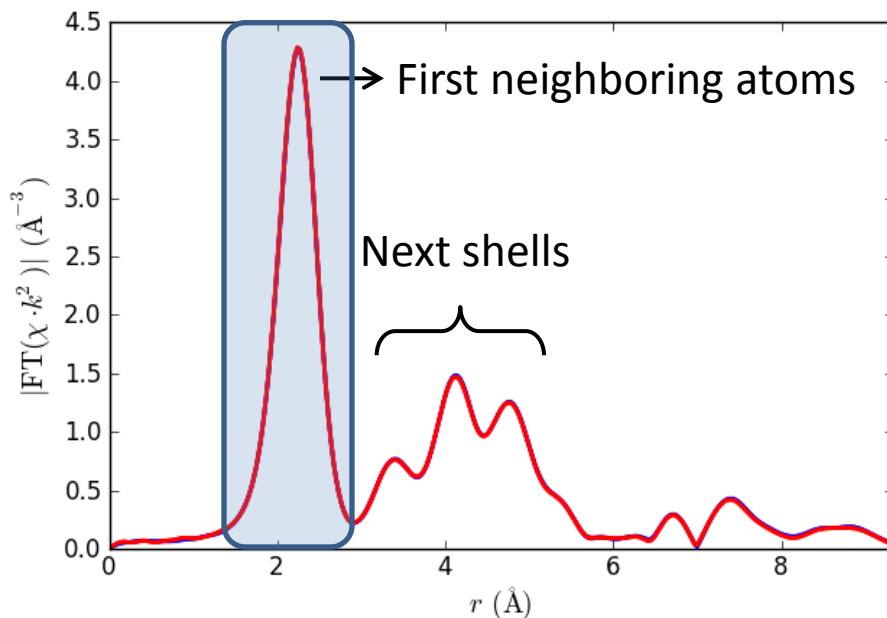
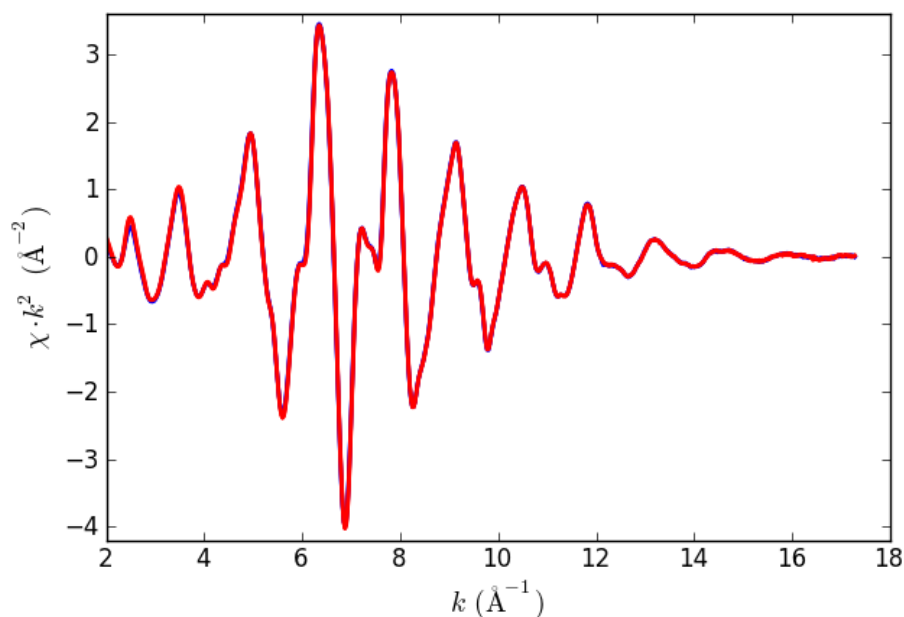


# EXAFS:

## Filtered back Fourier analysis



First shell parameters can be obtained by fitting directly the filtered back Fourier transform.

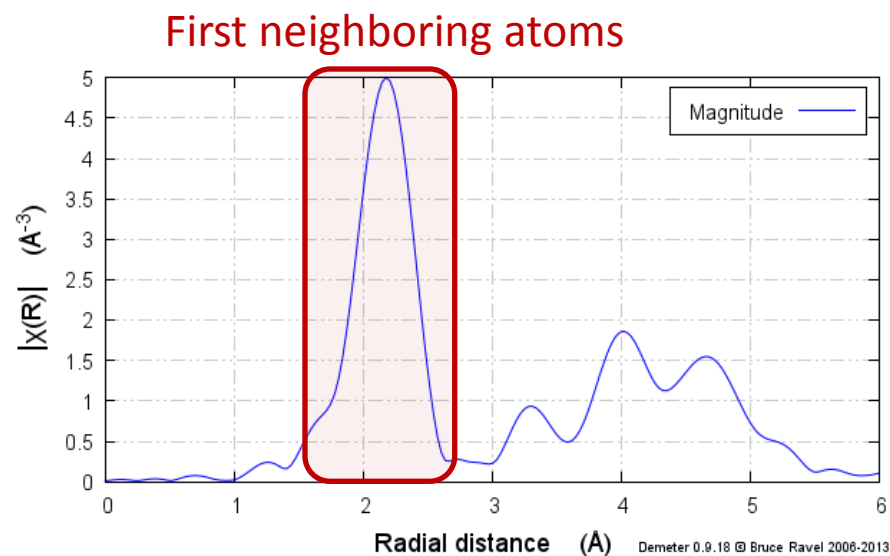
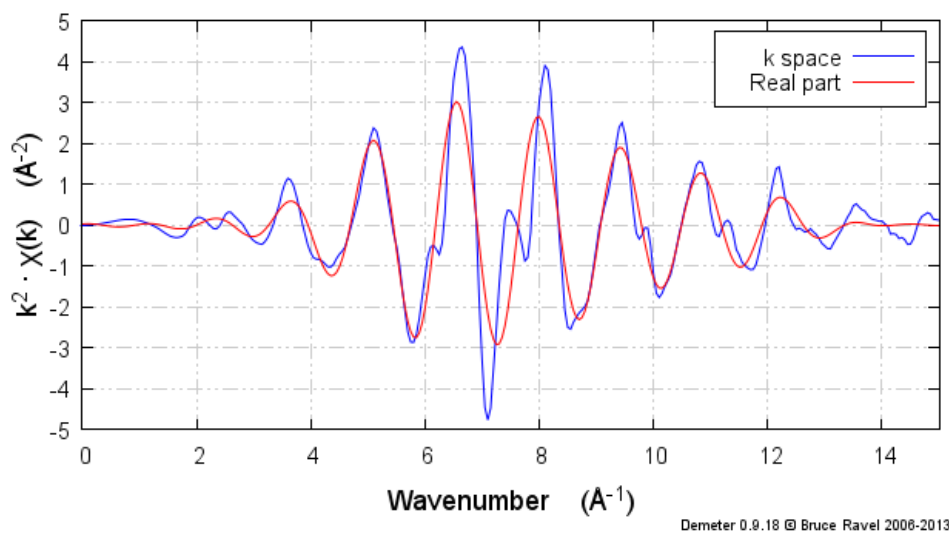


# EXAFS:

## Filtered back Fourier analysis



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# EXAFS:

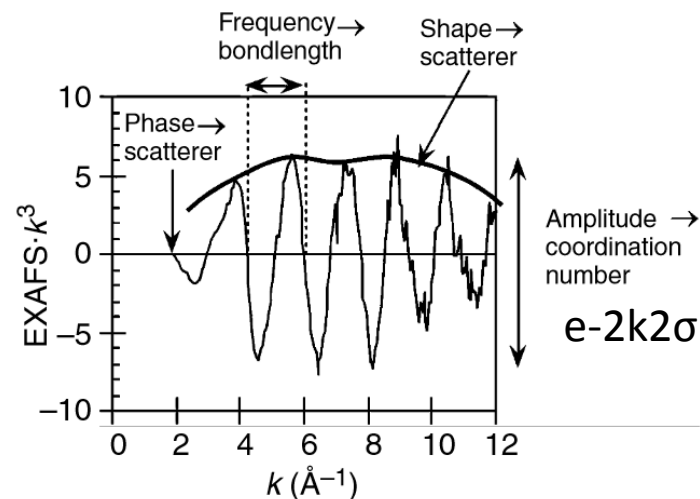
## The $\chi(k)$ function modeling



$$\chi(k) = \frac{\mu - \mu_0}{\mu_0} \quad k = \sqrt{2m_e(E - E_0)/\hbar^2}$$

$$\chi(k) = \sum_i \frac{N_i S_0^2}{k R_i^2} f_i(k, R_i) e^{-\frac{2R_i}{\lambda}} e^{-2k^2 \sigma_i^2} \sin[2kR_i + \delta_i(k)]$$

The amplitude of the oscillations is proportional to the number of scattering atoms  $N_i$ , the frequency of the oscillations is inversely proportional to the absorber–scatterer distance  $R_i$ , and the shape of the oscillations is determined by the energy dependence of the photoelectron scattering, which depends on the identity of the scattering atom.





# EXAFS:

## The $\chi(k)$ function modeling

$$\chi(k) = \frac{\mu - \mu_0}{\mu_0} \quad k = \sqrt{2m_e(E - E_0)/\hbar^2}$$

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$N_i = 3 \cos^2 \vartheta_i$   $\longrightarrow$  number of neighboring atoms at a distance  $R_i$  at an angle  $\vartheta_i$  with respect to the x-ray beam polarization.

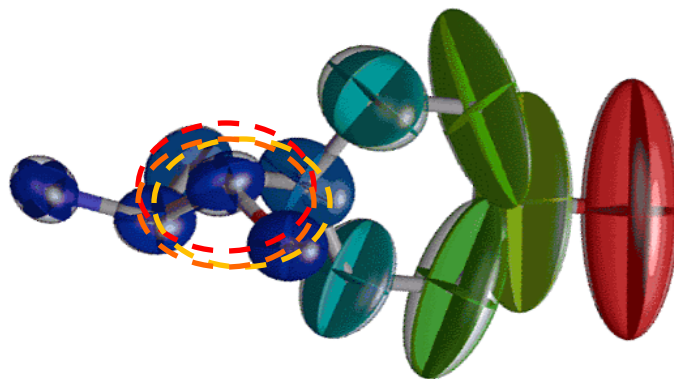
- $S_0^2$   $\longrightarrow$  passive electrons reduction factor (to allow for inelastic loss processes)
- $f_i(k, R_i)$   $\longrightarrow$  backscattering amplitude (energy dependence of the photoelectron scattering)
- $\lambda$   $\longrightarrow$  photoelectron mean free path
- $\delta_i$   $\longrightarrow$  phase shift (that the photoelectron wave undergoes when passing through the potential of the absorbing and scattering atoms)
- $\sigma_i^2$   $\longrightarrow$  correlated Debye–Waller factor (root-mean-square deviation in absorber–scatterer distance)

# EXAFS:

Static and dynamic disorder

Temperature dependence of the DW

$$\chi(k) = \sum_i \frac{N_i S_0^2}{k R_i^2} f_i(k, R_i) e^{-\frac{2R_i}{\lambda}} e^{-2k^2 \sigma_i^2} \sin[2kR_i + \delta_i(k)]$$



$$\sigma^2 = \sigma_0^2 + \sigma^2(T)$$

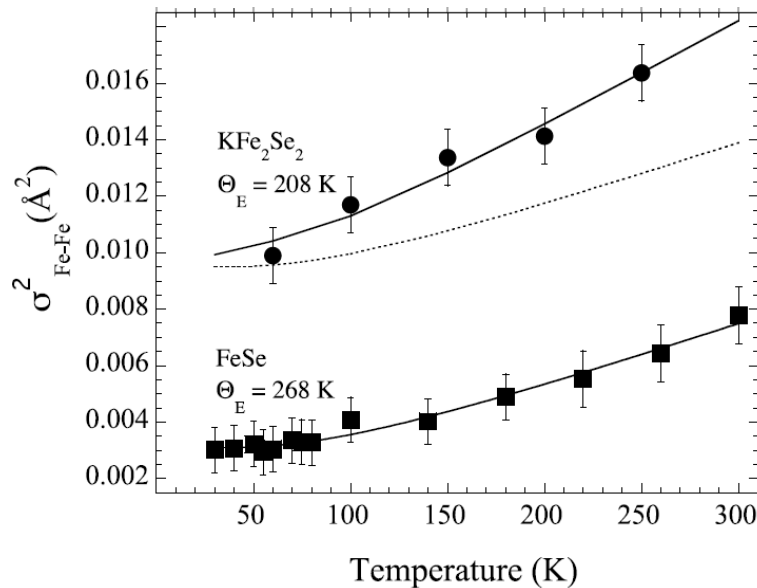
$$\sigma^2(T) = \frac{\hbar^2}{2\mu k_B \theta_E} \coth \frac{\theta_E}{2T}$$

# EXAFS:

Static and dynamic disorder

Temperature dependence of the DW

$$\chi(k) = \sum_i \frac{N_i S_0^2}{k R_i^2} f_i(k, R_i) e^{-\frac{2R_i}{\lambda}} e^{-2k^2 \sigma_i^2} \sin[2kR_i + \delta_i(k)]$$



$$\sigma^2 = \sigma_0^2 + \sigma^2(T)$$

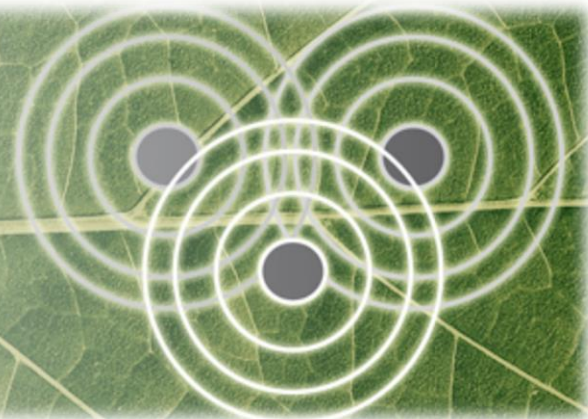
Einstein's formula:

$$\sigma^2(T) = \frac{\hbar^2}{2\mu k_B \theta_E} \coth \frac{\theta_E}{2T}$$

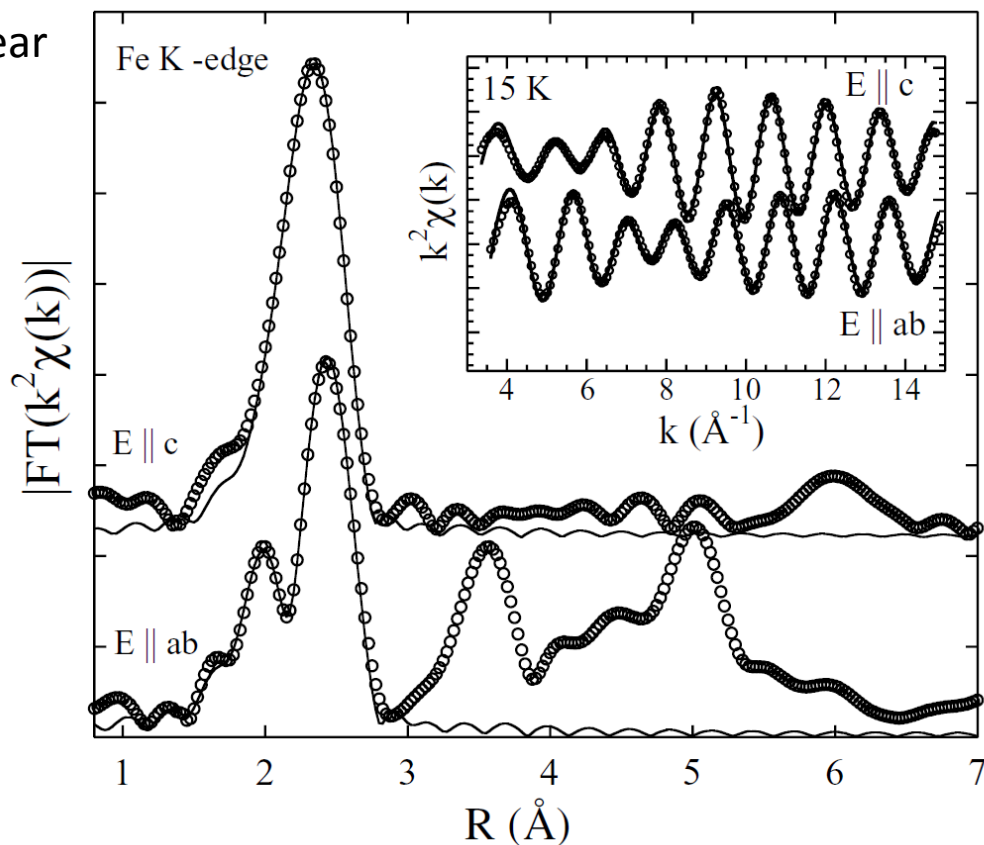
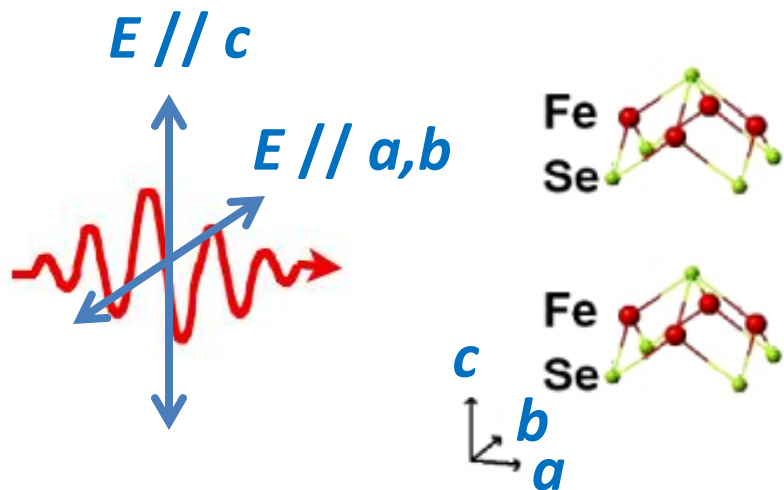


# EXAFS:

## Polarization analysis



The contributions of Fe-Se and Fe-Fe appear mixed in the  $E_{ab}$  polarization, While only the Fe-Se contribute to the  $E_c$  main peak in the Fourier transform.



# EXAFS & XANES:

## Complementary information



## XANES

- **Oxidation state**
- **Unoccupied electronic states**
  - **Spin state**
  - **Local structure**
- **direct information about bond angles.**

## EXAFS

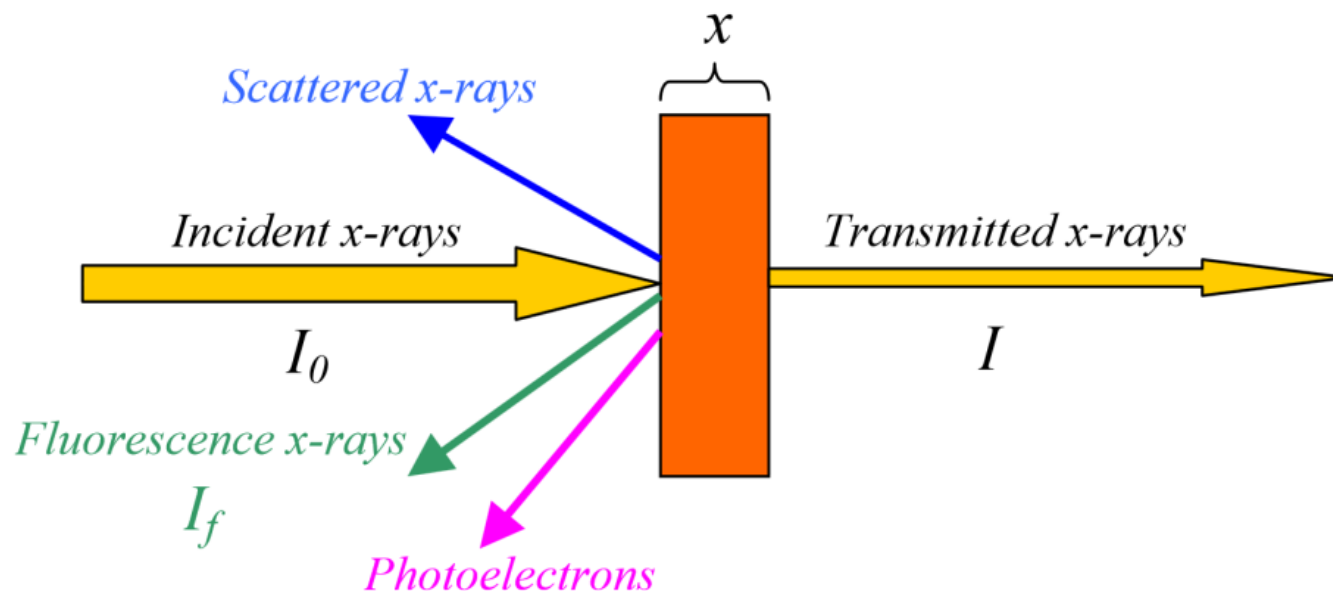
- **Bond distances**
- **Coordination number**
- **Static and dynamic disorder**

# XANES & EXAFS:

## Detection methods:

### Transmission & fluorescence modes

Absorption of an ionizing X-ray results in photoelectron ejection, leaving behind a highly excited core-hole state. This can relax by a variety of mechanisms, with the two most important being emission of an Auger electron and X-ray fluorescence.





# XANES & EXAFS:

Detection methods:

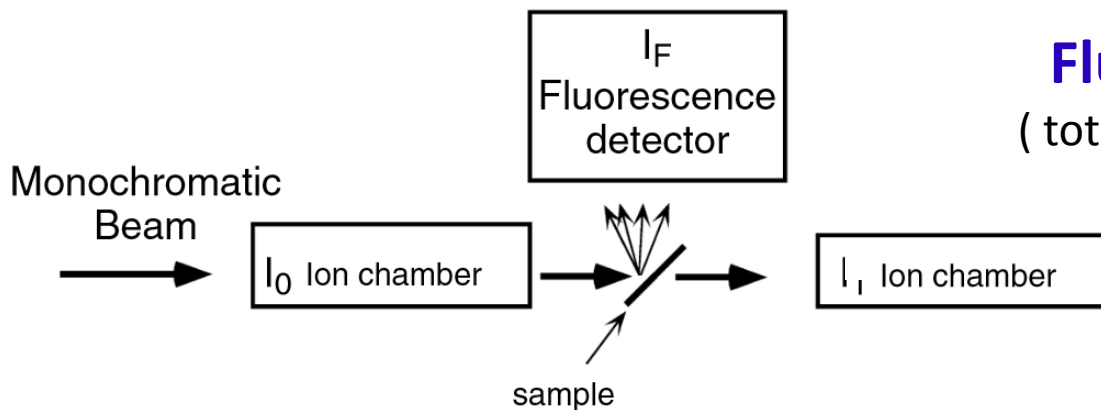
Transmission & fluorescence modes

Providing the sample is dilute or thin (total absorbance  $\ll 1$ ), the intensity of the fluorescence X-rays is proportional to the X-ray absorption cross-section.

In order to have good sensitivity, the fluorescence detector needs some kind of energy resolution to distinguish between the signal and background X-rays.

**Transmission:**  $\mu x = \ln(I_0 / I_1)$

**Fluorescence:**  $\mu \propto (I_F / I_0)$   
( total absorbance  $\ll 1$   $e^{-\mu x} \rightarrow 1 - \mu x$  )



# XANES & EXAFS:

Fluorescence method and the self absorption effect

## FLUORESCENCE MODE:

$\mu \propto (I_F / I_0)$  is only true in the limit of very thin or very dilute samples

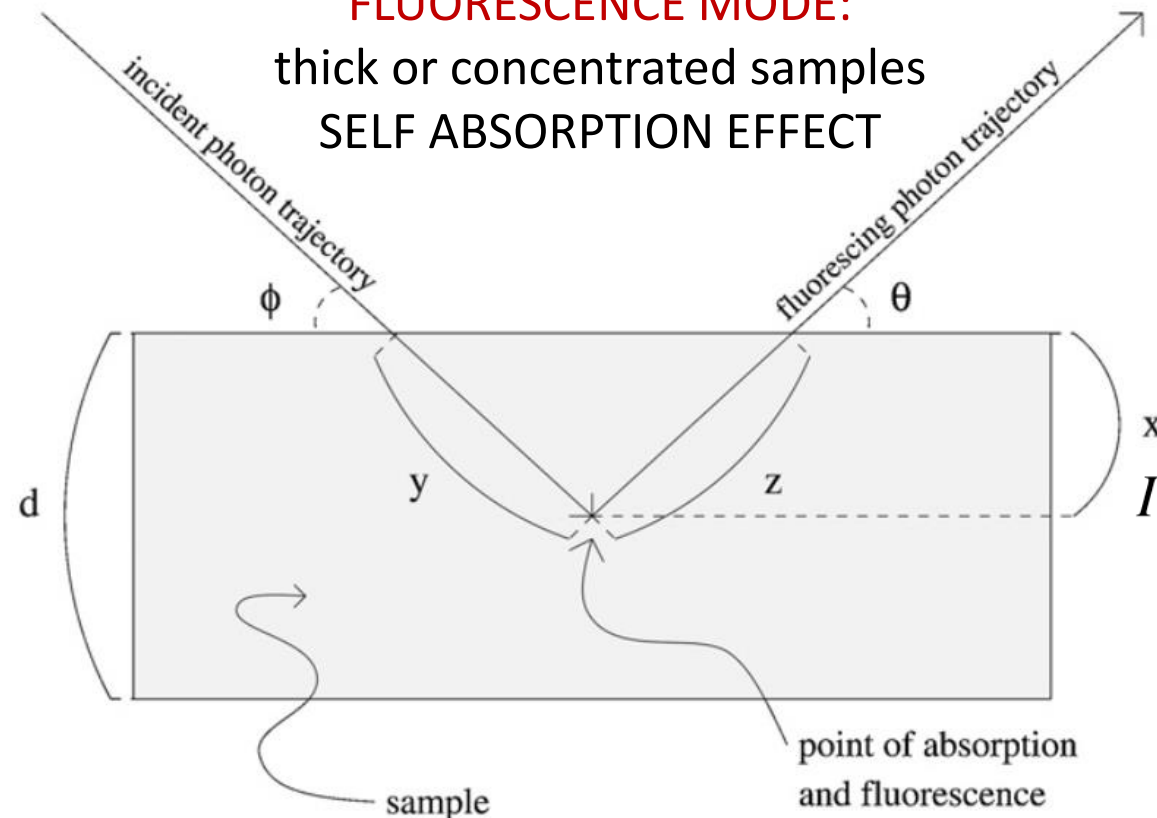
**What is happening for thick or concentrated samples?**

# XANES & EXAFS:

## Fluorescence method and the self absorption effect

### FLUORESCENCE MODE:

thick or concentrated samples  
SELF ABSORPTION EFFECT



The depth into which the incident beam can penetrate changes as fine structure of  $\mu(E)$  changes. As the oscillatory part wiggles up, the penetration depth diminishes. As it wiggles down, the depth increases.

This serves to attenuate the oscillatory structure.

$$I_f = I_0 e^{-\mu(E)y} e^{-\mu(E_f)z} \epsilon_a(E) \mu_a(E)$$

$E_f$  = energy of the fluorescing photon

$\epsilon_a$  = fluorescence efficiency per unit solid angle

$\mu_a$  = absorption due to the given core excitation of the absorbing atom



# XANES & EXAFS:

## Fluorescence method: the self absorption effect



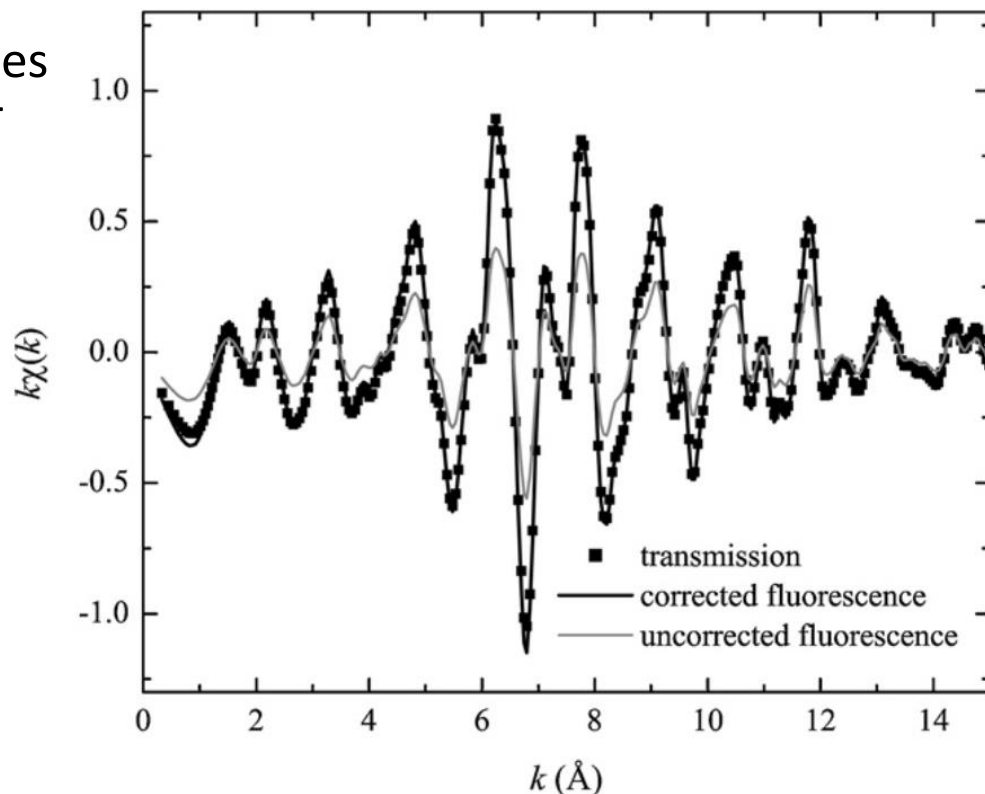
### FLUORESCENCE MODE:

thick or concentrated samples  
SELF ABSORPTION EFFECT

$$\chi = \frac{\chi_{exp}}{1 - \frac{\bar{\mu}}{\alpha} \chi_{exp} - \frac{\bar{\mu}_a}{\alpha}}$$

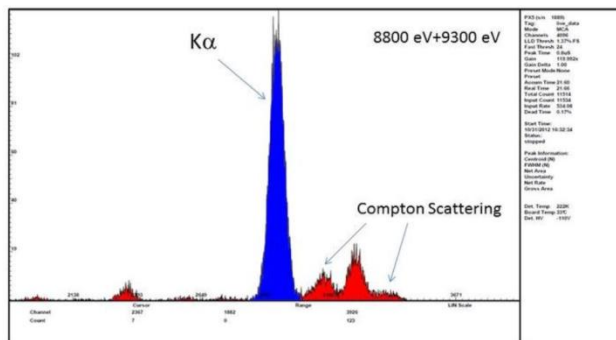
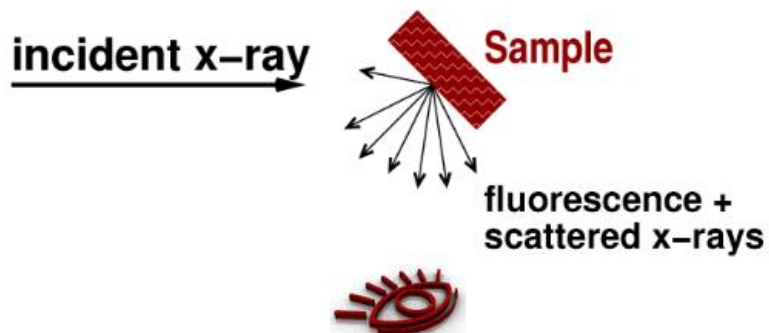
$$\alpha \equiv \bar{\mu}_T + g\mu_f$$

$$g \equiv \sin \phi / \sin \theta$$

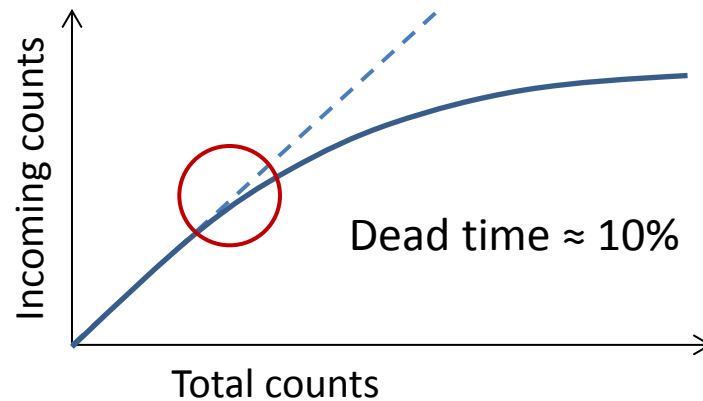


# XANES & EXAFS:

## Fluorescence method: detector dead time

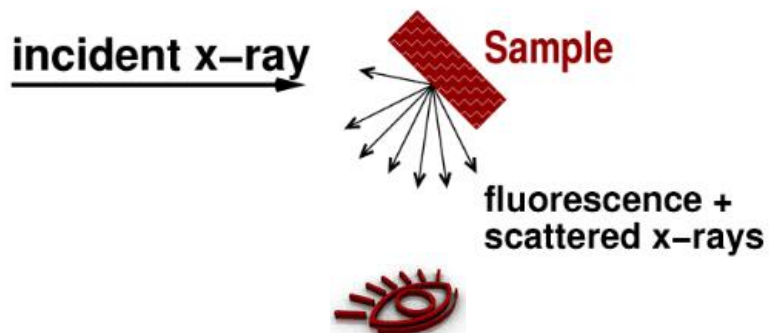


Dead time: the electronic energy discrimination takes a finite amount of time, which limits the total amount of signal that can be processed. When the count rates are exceeded, the detector is effectively unable to count all the fluorescence for some fraction of the time. The limit of total intensity incident on these detectors can limit the quality of the measured XAFS.



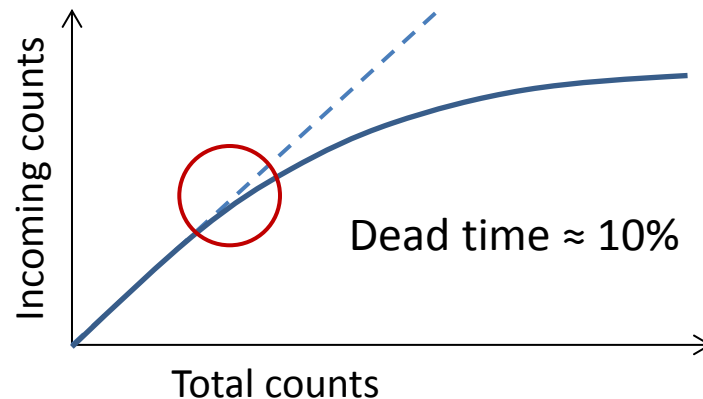
# XANES & EXAFS:

## Fluorescence method: detector dead time



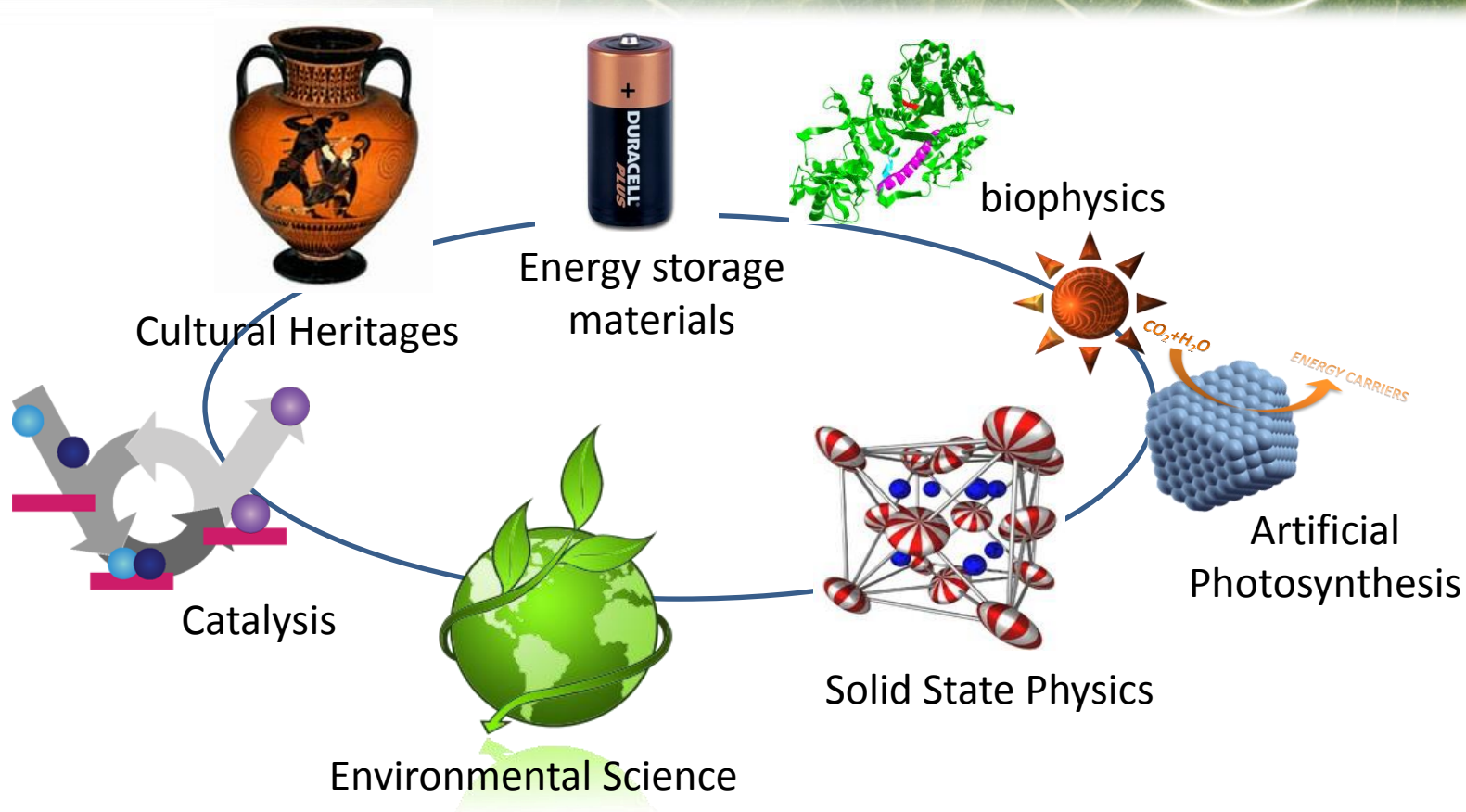
*(E discrimination can potentially allow to suppress the scatter peak and other fluorescence lines, collecting only the intensity of the fluorescence lines of interest.)*

Dead time: the electronic energy discrimination takes a finite amount of time, which limits the total amount of signal that can be processed. When the count rates are exceeded, the detector is effectively unable to count all the fluorescence for some fraction of the time. The limit of total intensity incident on these detectors can limit the quality of the measured XAFS.





# Research fields: To give an idea...



# Why so many?

To give an idea...



**XAS can be applied to any  
kind of materials:  
Crystals, glass, liquids, etc...**



# SUMMARY

## XAS (XANES/EXAFS)

element sensitive local probe that can be applied to any kind of material

### COMPLEMENTARY INFORMATION

#### XANES

- Oxidation state
- Unoccupied electronic states
  - Spin state
  - Local structure
- direct information about bond angles.

#### EXAFS

- Bond distances
- Coordination number
- Static and dynamic disorder



# REFERENCES

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