

Growth of cobalt iron oxides on Ru(0001) by real-time LEEM and real-time PEEM

Juan de la Figuera



**X AUSE Conference &
VALBA User's Meeting**

5 – 8 September 2022

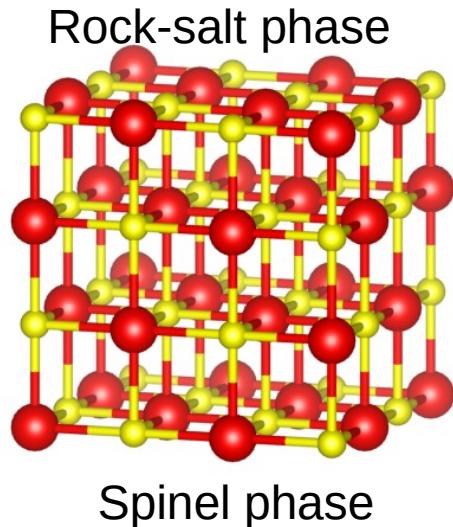
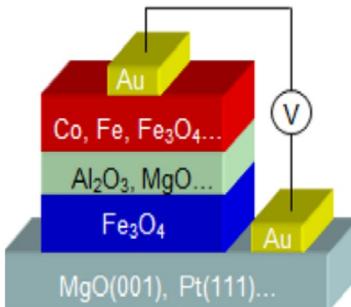
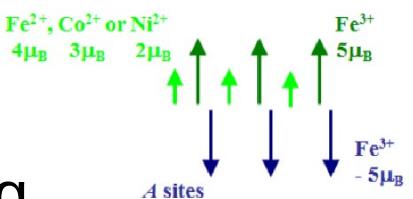
ALBA Synchrotron, Barcelona

Cobalt-iron oxides

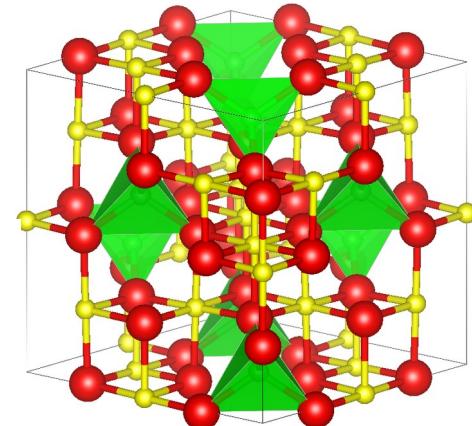
- Cobalt iron oxides show a mixture of spinel and rock salt phases
 - Rock salt (aka halite) phases: only octahedral sites, occupied by divalent cations. **Antiferromagnetic.**
 - Spinel phases: octahedral and tetrahedral sites, with divalent and trivalent cations. Mostly **ferrimagnetic**, some antiferromagnetic

They can be used for:

- Spin filters
- Conductive or insulating ferrimagnetic layers



Rock-salt phase

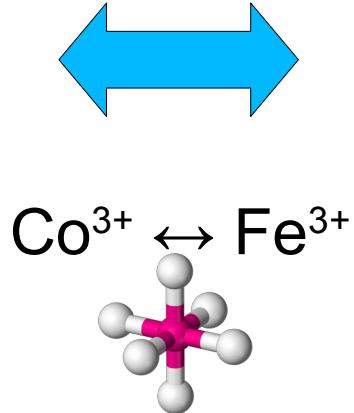


Spinel phase

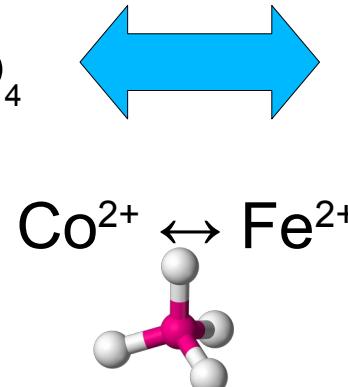
Magnetism in mixed cobalt and iron oxides

Spinel phases

Co_3O_4 :
 $(\text{Co}^{2+})[\text{Co}^{3+}]_2\text{O}_4$
Insulating **AFM**
 $T_{\text{Neel}} = 40 \text{ K}$
Normal Spinel

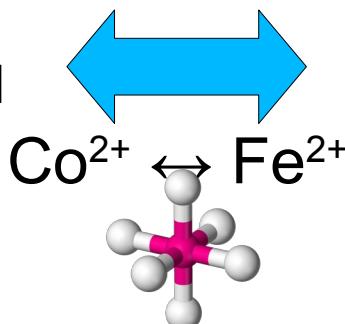


CoFe_2O_4 :
 $(\text{Fe}^{3+})[\text{Co}^{2+}\text{Fe}^{3+}]_2\text{O}_4$
Insulating **Ferrim**
 $T_{\text{Curie}} = 800 \text{ K}$
Inverse Spinel



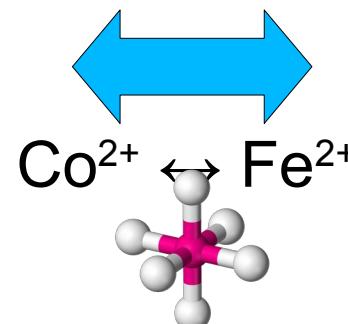
Fe_3O_4 :
 $(\text{Fe}^{3+})[\text{Fe}^{2+}\text{Fe}^{3+}]_2\text{O}_4$
Conducting **Ferrim**
 $T_{\text{Curie}} = 850 \text{ K}$
Inverse Spinel
Half-metal
Verwey Transition

CoO :
Insulating **AFM**
 $T_{\text{Neel}} = 300 \text{ K}$
Mott insulator



Rock-salt phases

$\text{Co}_x\text{Fe}_{1-x}\text{O}$:
Insulating, ??
 $T_{\text{neel}} = ?? \text{ K}$



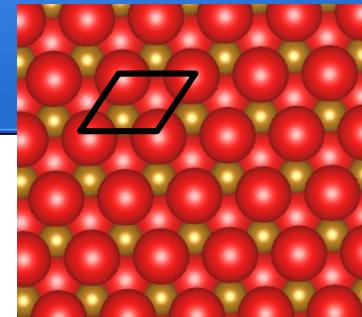
FeO :
Insulating **AFM**
 $T_{\text{Neel}} = 200 \text{ K}$
Mott insulator?

Spinel vs rock-salt along the (111) orientation

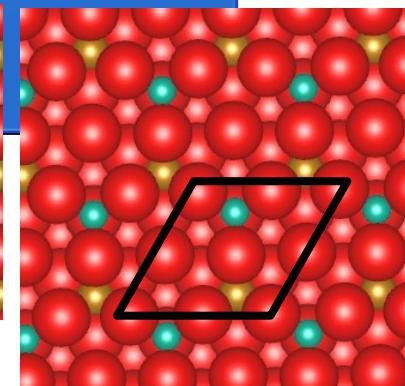
Red- oxygen anions

Yellow- Fe in octahedral sites

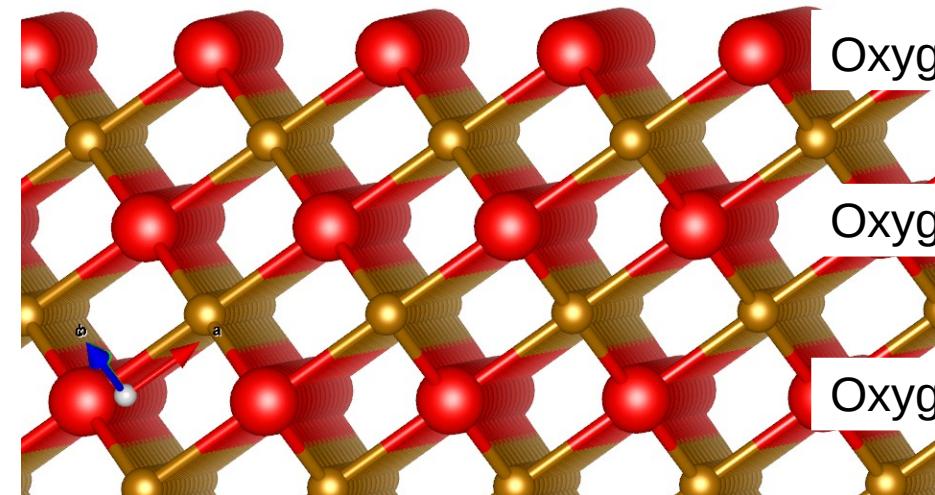
Blue- Fe in tetrahedral sites



Rock-salt unit cell



Full spinel unit cell

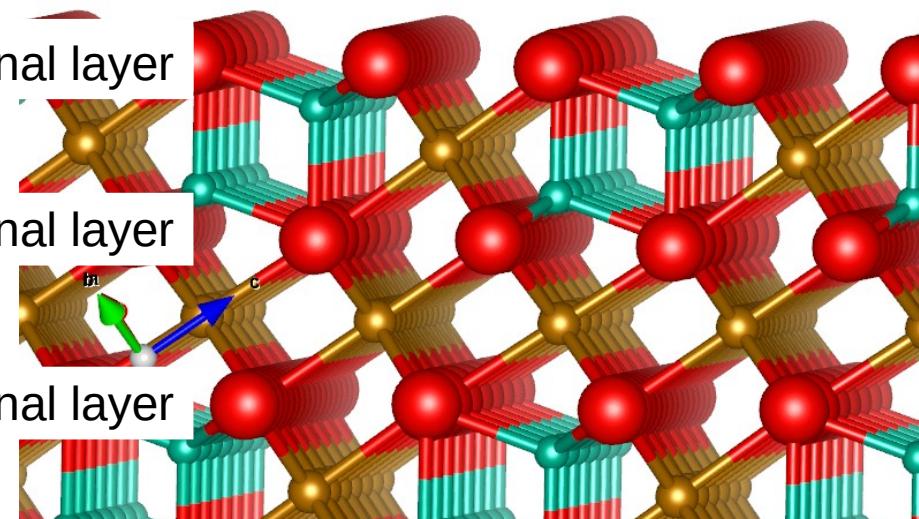


Rocksalt

Oxygen hexagonal layer

Oxygen hexagonal layer

Oxygen hexagonal layer

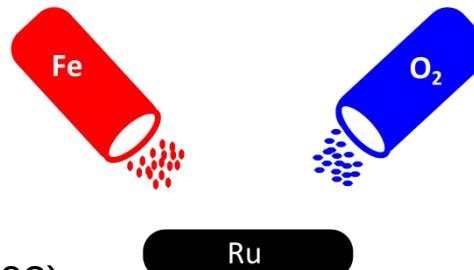
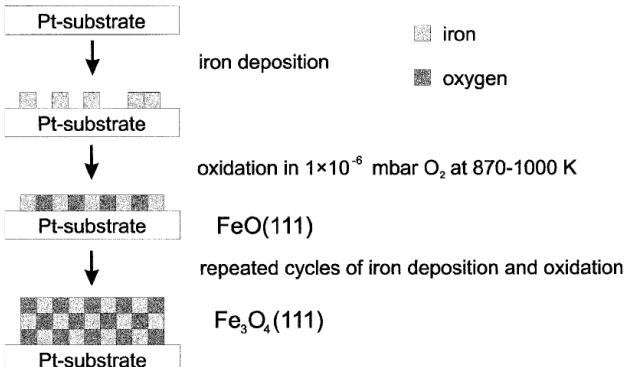


Spinel

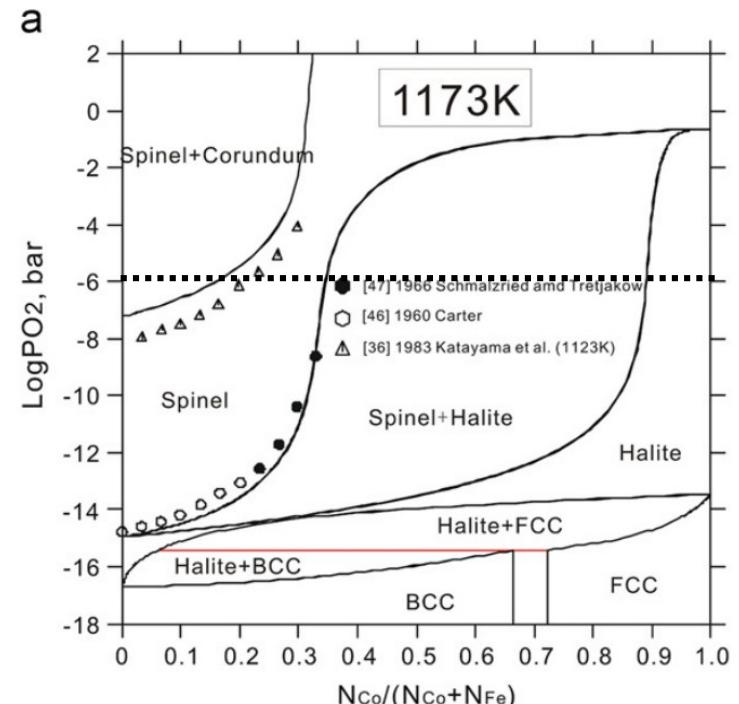
Easily distinguishable by LEED

Growing oxides in UHV: lets look!

- Two methods used in the field:
 - Sequential metal growth and oxidation
 - Oxidation (O_2) while depositing



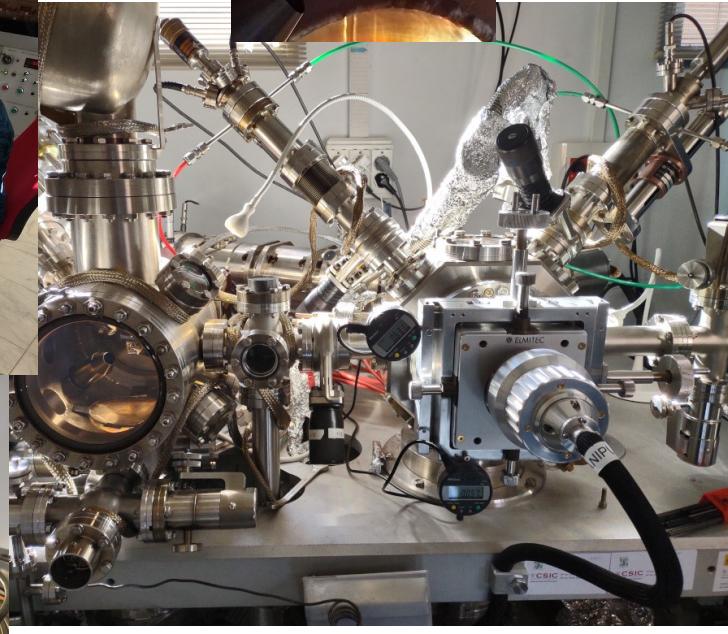
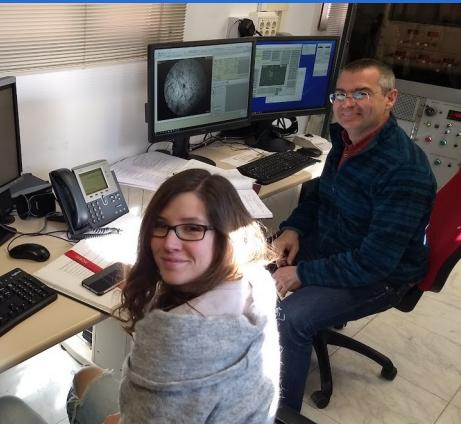
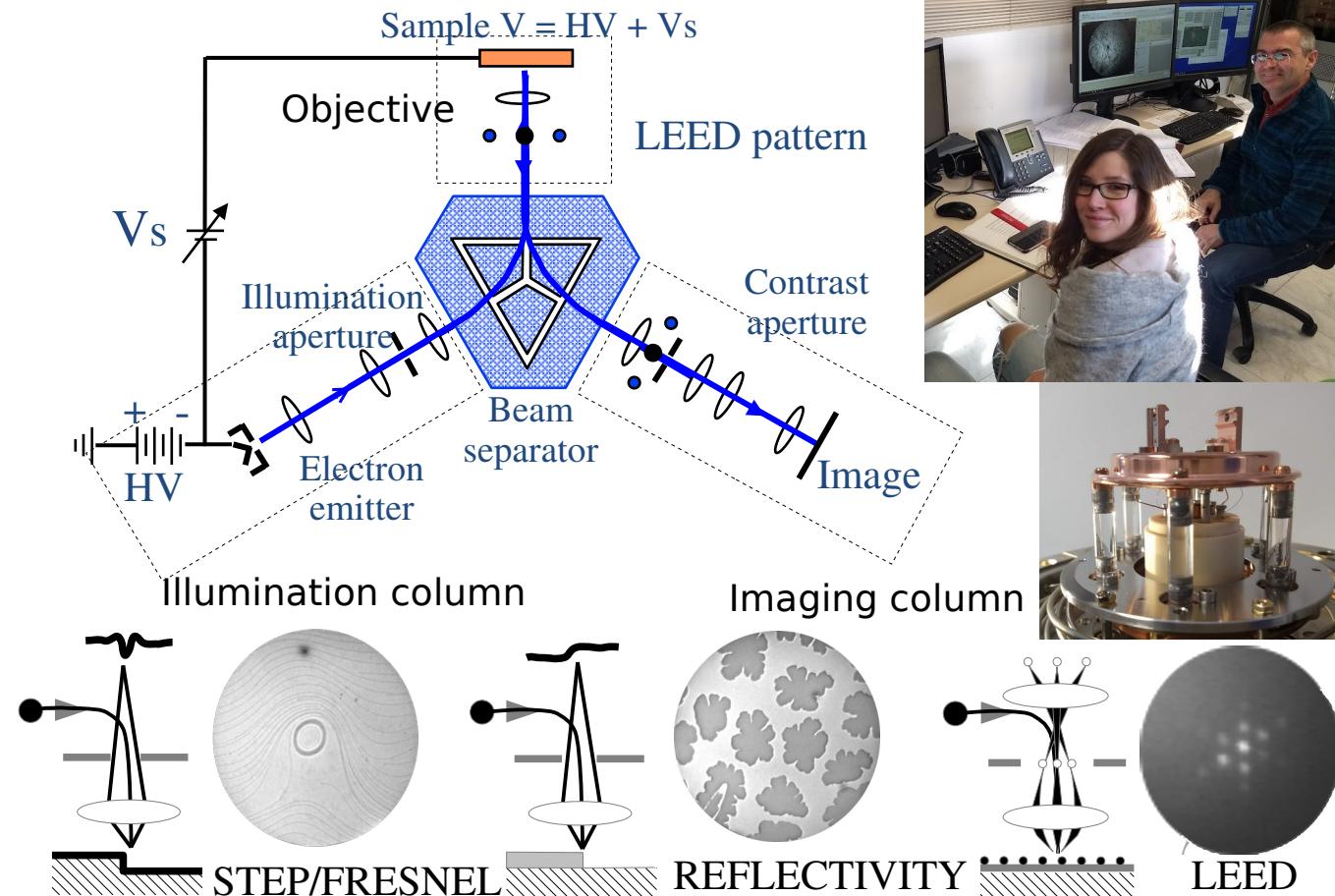
Weiss&Ranke, Prog. Surf. Sci (2002)



Zhang & Chen, CALPHAD 41 (2013) 76

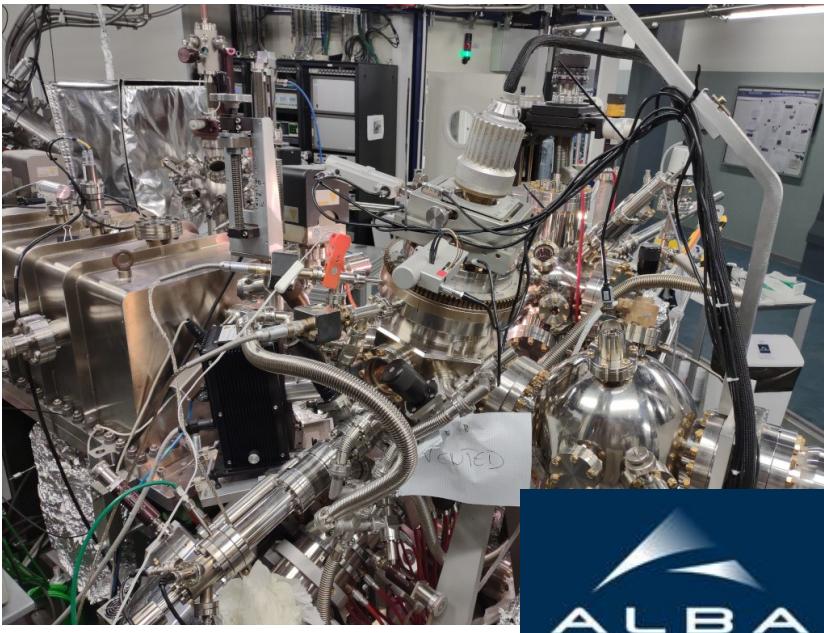
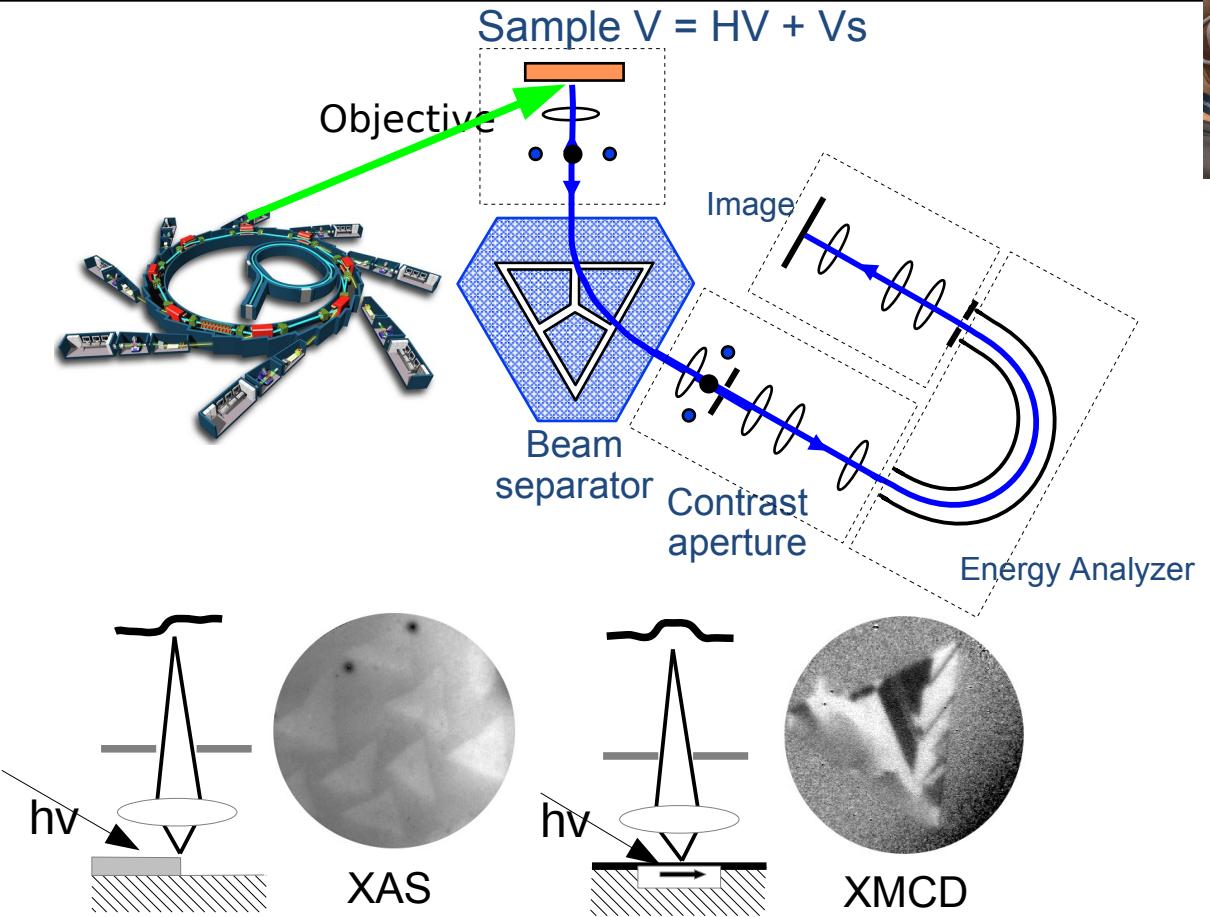
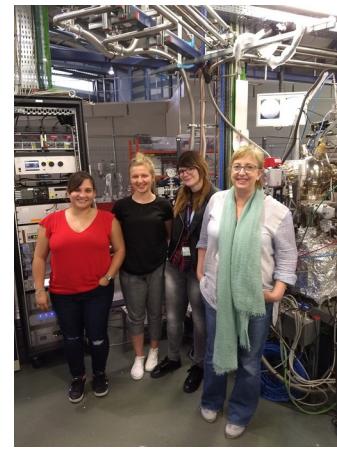
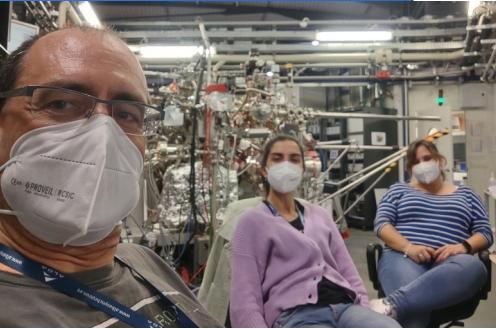
In LEEM we can image in $p < 10^{-5}$ Torr O_2

Low-Energy Electron Microscopy

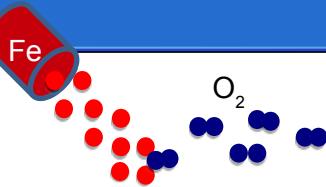


Eelmitec III LEEM at
Instituto de Química Física
“Rocasolano”, CSIC (Madrid)

Adding x-rays: PEEM



Growth of magnetite crystals on Ru(0001)



FOV 10 μm

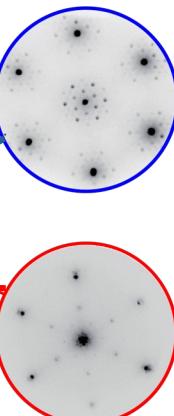
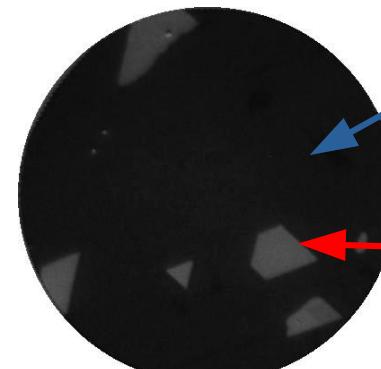
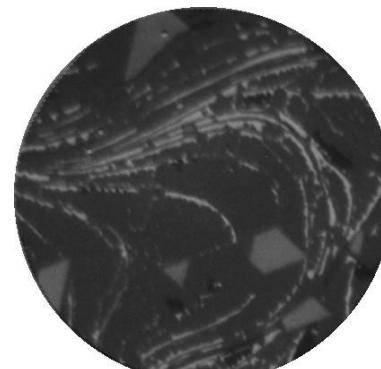
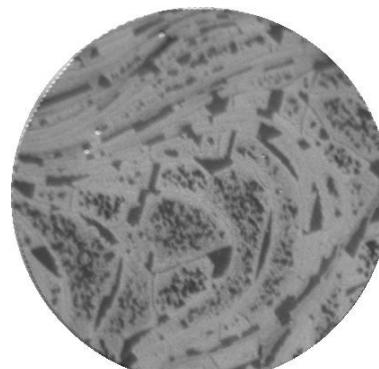
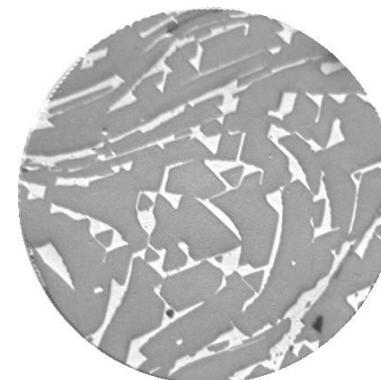
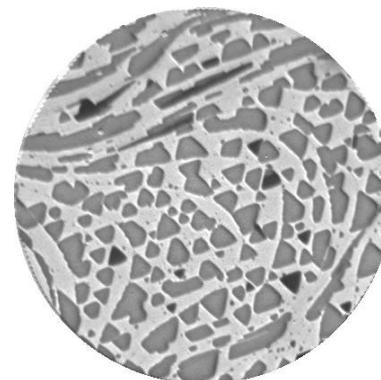
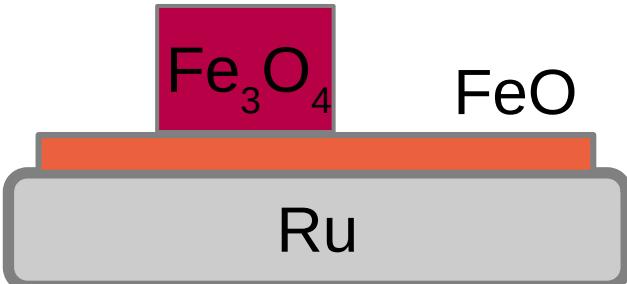
Ru

Reactive-MBE

O₂ pressure~ 10⁻⁶ Torr

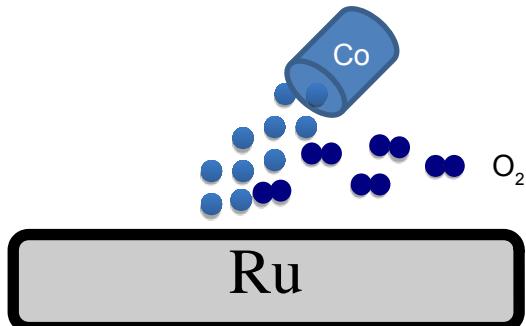
Fe flux~ 1 ML/10

Temperature ~ 600-800°C

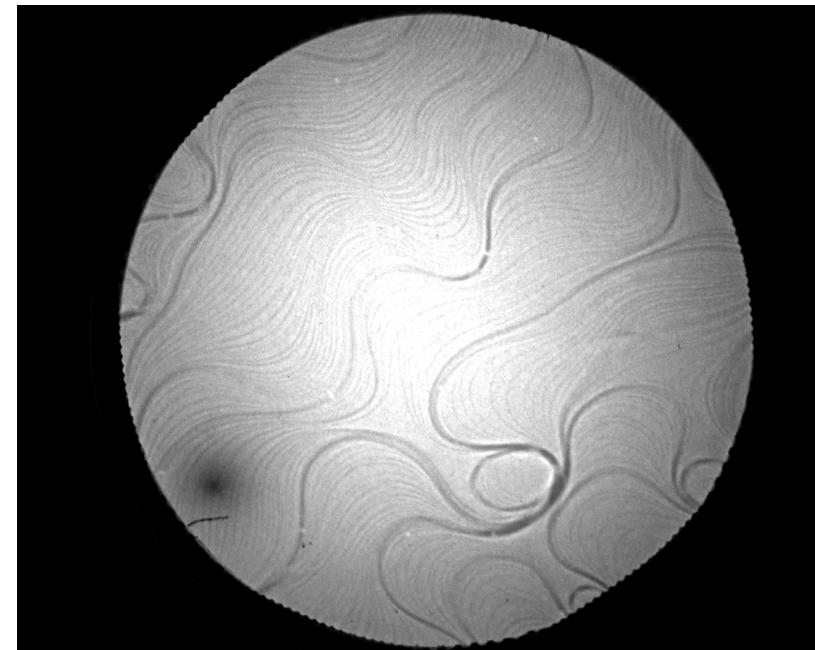


Growth of Co oxides on Ru (0001)

- Growth by HOMBE (High-Temperature, O-assisted MBE):
- Thermal evaporation of Co in O_2 ($p = 1.0 \times 10^{-6}$ mbar) atmosphere.
- Ru(0001) substrate, $T \sim 850$ °C

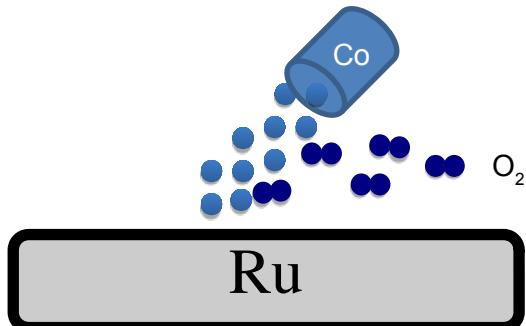


CoO / Ru(0001); FoV: 20 μ m
In-situ monitoring by LEEM

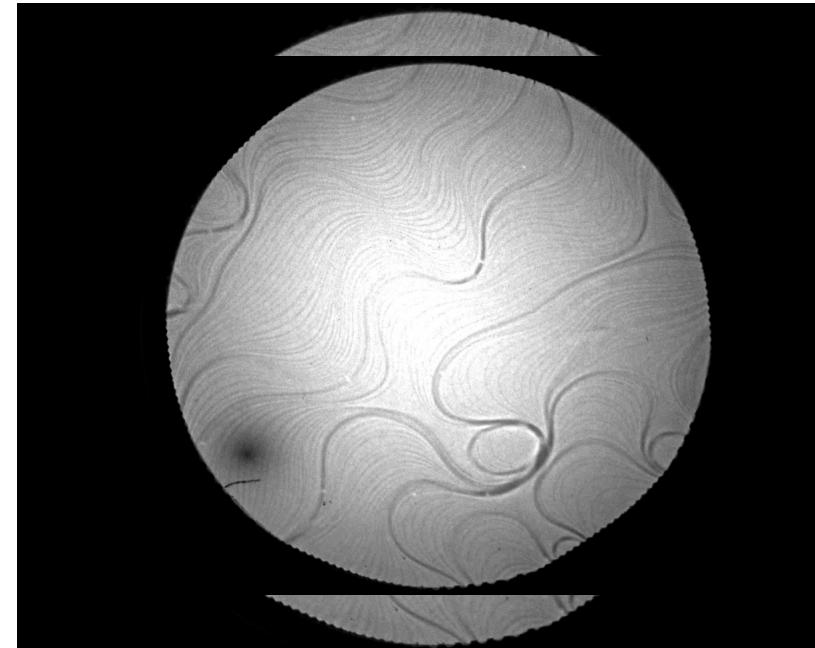


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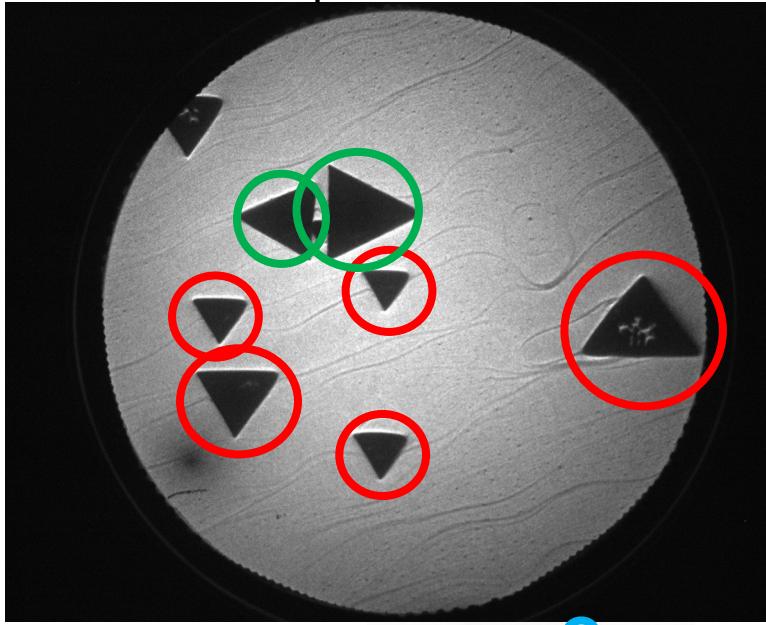


CoO / Ru(0001); FoV: 20 μ m
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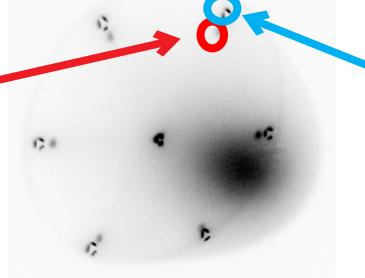


Epitaxial CoO islands on Ru(0001)

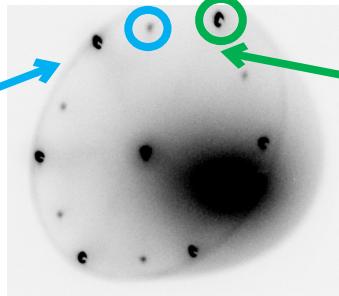
FoV = 20 μm , SV = 50 eV



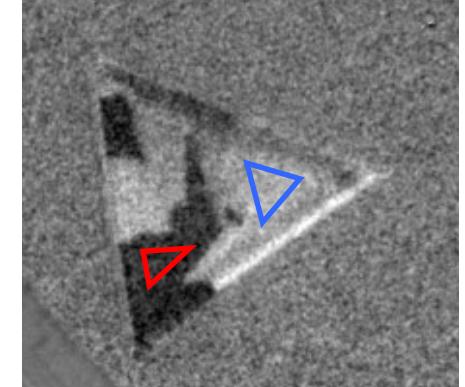
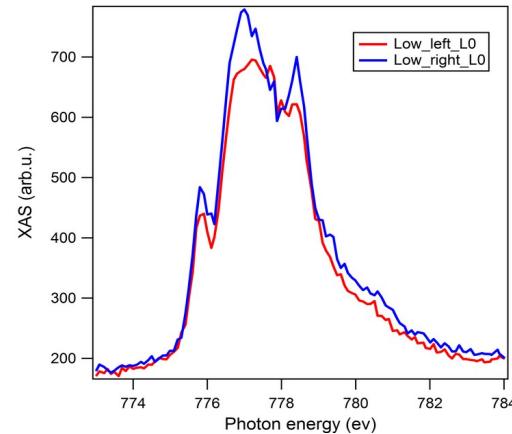
“majority”
island



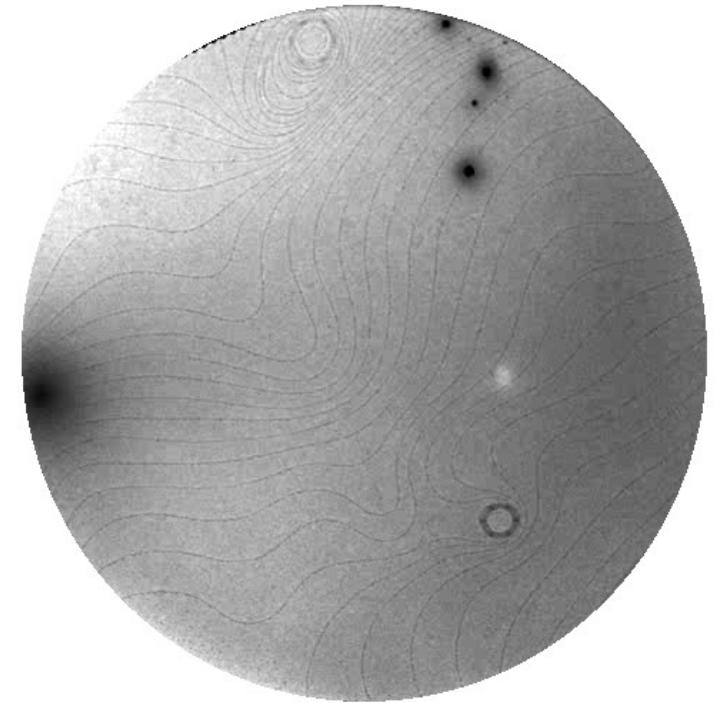
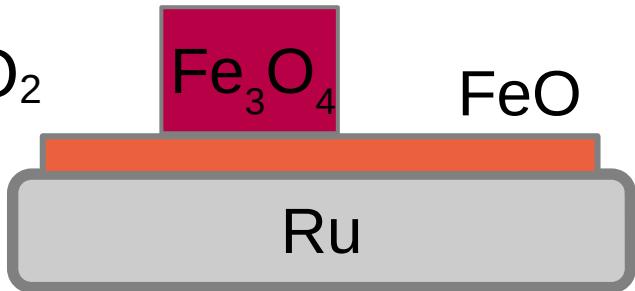
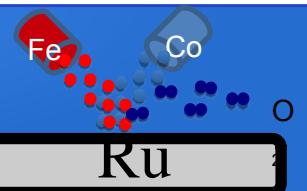
Spots of
substrate



“30° rot.”
island

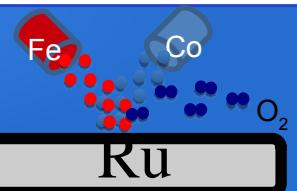


Does it work for ternary spinels?

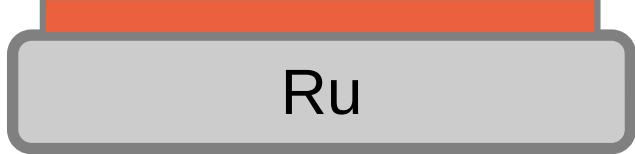


Fe+Co (2:1), in 10^{-6} mbar O_2 on Ru(0001),
FOV 10 μm , 770°C

Does it work for ternary spinels?

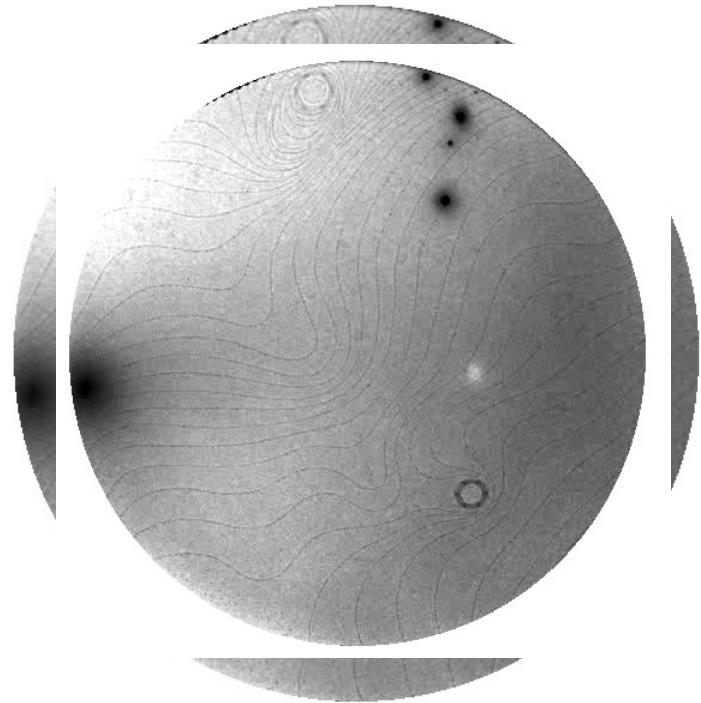


$Fe + O_2$



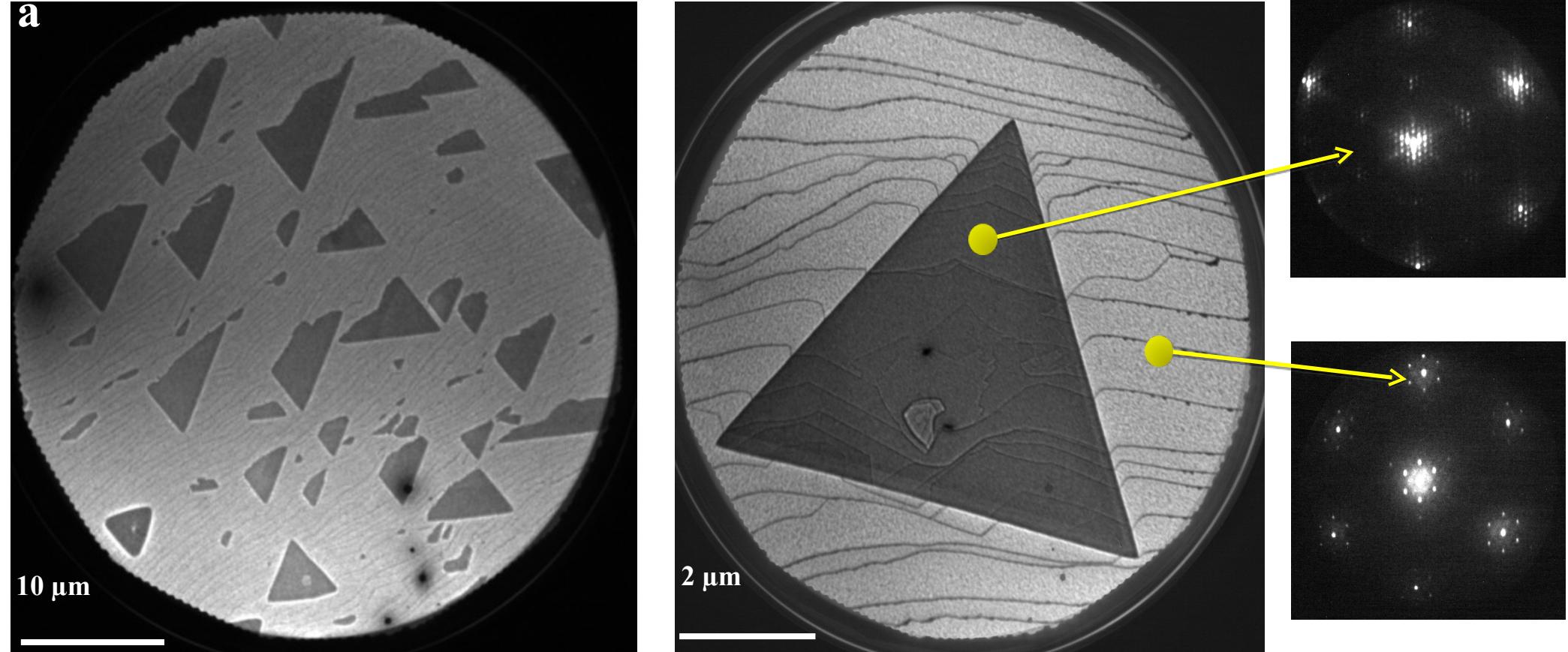
$Fe + Co + O_2$

$Co + O_2$

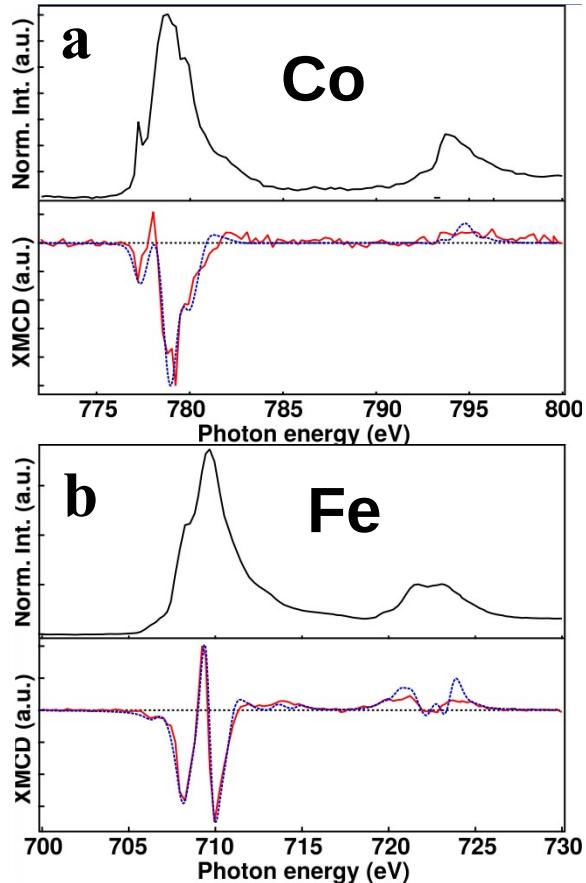


Fe+Co (2:1), in 10^{-6} mbar O_2 on Ru(0001),
FOV 10 μ m, 770°C

It works: wetting layer plus islands



XAS provides chemical characterization

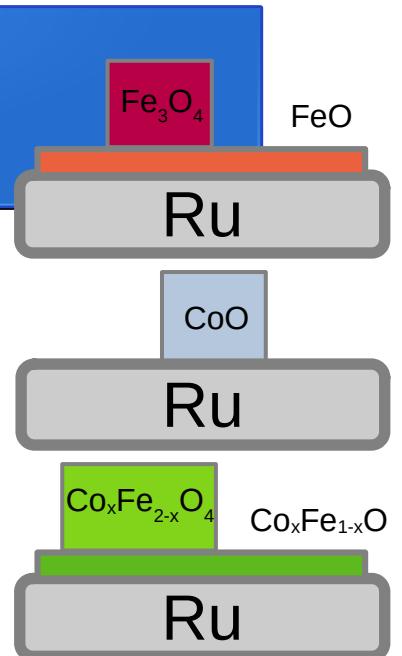


- Deposited ratio: Co:Fe 1:2
- Islands are spinel
- Wetting layer is a Co(II), Fe(II) oxide
- Experimental composition of:
 - Islands: $\text{Co}_{0.5}\text{Fe}_{2.5}\text{O}_4$
 - Wetting layer: $\text{Co}_{0.5}\text{Fe}_{0.5}\text{O}$

Composition does not match
the deposited ratio

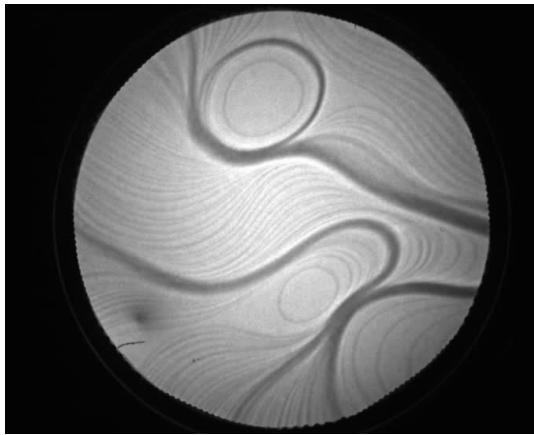
Too many questions...

- Co+O₂ gives Co 3D islands
- Fe+O₂ gives Fe₃O₄ 3D islands on FeO
- 1 Co: 2 Fe + O₂ gives CoFeO₄ 3D islands plus CoFeO
- How does the growth mode depend on the Co:Fe ratio?
- Is the composition of CFO islands and wetting layer constant for a given ratio?
- Can we engineer particular nanostructures?

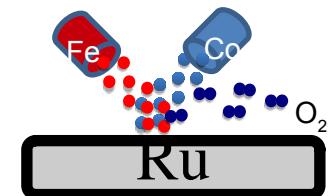
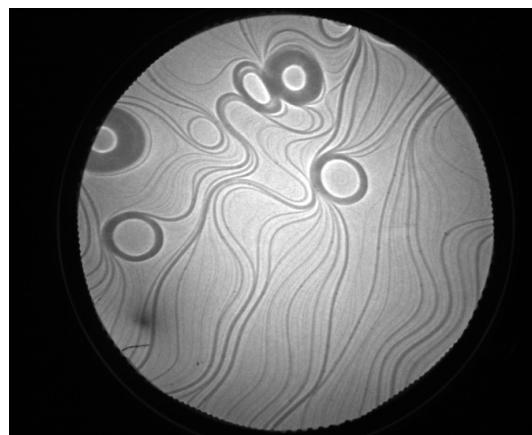


Initial growth with different Co:Fe ratios

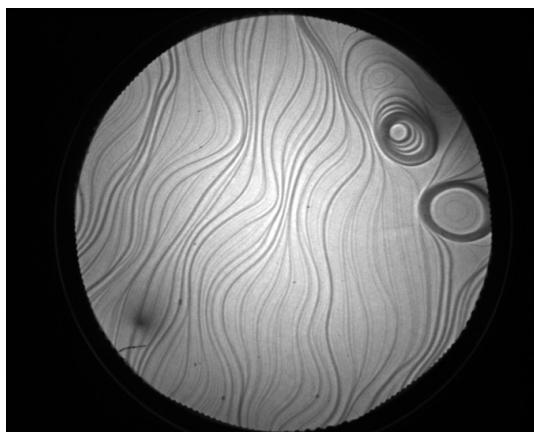
Co:Fe 2:1



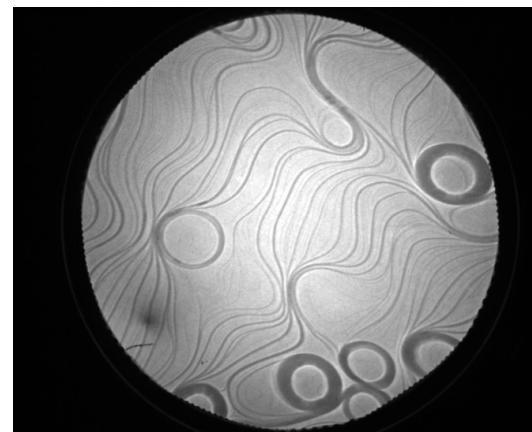
Co:Fe 5:1



Co:Fe 3:1



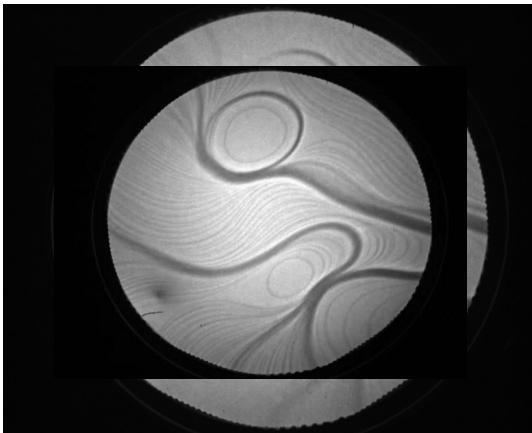
Co:Fe 17:1



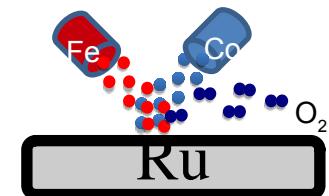
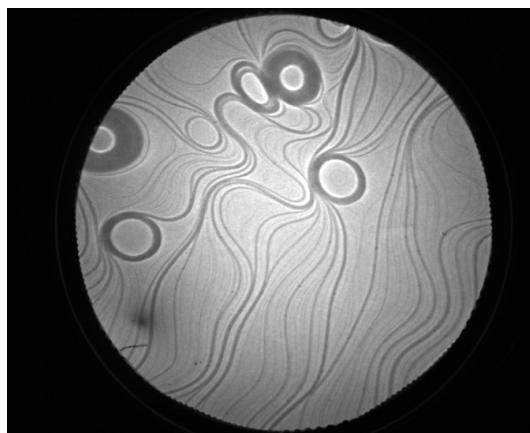
Depositing 1ML Co in 8 min
FOV 10 um
Substrate 850 °C
Pressure oxygen 1×10^{-6} mbar

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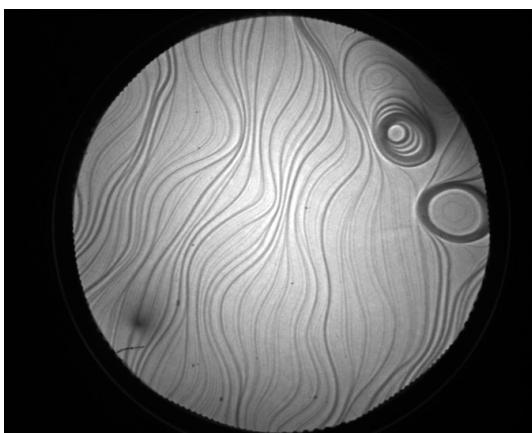
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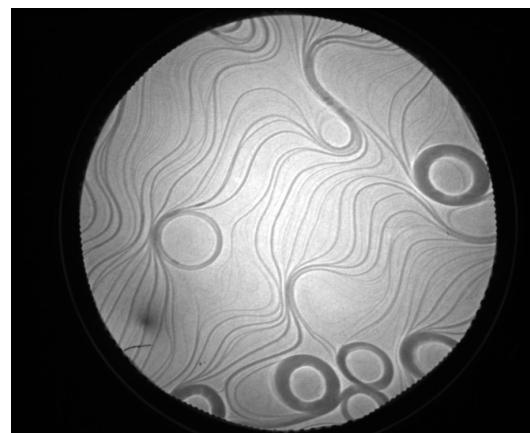
Co:Fe 5:1



Co:Fe 3:1



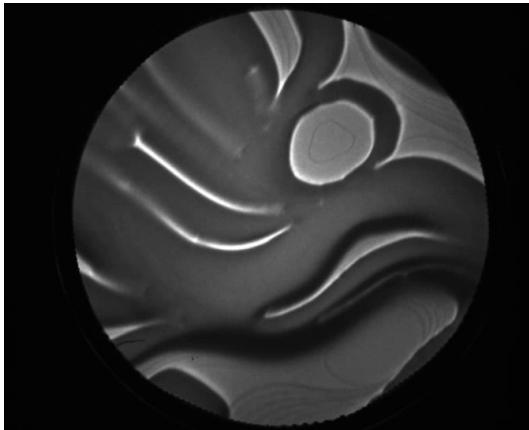
Co:Fe 17:1



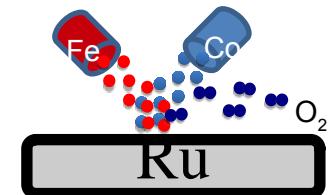
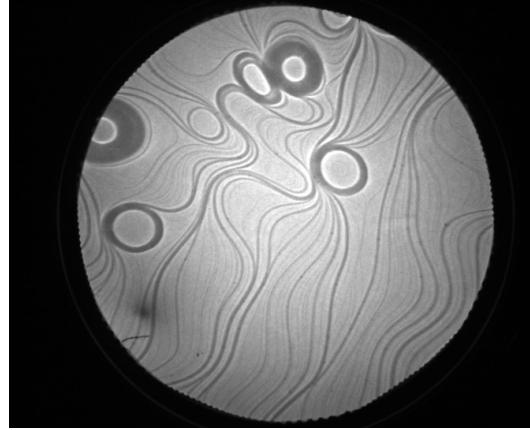
Depositing 1ML Co in 8 min
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Substrate 850 °C
Pressure oxygen 1x10⁻⁶ mbar

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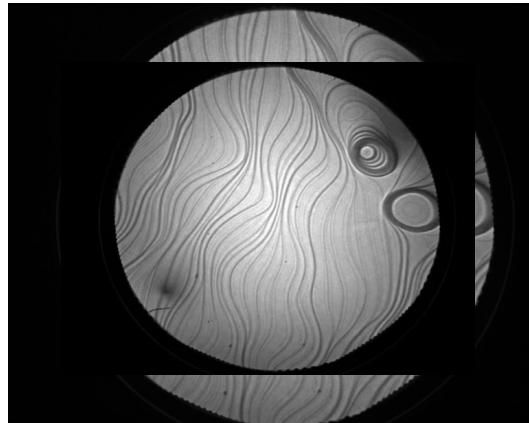
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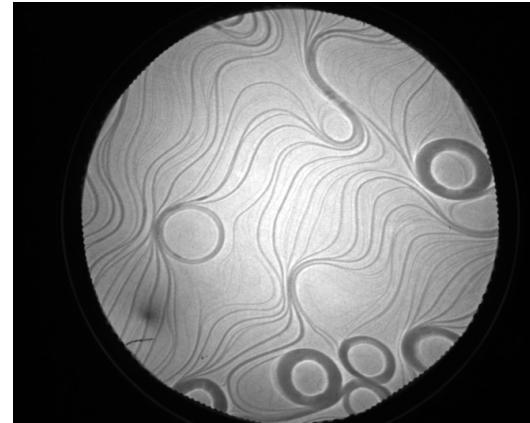
Co:Fe 5:1



Co:Fe 3:1



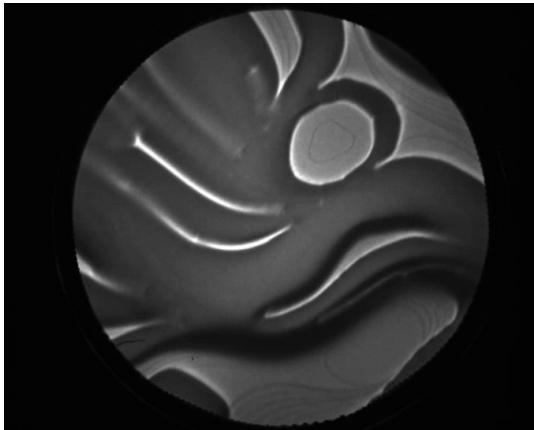
Co:Fe 17:1



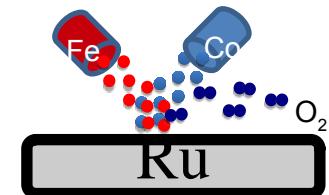
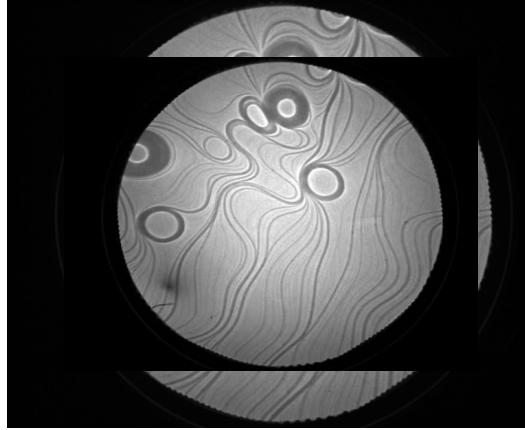
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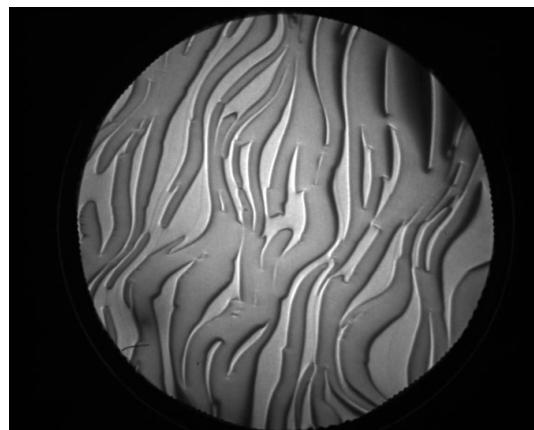
Co:Fe 2:1



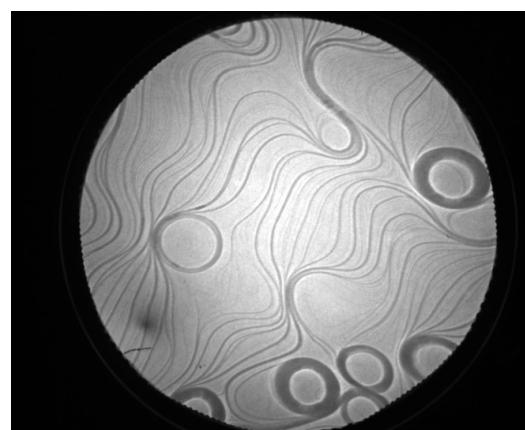
Co:Fe 5:1



Co:Fe 3:1



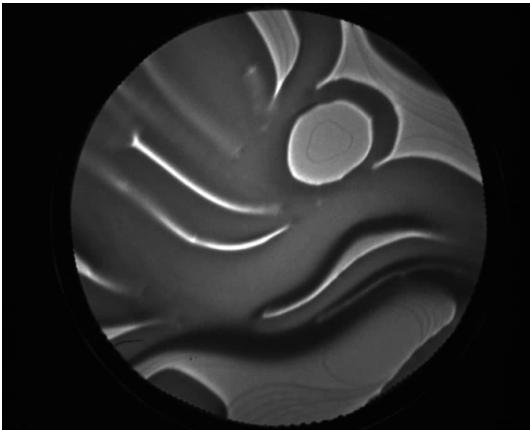
Co:Fe 17:1



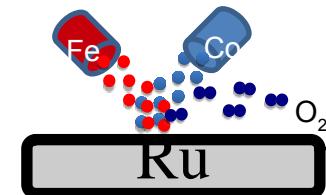
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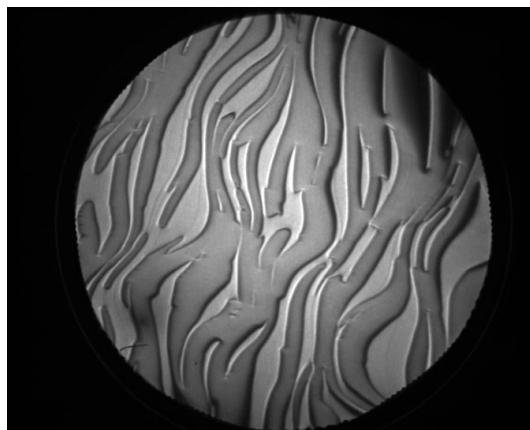
Co:Fe 2:1



Co:Fe 5:1



Co:Fe 3:1



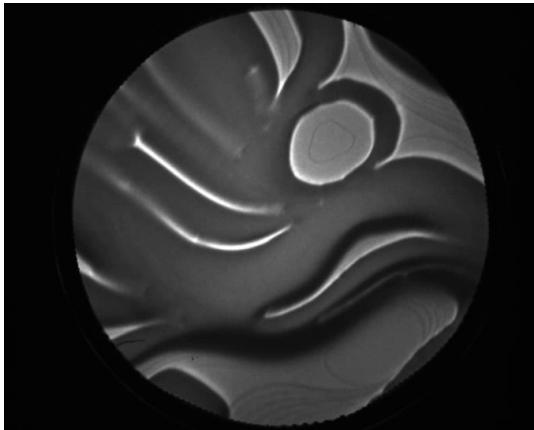
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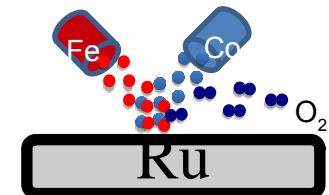
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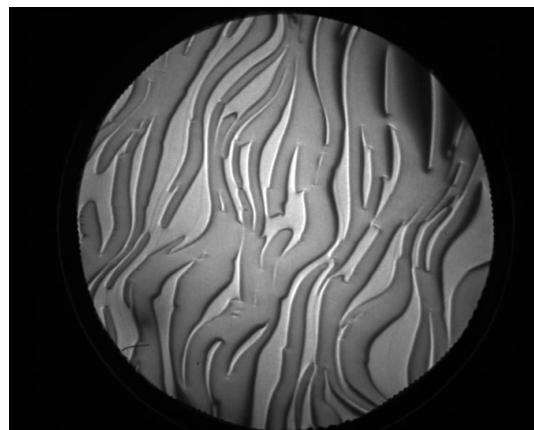
Co:Fe 2:1



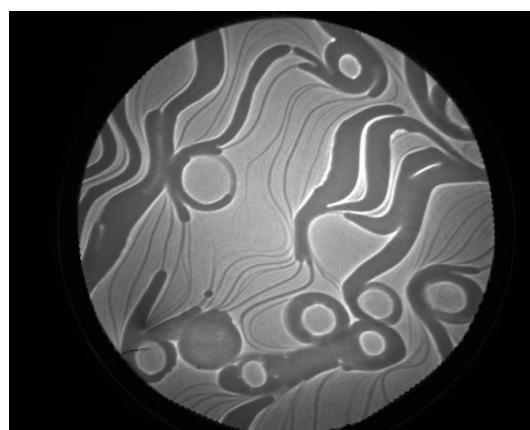
Co:Fe 5:1



Co:Fe 3:1

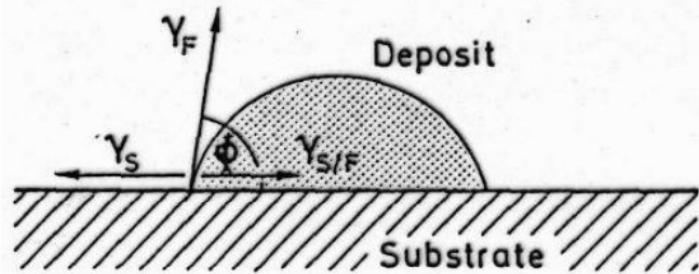


Co:Fe 17:1



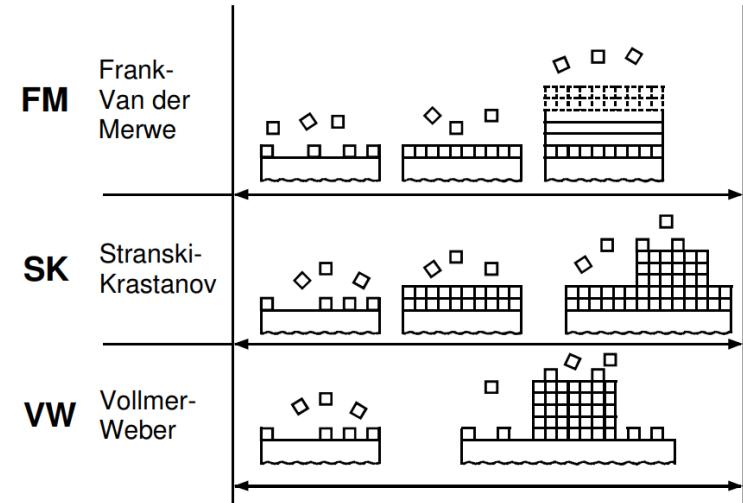
Depositing 1ML Co in 8 min
FOV 10 μm
Substrate 950 °C
Pressure oxygen 1×10^{-6} mbar

Why the transition from layer-by-layer to 2D growth?



Layer-by-Layer (FM): $\Phi = 0$, $\gamma_S \geq \gamma_F + \gamma_{S/F}$

3D growth (VW): $\Phi > 0$, $\gamma_S < \gamma_F + \gamma_{S/F}$

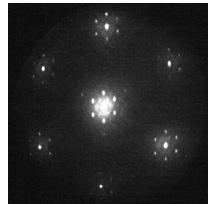


Bulk CoO/FeO: lattice 0.301/0.305 nm

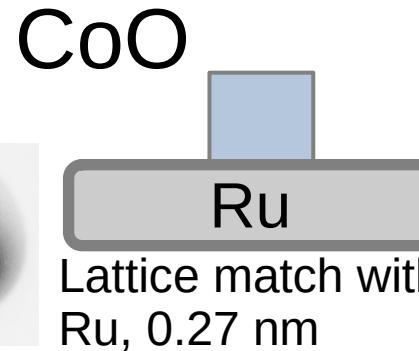
$\text{Co}_x\text{Fe}_{1-x}\text{O}$, with $0 < x < 0.95$



Lattice 0.32 nm



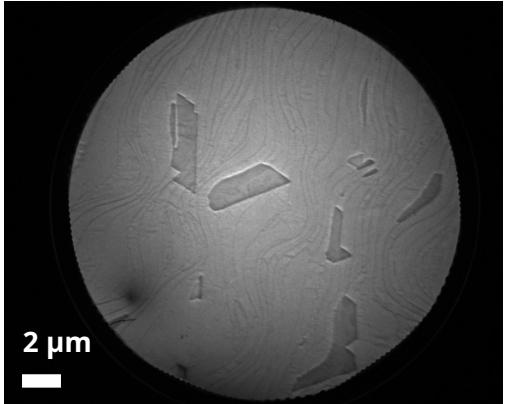
Moiré underneath



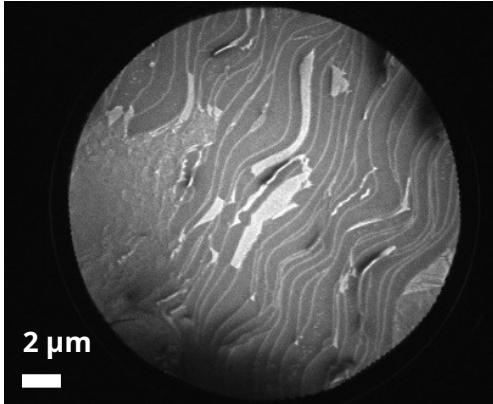
Lattice match with
Ru, 0.27 nm

But at later stages things differ strongly!!!

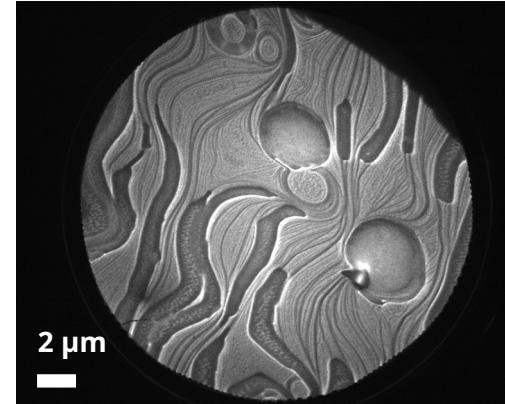
Work in progress



Co:Fe 2:1



Co:Fe 3:1

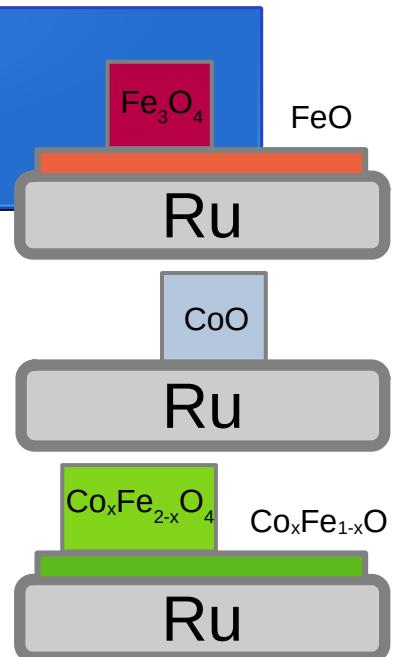


Co:Fe 17:1

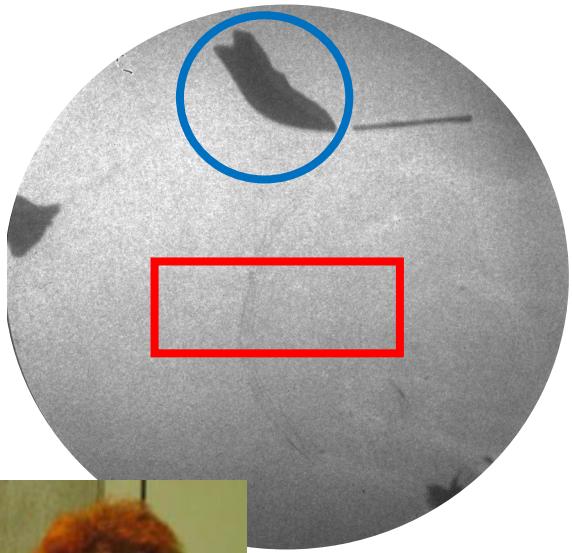
- 5 % Fe enough to favor initially a 2D-growth (pure FeO) over a 3D-growth (pure CoO)
- 30% Fe required to grow a homogeneous film

Too many questions...

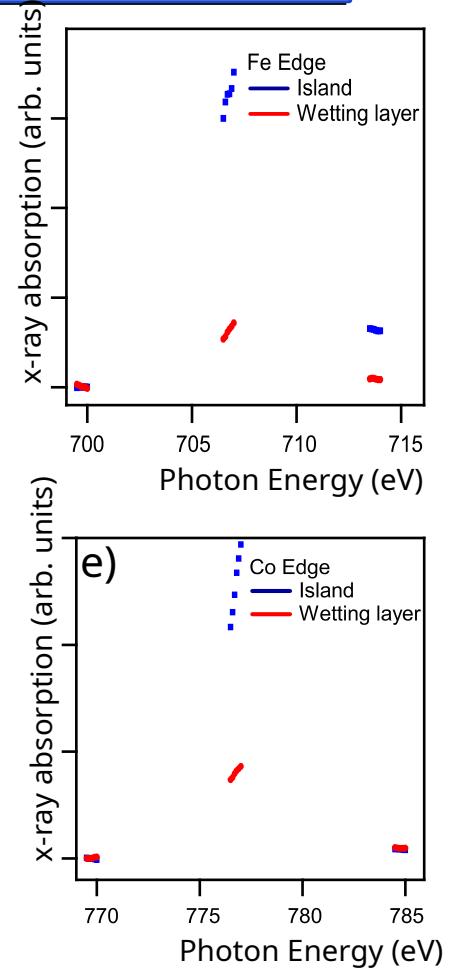
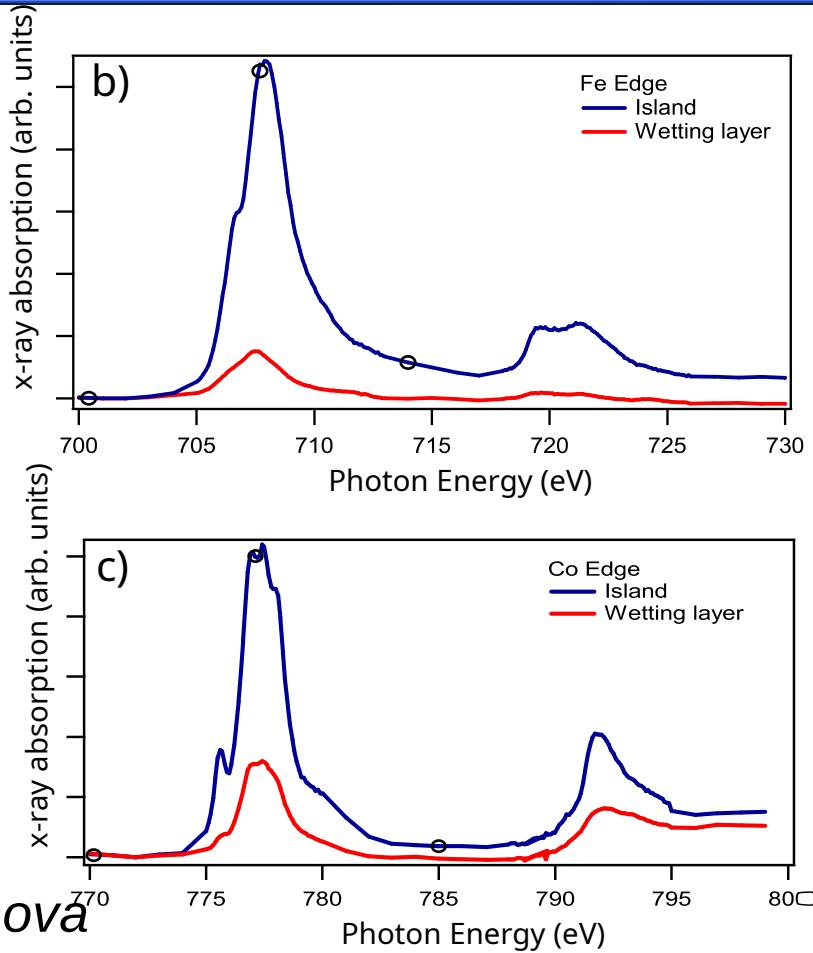
- Co+O₂ gives Co 3D islands
- Fe+O₂ gives Fe₃O₄ 3D islands on FeO
- 1 Co: 2 Fe + O₂ gives CoFeO₄ 3D islands plus CoFeO
- How does the growth mode depend on the Co:Fe ratio?
- Is the composition of CFO islands and wetting layer constant for a given ratio?
- Can we engineer particular nanostructures?



How to follow growth in PEEM

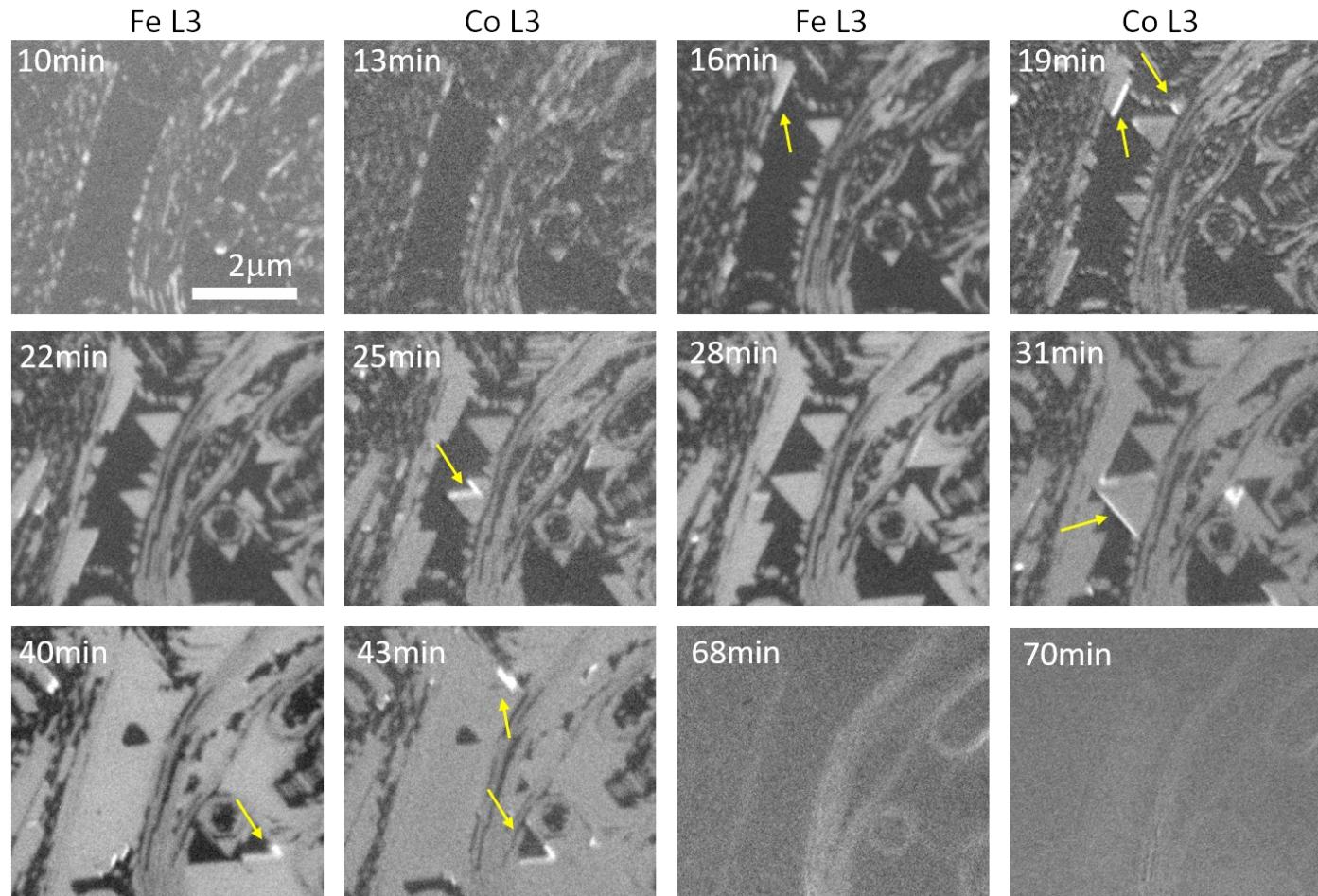
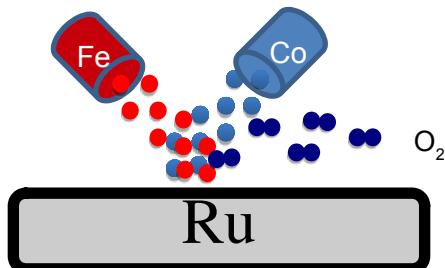


Suggested by Maya Kiskinova

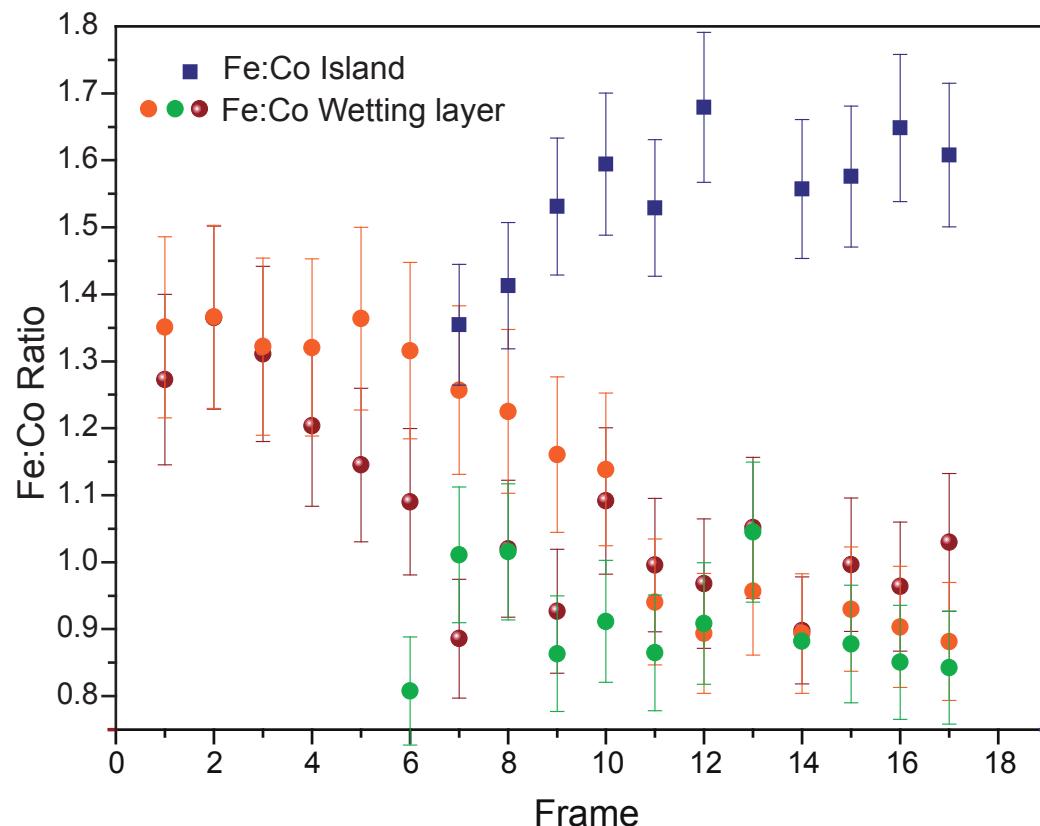
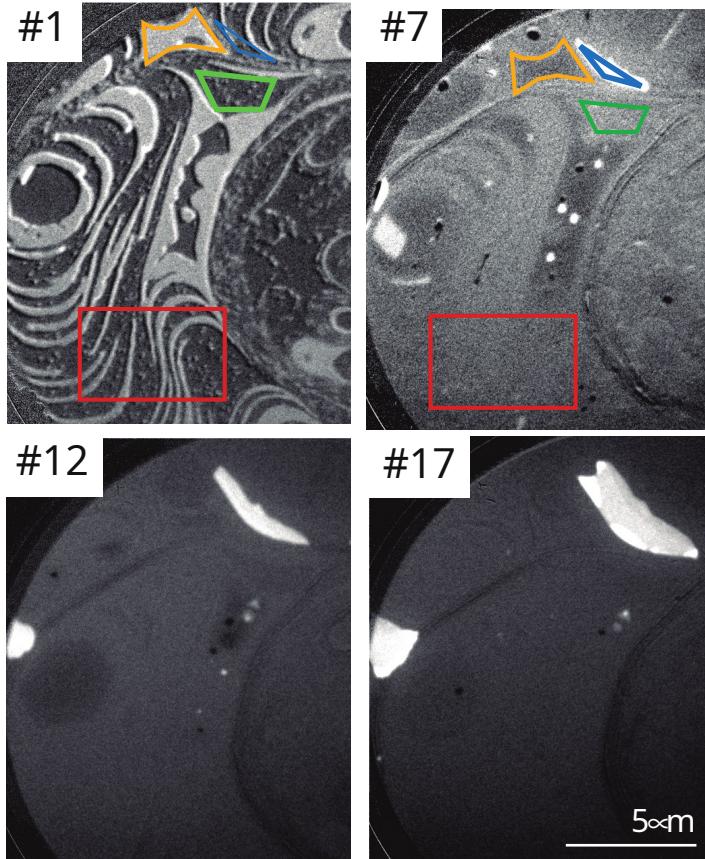


A PEEM growth movie

Following growth
in PEEM,
switching between
Fe and Co XAS
edges

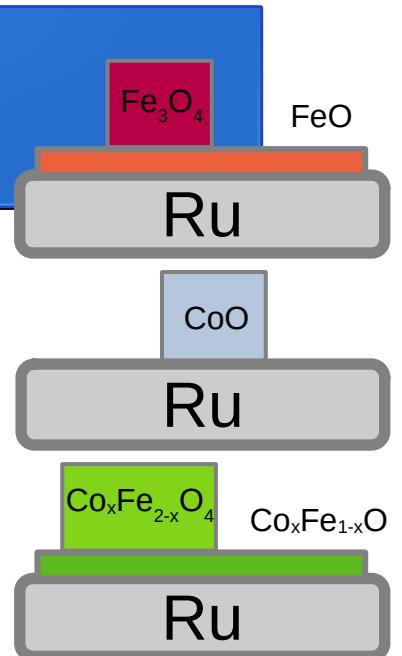


Evolution of the composition of the film



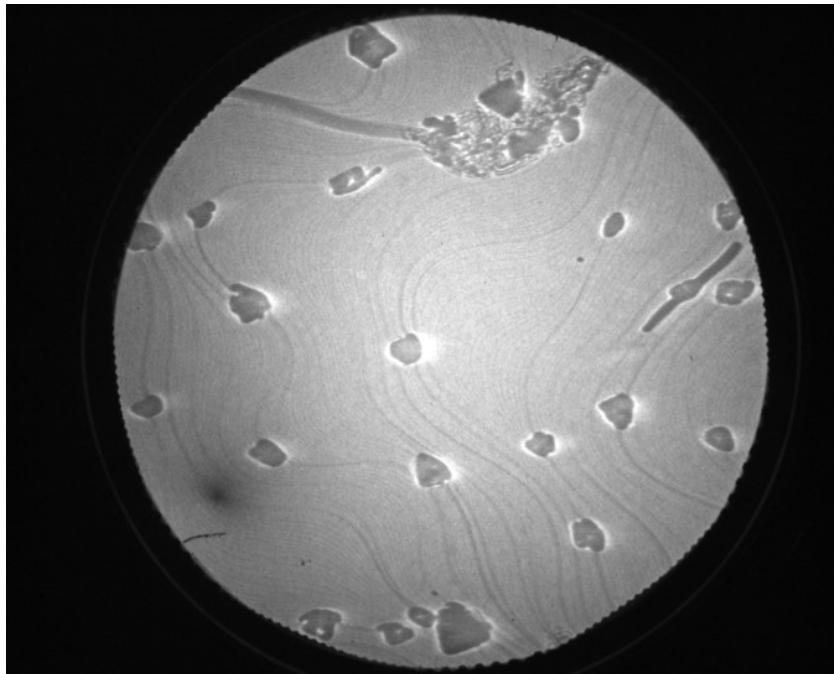
Too many questions...

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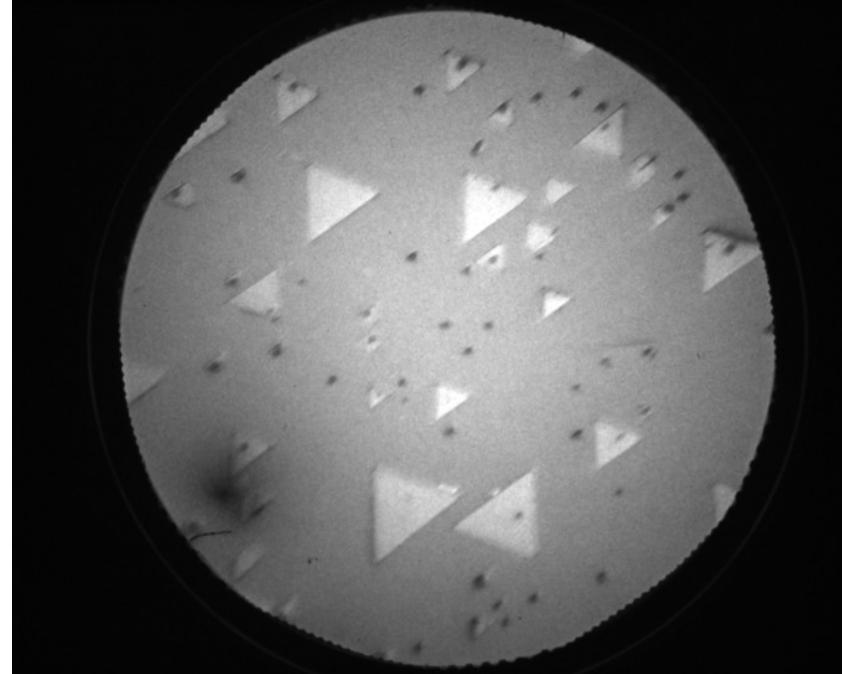


CoO/Fe₃O₄ nanostructures

grow CoO, then Fe₃O₄



growth Fe₃O₄, then CoO

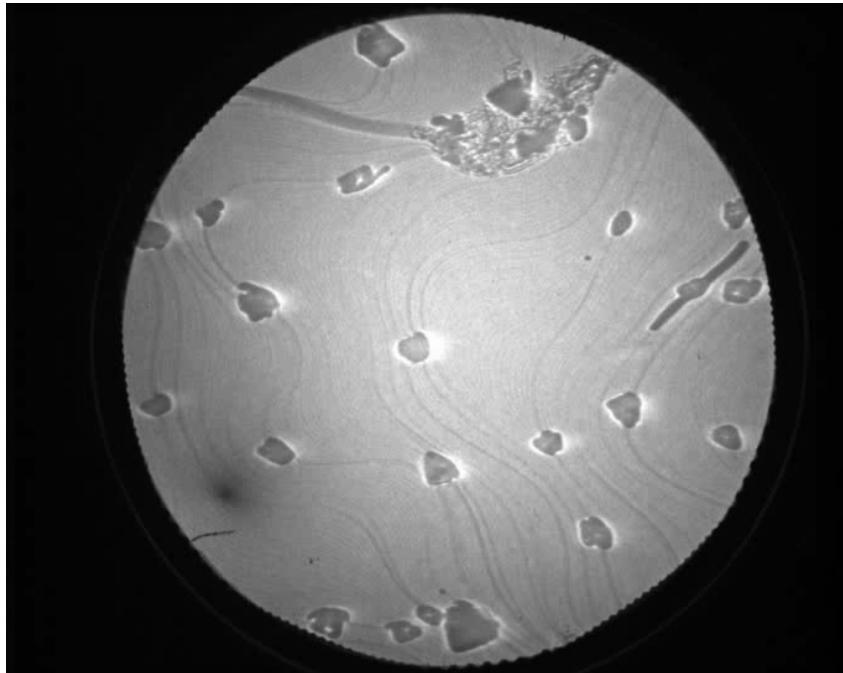


CoO islands dissolved upon Fe deposition

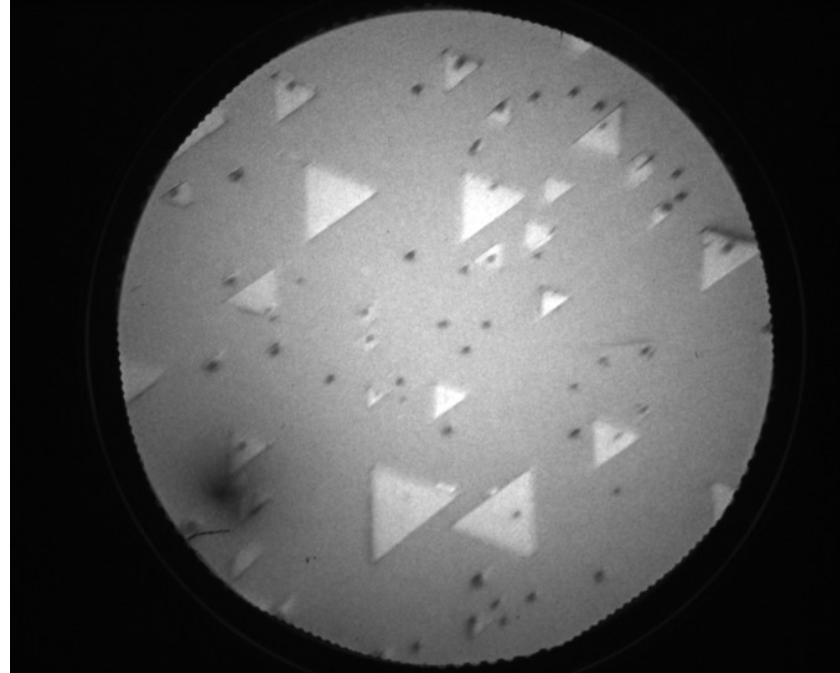
CoO islands covered Fe₃O₄ ones???

CoO/Fe₃O₄ nanostructures

grow CoO, then Fe₃O₄



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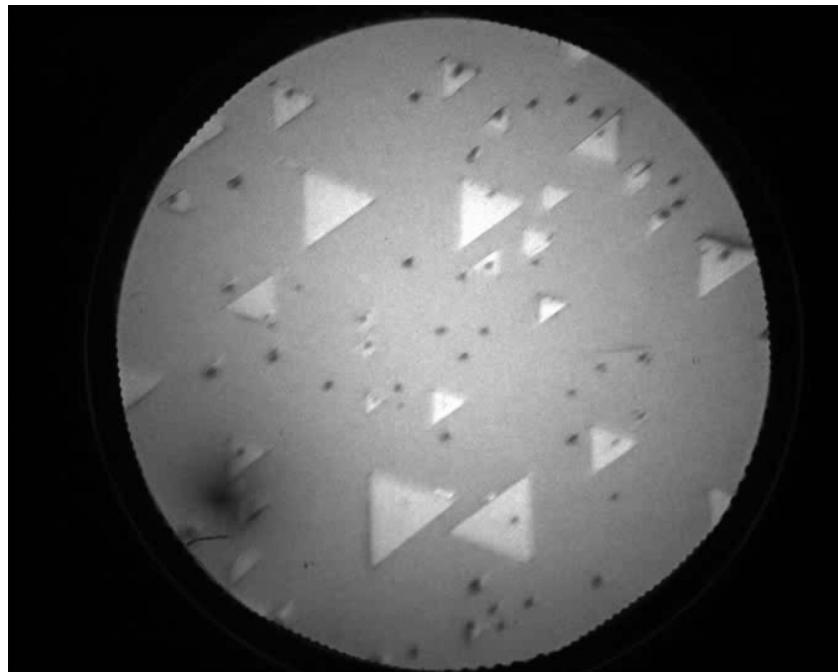
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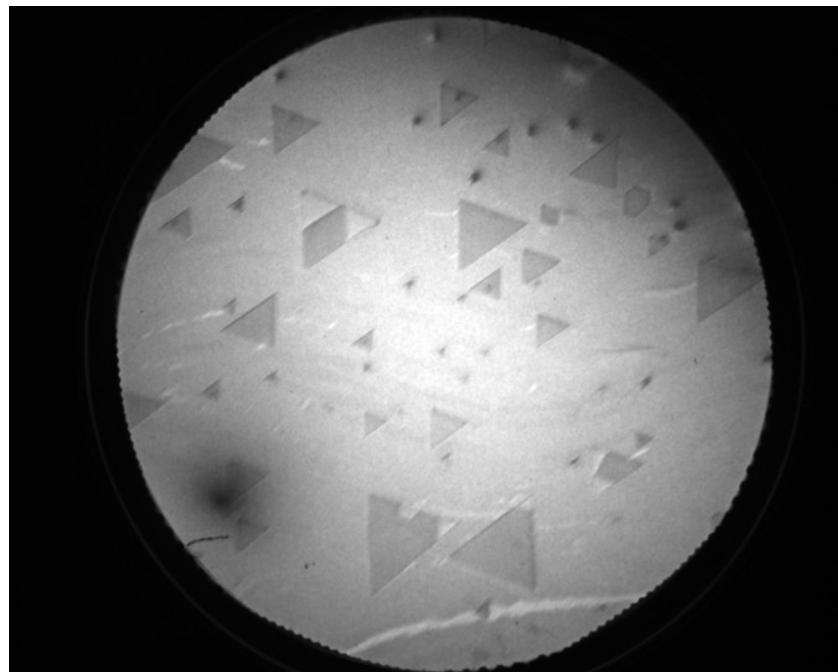
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CoO/Fe₃O₄ nanostructures

grow CoO, then Fe₃O₄



growth Fe₃O₄, then CoO



CoO islands dissolved upon Fe deposition

CoO islands covered Fe₃O₄ ones???

Conclusions

- Electron microscopies are great for growth characterization:
 - LEEM gives a real-space view of the growth front
 - PEEM gives composition. With some effort, it can be used in real time also.
- Co-Fe oxides are a fascinating playground of different phases with different magnetic properties
 - Simultaneous deposition of 2:1 Fe/Co in O₂ produces a rock-salt wetting layer and 3-dimensional spinel islands.
 - The growth mode can be tuned by changing the Fe/Co ratio
 - The composition evolves with time even upon a given Fe/Co deposition ratio when there are several phases present. **MORE PEEM MOVIES?**
- Combined CoO/Fe₃O₄ structures promise highly perfect systems to study AFM/FM magnetic coupling

And thanks to:

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Natalia Kwiatek



Lucía Aballe



Michael Foerster



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