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THE ROLE OF ANHARMONICITY ON THE THERMOELECTRIC PROPERTIES OF HALF-HEUSLER ALLOYS

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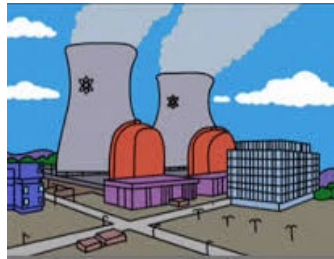


WHY THERMOELECTRIC MATERIALS?

Thermal power plant



Nuclear power plant



Waste incinerator



Factories



Natural gas



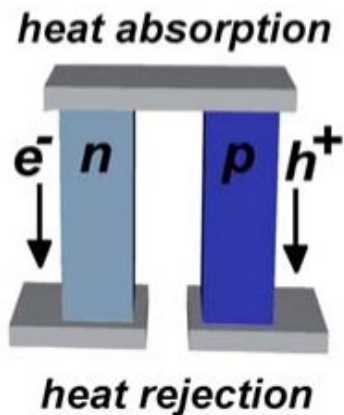
Automobile



Energy Loss/Waste Heat
66%

Energy Used
34%

Recovery



Seebeck effect

Thermoelectrics

Electric energy

MOTIVATION

To make a thermoelectric device competitive, an average $ZT > 2$ (in the application T-range) is required.

• Why half-Heusler alloys?

Pb-based alloys

high ZT, but high toxicity and weak mechanical strength.

Skutterudites

high ZT, but poor thermal stability and limited supply of rare-earth elements.

Half-Heusler alloys

environmentally friendly and mechanically and thermally robust.

OBJETIVE:

- Increase ZT

Reduction of the lattice contribution to the thermal conductivity: κ_L

Increase of the electrical contribution to transport: $PF = S^2\sigma$

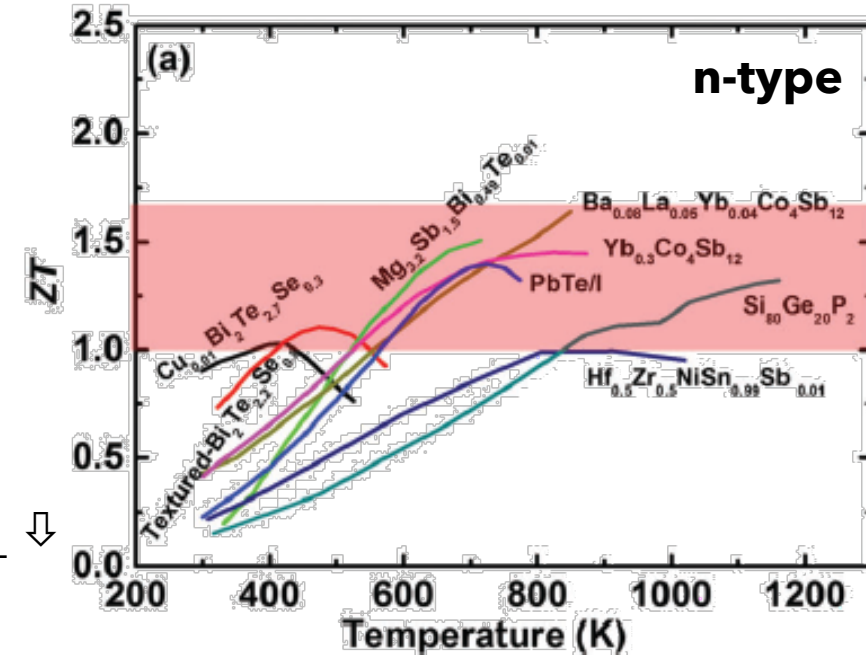
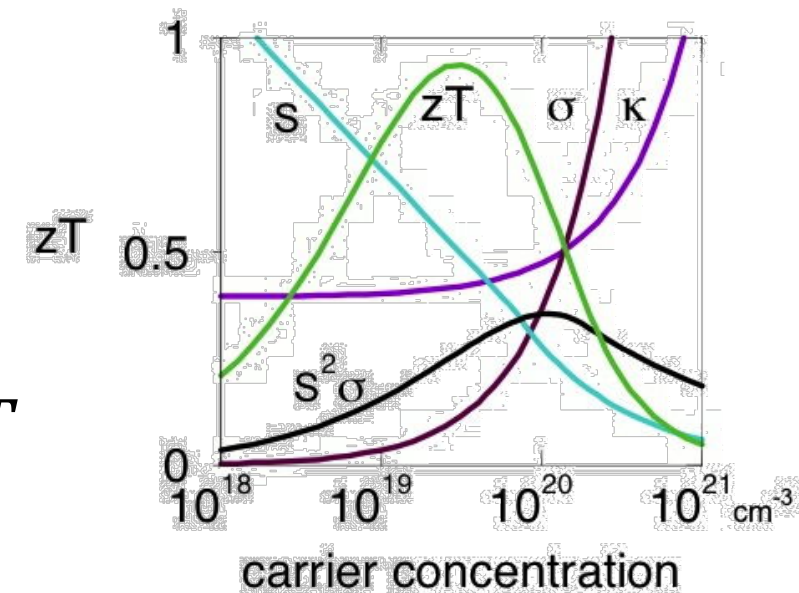
Nanostructures

↑ Phonon scattering → Thermal conductivity κ_L ↓

Doping

Modification of the charge carrier density

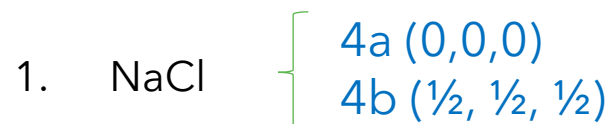
$$ZT = \frac{S^2\sigma}{\kappa_L + \kappa_e} T$$



HALF-HEUSLER ALLOYS

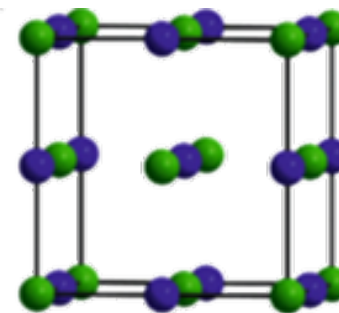
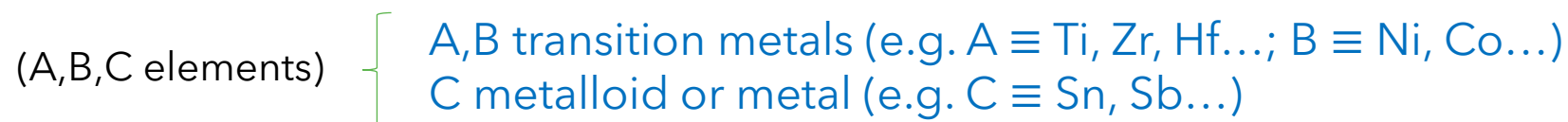
Half-Heusler (HH) alloys can be derived by combining a rocksalt- and a zincblende-type crystal structures.

Wickoff positions:

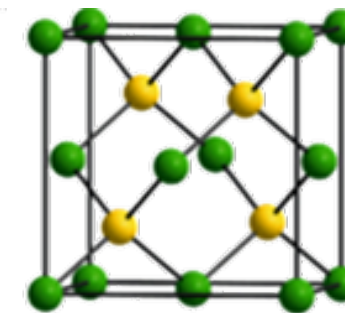


and $4d (\frac{3}{4}, \frac{3}{4}, \frac{3}{4})$ vacant.

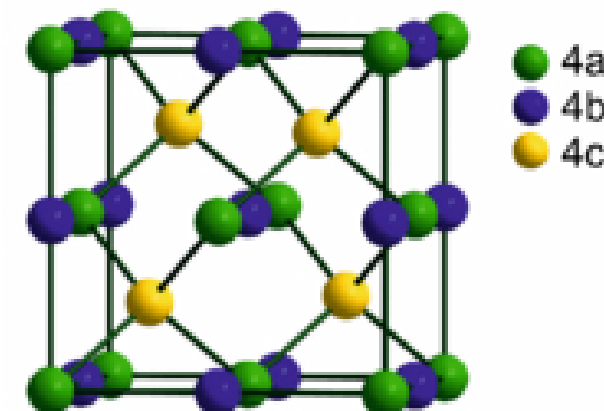
Space group: F-43m (no. 216); stoichiometry: 1:1:1 (ABC)



rock salt-type structure



zinc blende-type structure



Half-Heusler structure

EXPERIMENTAL TECHNIQUES

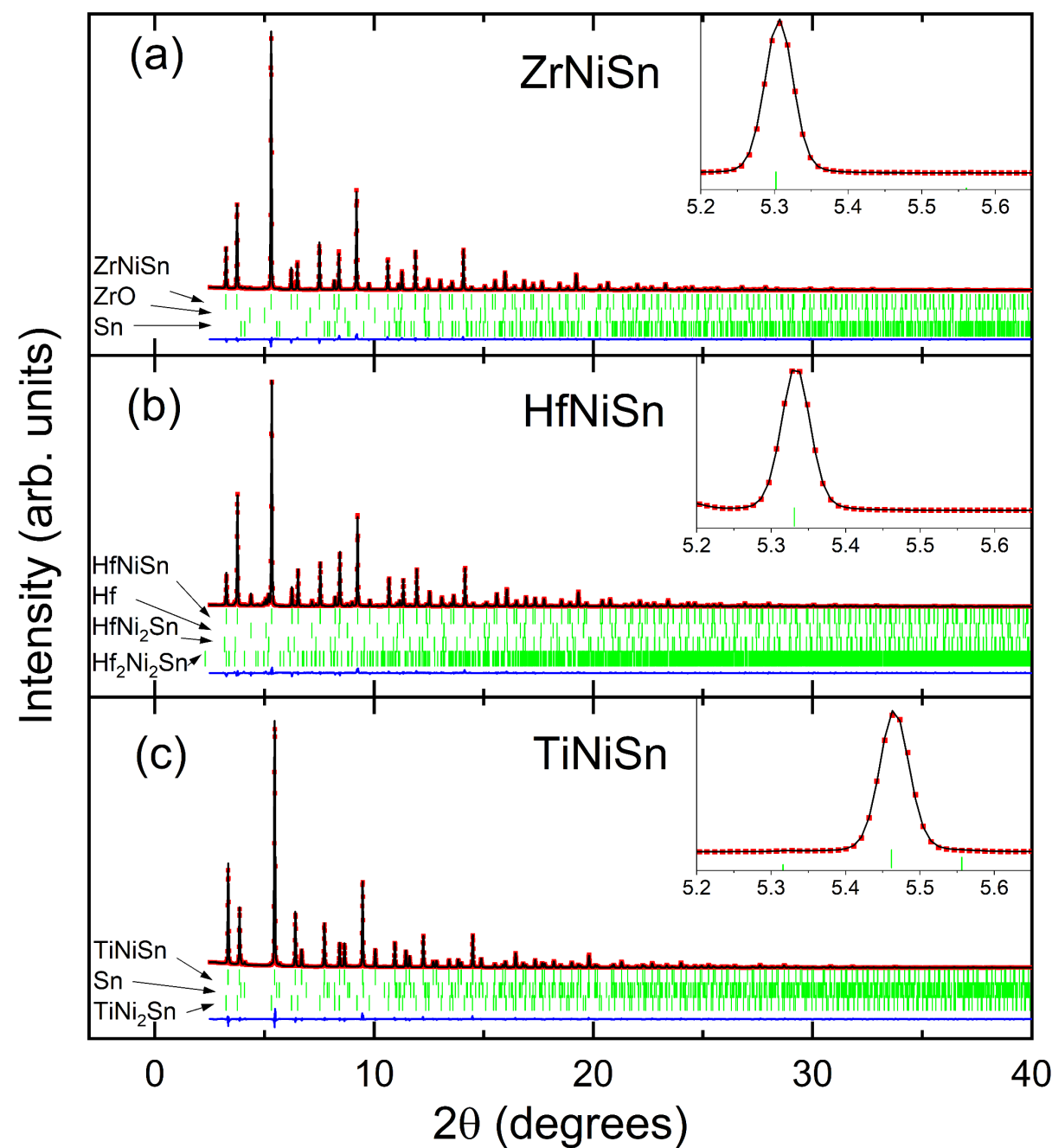
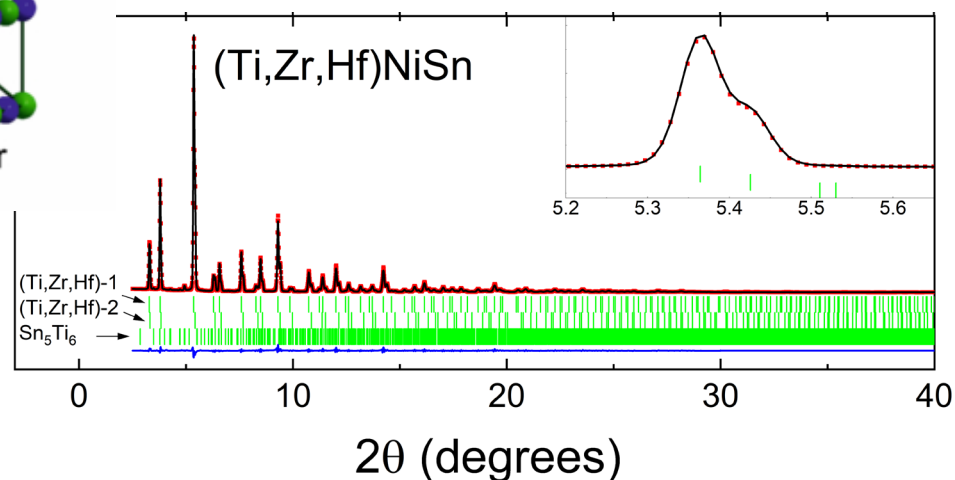
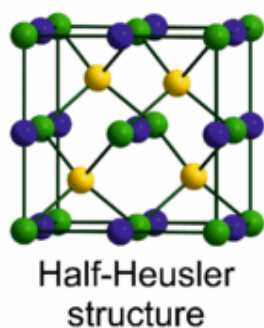
(Ti,Zr,Hf)NiSn

Arc melting + Thermal annealing @ 900°C, 1week – water quenched



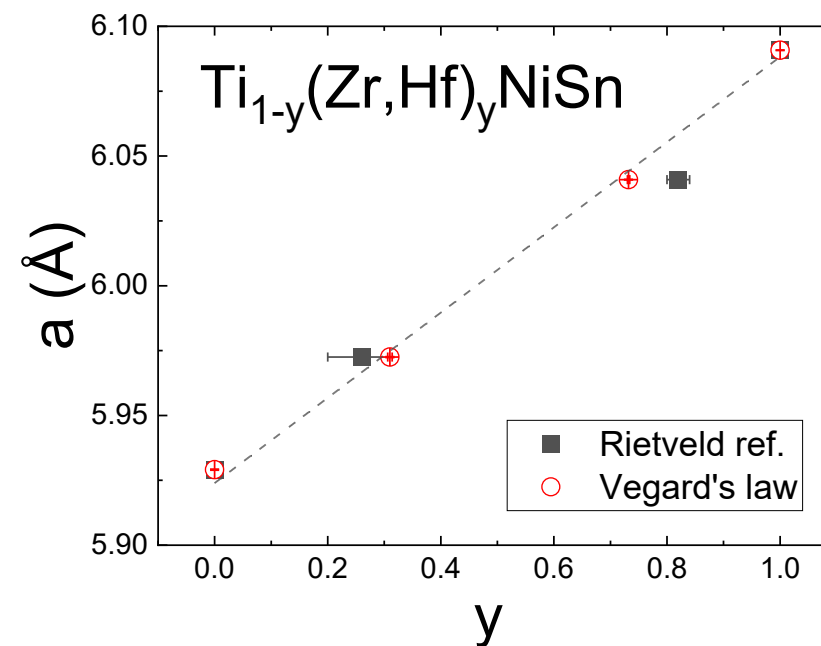
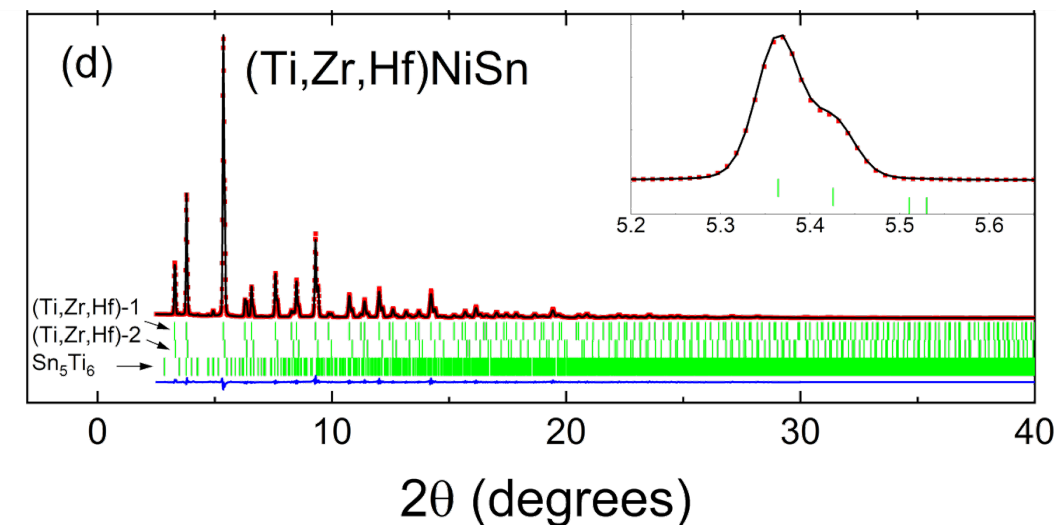
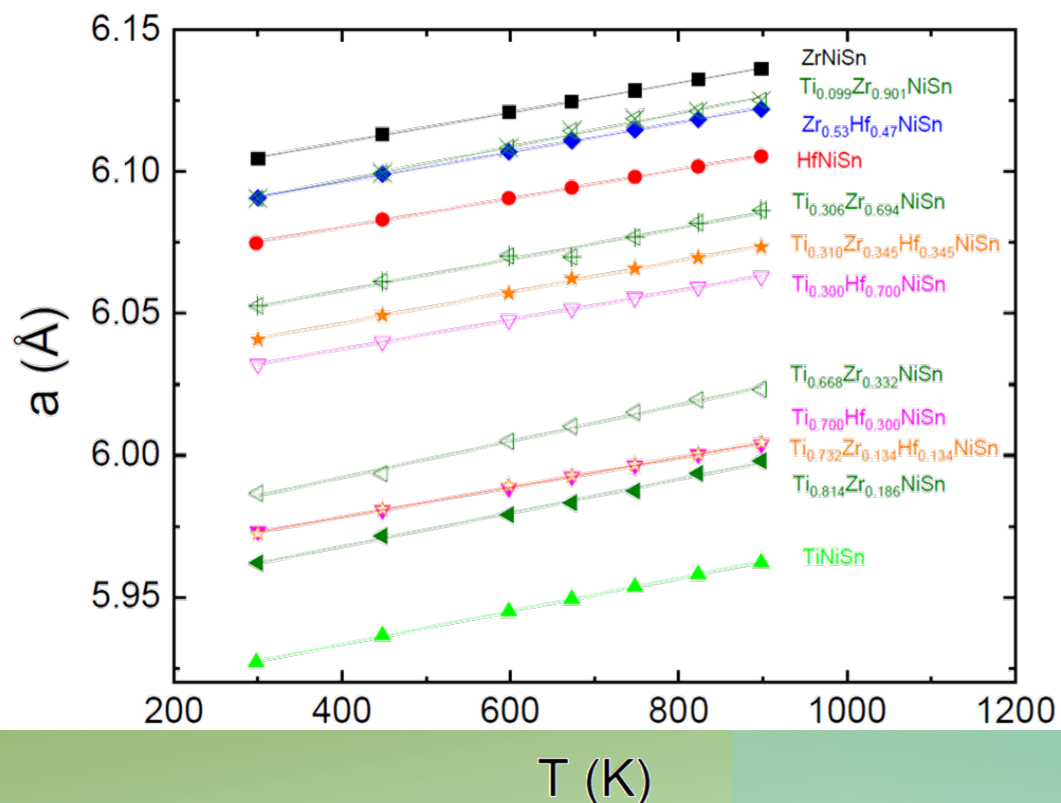
STRUCTURAL CHARACTERIZATION

- High resolution synchrotron X-ray diffraction
- **ID22@ESRF** (Grenoble, France), $\lambda = 0.2 \text{ \AA}$.
- Single hH phase - ternary alloys
- Segregation of hH phases - pseudoternary alloys
- Minor secondary phases (< 3 wt.%)



STRUCTURAL CHARACTERIZATION

- Linear coefficient of **thermal expansion** (CTE) for every hH phase.
- $\alpha = 8.4 - 9.8 \times 10^{-6} K^{-1} \rightarrow$ small expansion with T



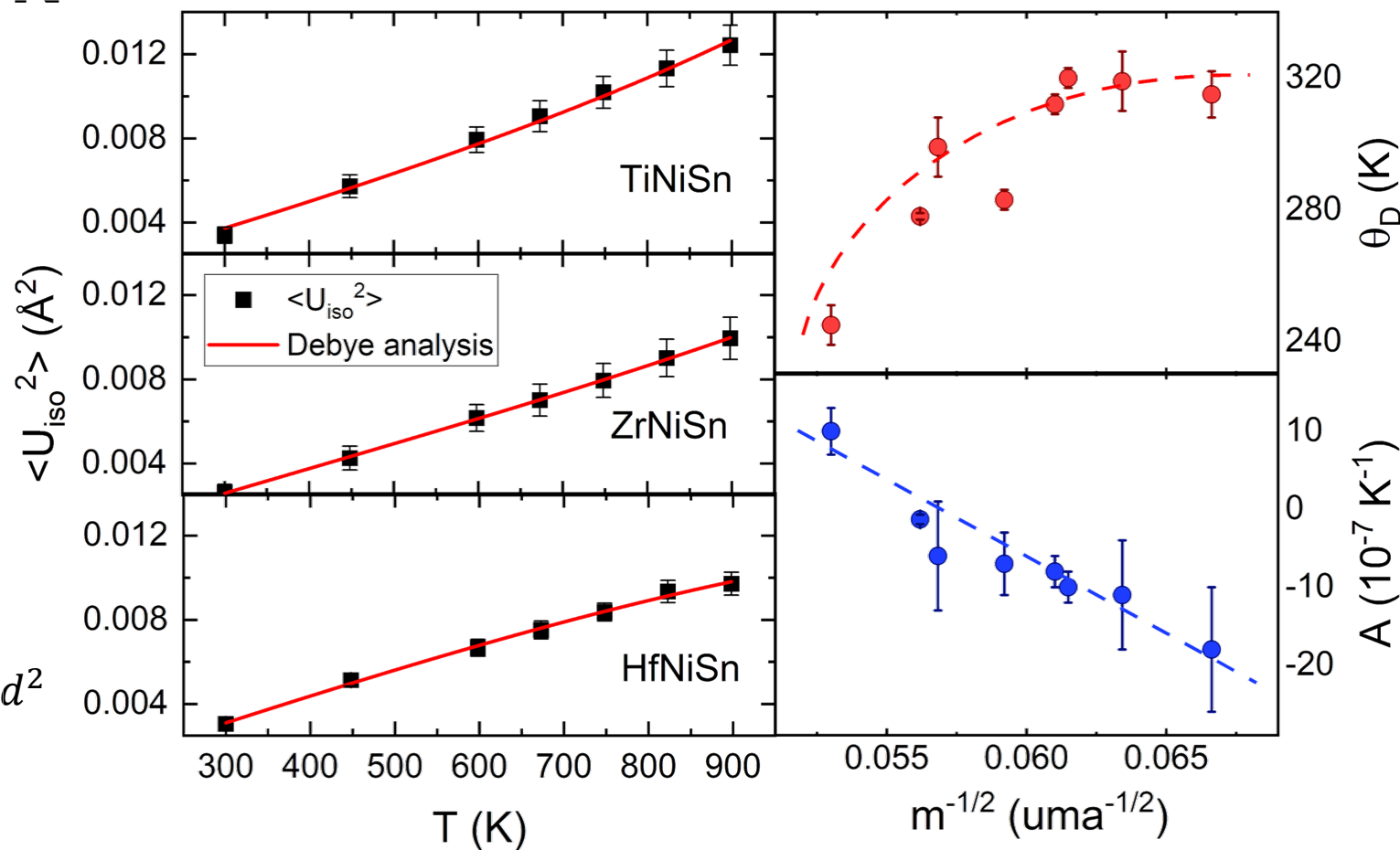
STRUCTURAL CHARACTERIZATION

- **Atomic (isotropic) displacement parameters (ADPs):**

$$B_{iso} = 8\pi^2 \langle U_{iso}^2 \rangle$$

- Debye model:
 - Simple crystal structures
 - Single mass $m \rightarrow$ weighted average of every atomic species

$$\langle U_{iso}^2 \rangle = \frac{3\hbar^2 T}{mk_B \theta_D^2} \left[\frac{T}{\theta_D} \int_0^{\frac{\theta_D}{T}} \frac{x}{e^x - 1} dx + \frac{\theta_D}{4T} \right] + d^2$$



STRUCTURAL CHARACTERIZATION

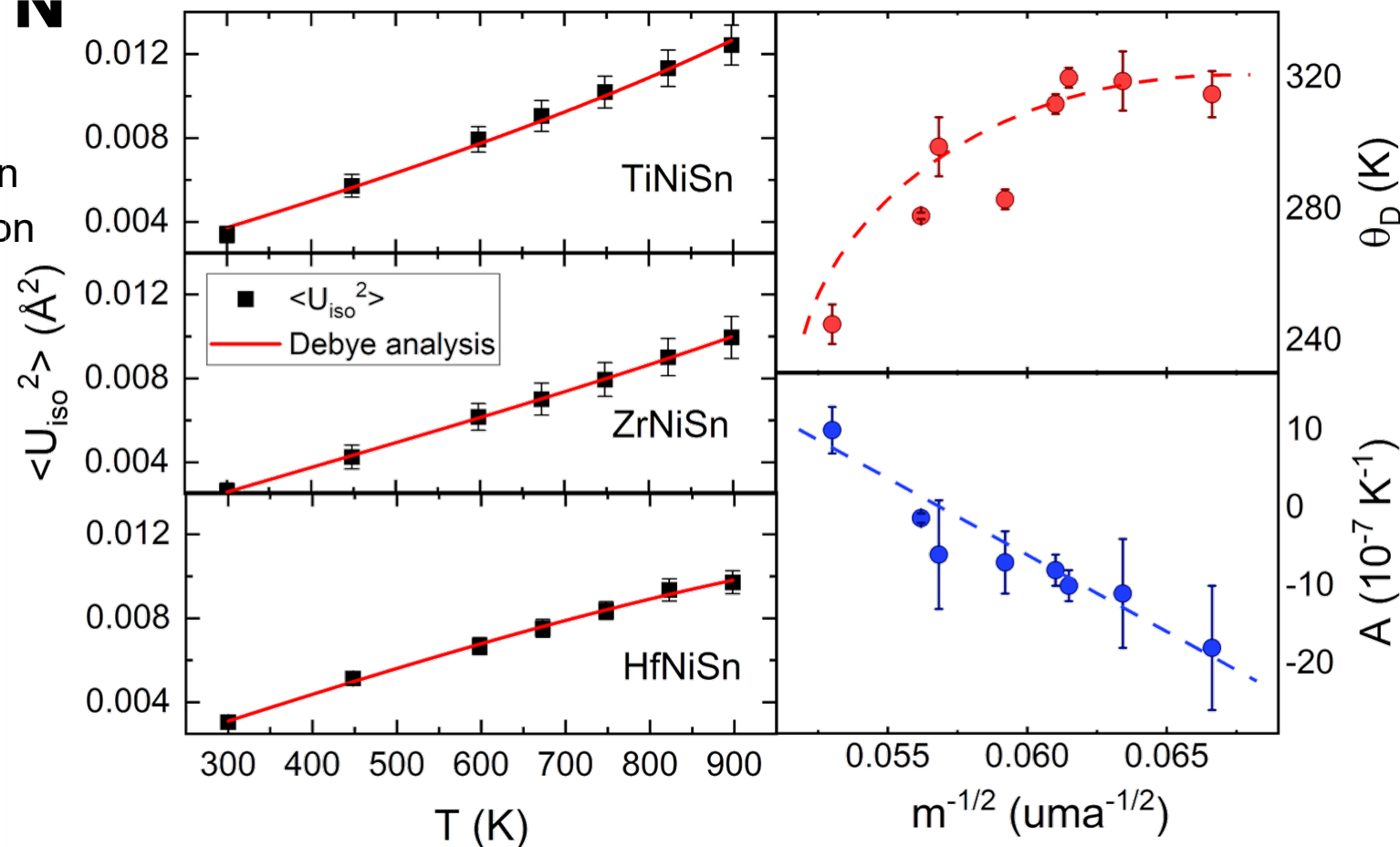
- θ_D generally depends on T
 - It can be calculated using the Thirring-Stern expansions of the quasiharmonic expression for the thermal energy θ_W^h :

$$\theta_W^h = \theta_1 + \sum_{n=1}^{\infty} b_n T^{-n}$$

(b_n coefficients)

- Also an anharmonic contribution has to be considered, so, the T-dependent Debye temperature:

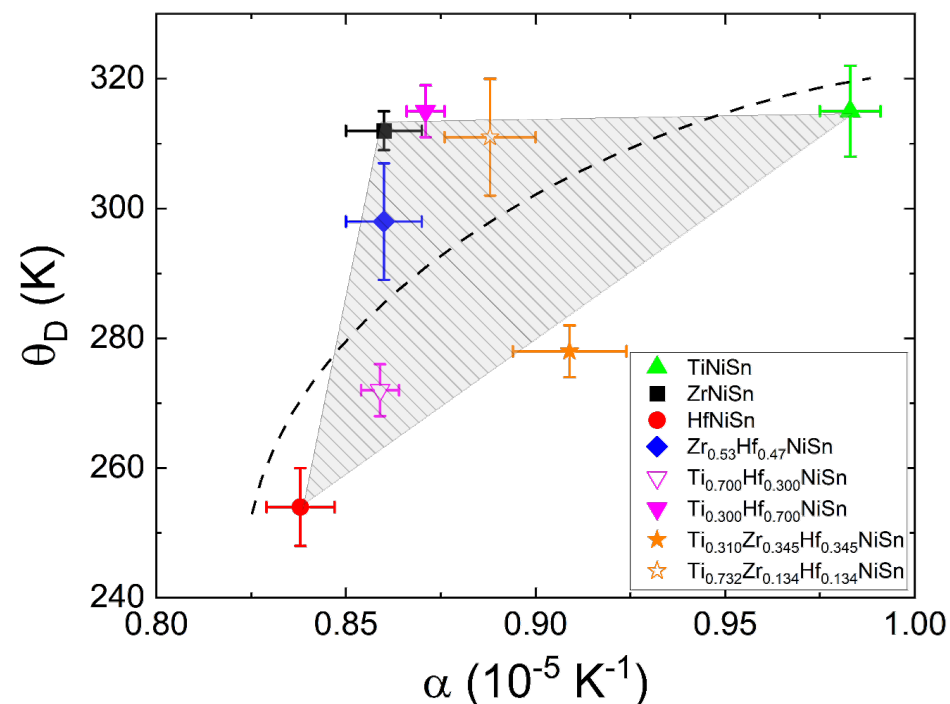
$$\theta_W = \theta_W^h + \left[\frac{1}{2} A T^2 \middle/ \frac{d \left(\frac{W_{th}^h}{3 N k_B T} \right)}{d \left(\frac{\theta_W^h}{T} \right)} \right]$$



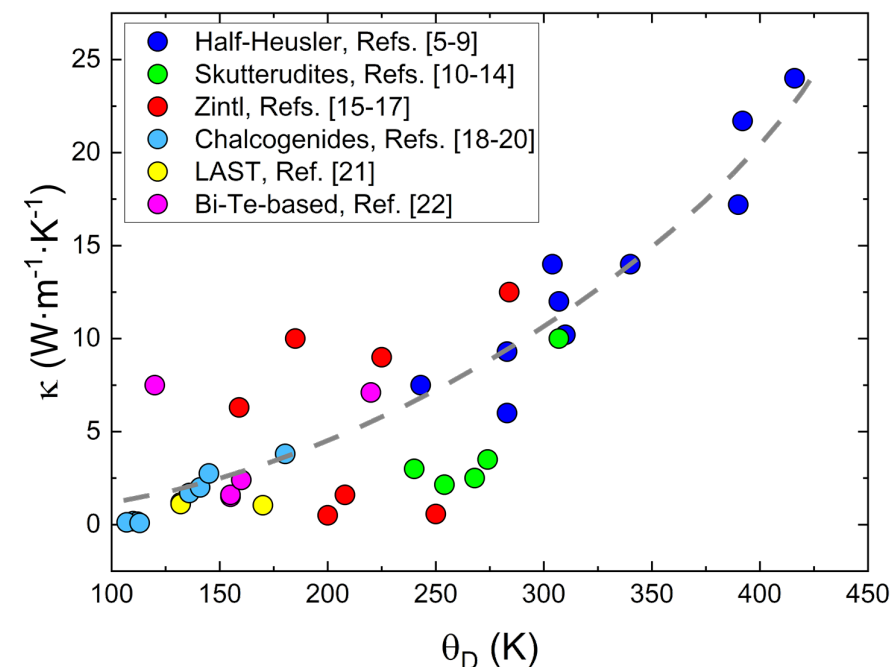
The lower $\theta_D \rightarrow$ the larger anharmonicity A

STRUCTURAL CHARACTERIZATION

- Relationship between Debye temperature and CTE:



- Implications in the thermoelectric properties: thermal conductivity

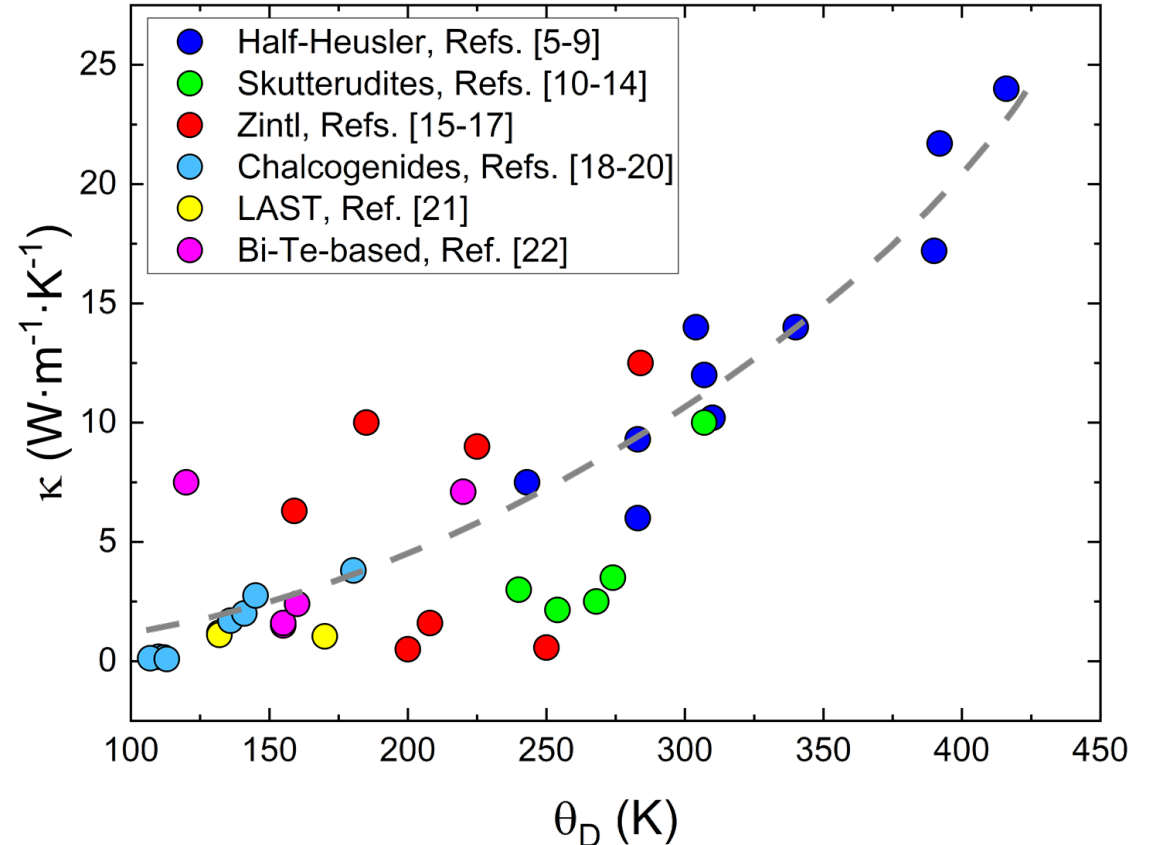


The lower CTE $\rightarrow$ the lower $\theta_D \rightarrow$ the lower $\kappa \rightarrow$ the larger ZT

The larger anharmonicity $A \rightarrow$ the larger ZT

CONCLUSIONS

- High resolution synchrotron XRD experiments give precise insight on the phase separation in isoelectronic substituted half-Heusler Alloys.
- Temperature dependent XRD can provide information of the suitability of a material to be a good thermoelectric compound.
- **The larger the anharmonicity, the larger ZT.**





T₁

H₄

A₁

N₁

K₅

S₁